Supplementary material

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Enhancement of electrochemical performance of monolayer  $SnS_2$  for Li/Na-ion batteries through a sulphur vacancy: A DFT study

## C.A. Bekeur<sup>1</sup>, R.E. Mapasha<sup>1\*</sup>

Department of Physics, University of Pretoria, Hatfield campus, Pretoria 0002, RSA<sup>1</sup>

\*Corresponding author Email: edwin.mapasha@up.ac.za



(c) Li at  $H_{SnS_2}$  site on pristine  $SnS_2$ 

(d) Na at  $\mathbf{H}_{SnS_2}$  site on pristine  $\mathbf{SnS_2}$ 





(e) Li at  $T_{Sn}$  site on pristine  $SnS_2$ 



(g) Li at  $T_S$  site on pristine  $SnS_2$ 





(f) Na at  $T_{Sn}$  site on pristine  $SnS_2$ 



(h) Na at  $T_S$  site on pristine  $SnS_2$ 



(i) Li at  $T_{Sn}$  site on S-vacancy  $SnS_2$ 





(k) Li at  $V_S$  site on S-vacancy  $SnS_2$ 





(j) Na at  $T_{Sn}$  site on S-vacancy  $SnS_2$ 





(l) Na at  $\mathbf{V}_S$  site on S-vacancy  $\mathbf{SnS}_2$ 



(o) Li at  $T_S$  site on S-vacancy  $SnS_2$ 



Figure 1: The relaxed various configurations (a-p) of Li/Na adatoms on the pristine and S-vacancy SnS<sub>2</sub> monolayer  $5 \times 5$  supercell.



(c) 25 Li adatoms on  $SnS_2$  with S-vacancy



Figure 2 : The total density of states of multiple Li/Na adatoms on pristine  $SnS_2$  monolayer as well as on  $SnS_2$  monolayer with S-vacancy.