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ISSN 2414-3146

Bis[(4-methylphenyl)diphenylphosphane- κP]- (nitrate- $\kappa^2 O, O'$)silver(I)

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Received 23 September 2022

Accepted 31 October 2022

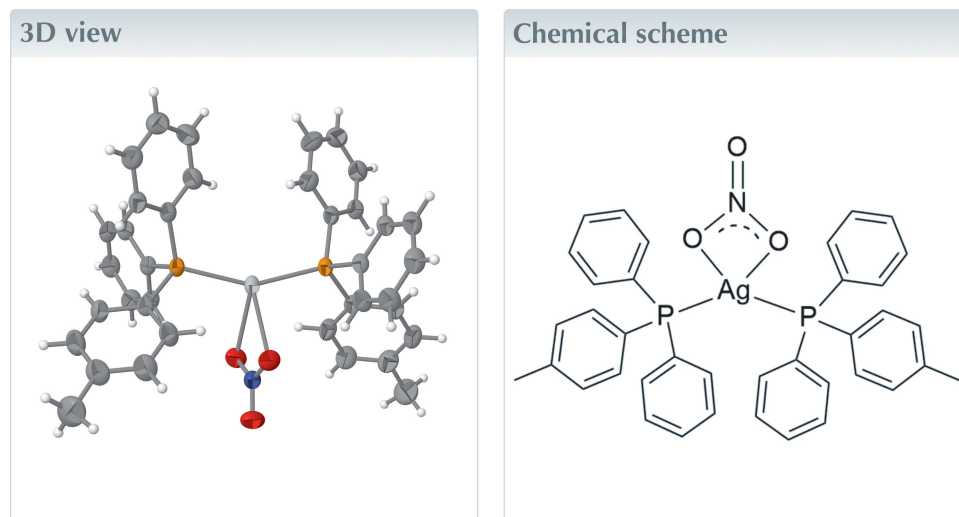
Edited by E. R. T. Tiekink, Sunway University, Malaysia

Keywords: silver(I) complex; diphenyl-*p*-tolyl phosphine; nitrate; crystal structure.

CCDC reference: 2216475

Structural data: full structural data are available from iucrdata.iucr.org

The molecular structure of the title Ag^I complex, [Ag(NO₃)(C₁₉H₁₇P)₂] or [Ag{(p-CH₃C₆H₄)(C₆H₅)₂P- κP }-NO₃- $\kappa^2 O, O'$ }], is described, where a distorted trigonal-planar coordination environment is exhibited about the central Ag^I atom; in this description, the two O atoms are assumed to occupy one position in the coordination sphere. The compound crystallized with half a molecule in the asymmetric unit having the Ag^I atom lying on a twofold axis.



Structure description

Silver(I) phosphine complexes have been found to exhibit high antimicrobial, anti-bacterial and anticancer activity (Potgieter *et al.*, 2016). The continuously expanding library of active compounds leads to a growing interest into their solid- and solution-state characterization, including by single-crystal X-ray diffraction.

The molecular structure of the title compound is shown in Fig. 1. The complex crystallized in the monoclinic space group *C*2/*c*, *Z* = 4 with the asymmetric unit comprising one half of the silver complex molecule, as the central Ag^I atom, along with the nitrate-N and one nitrate-O atom, lying on a twofold axis. Coordinated to the Ag^I atom are two diphenyl-*p*-tolylphosphine ligands, and one nitrate ligand, *via* two O atoms; in this description, the two O atoms are assumed to occupy one position in the coordination environment. The symmetry present in the molecule results in identical Ag–P1 [2.4095 (9) Å] and Ag–O1 [2.522 (3) Å] bond lengths, which fall within the ranges of related compounds (Potgieter *et al.*, 2016). The distorted trigonal-planar coordination displayed by the central Ag^I atom stems from the three coordinating ligands, with corresponding bond angles P1–Ag1–P1ⁱ [152.89 (5)°], P1–Ag1–O1 [109.76 (7)°], and P1–Ag1–O1ⁱ [94.92 (7)°]; symmetry operation: (i) 1 – *x*, *y*, $\frac{3}{2}$ – *z*. The bidentate mode of coordination of the nitrate ligand is confirmed by the O1–Ag1–O1ⁱ bite angle of



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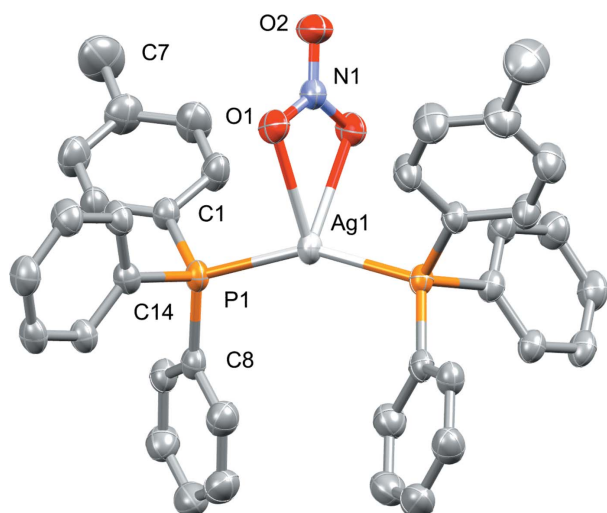


Figure 1
Perspective view of the molecular structure of the title compound showing thermal displacement ellipsoids at the 50% probability level. Hydrogen atoms are omitted for clarity. Unlabelled atoms are related by a twofold axis of symmetry.

50.63 (12)°. The *ipso*-aryl carbon atoms of each of the phosphine ligands overlap in a near-staggered fashion when viewed down the P1–Ag1–P1ⁱ axis, presumably due to the steric bulk

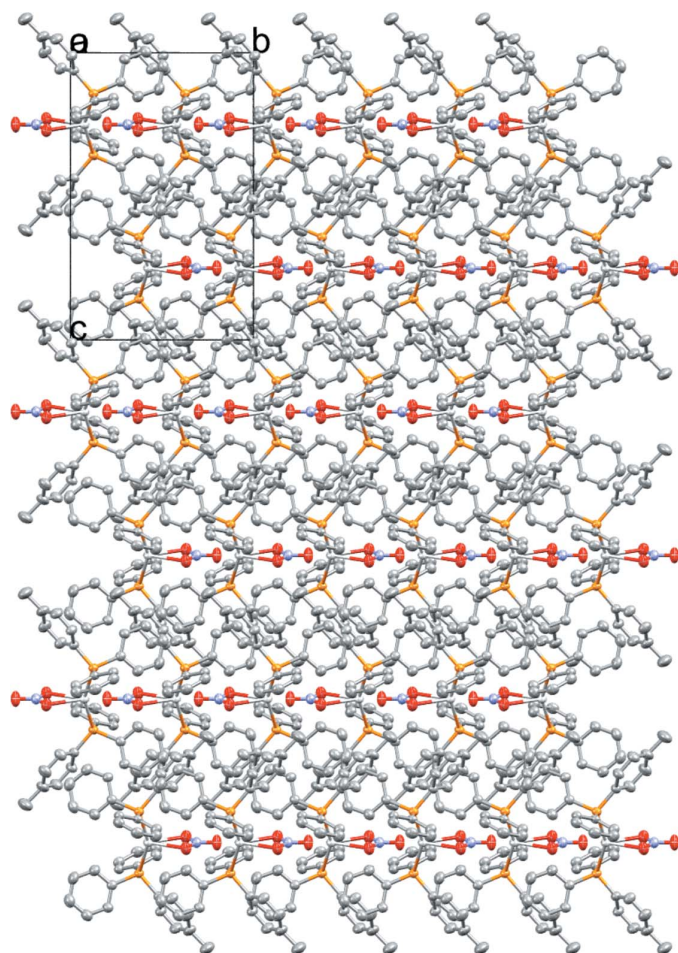


Figure 2
Packing diagram viewed along the *a*-axis.

Table 1
Experimental details.

Crystal data	
Chemical formula	[Ag(NO ₃)(C ₁₉ H ₁₇ P) ₂]
<i>M_r</i>	722.47
Crystal system, space group	Monoclinic, C2/c
Temperature (K)	150
<i>a</i> , <i>b</i> , <i>c</i> (Å)	18.1290 (6), 11.0936 (5), 17.9971 (7)
β (°)	104.849 (4)
<i>V</i> (Å ³)	3498.6 (2)
<i>Z</i>	4
Radiation type	Cu Kα
μ (mm ⁻¹)	5.77
Crystal size (mm)	0.21 × 0.18 × 0.10
Data collection	
Diffractometer	XtaLAB Synergy R, DW system, HyPix
Absorption correction	Multi-scan (CrysAlis PRO; Rigaku OD, 2022)
<i>T_{min}</i> , <i>T_{max}</i>	0.344, 1.000
No. of measured, independent and observed [<i>I</i> > 2σ(<i>I</i>)] reflections	20162, 3668, 3383
<i>R_{int}</i>	0.082
(sin θ/λ) _{max} (Å ⁻¹)	0.637
Refinement	
<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.048, 0.109, 1.06
No. of reflections	3668
No. of parameters	206
H-atom treatment	H-atom parameters constrained
Δρ _{max} , Δρ _{min} (e Å ⁻³)	0.60, −1.14

Computer programs: CrysAlis PRO (Rigaku OD, 2022), SHELXT (Sheldrick, 2015a), SHELXL (Sheldrick, 2015b) and OLEX2 (Dolomanov et al., 2009).

of the phosphine ligands. The corresponding torsion angles are P1ⁱ–Ag1–P1–C1 = −148.40 (15)°, P1ⁱ–Ag1–P1–C8 = −24.78 (17)° and P1ⁱ–Ag1–P1–C14 = 92.91 (14)°. The plane defined by atoms P1, Ag, P1ⁱ and N1 intercepts the plane defined by Ag1, O1, O1ⁱ and O2 at an angle of 72.33 (9)°.

In the crystal, the complex packs in three-dimensions as layers of isolated complexes. Within these layers a metal-containing NO₃ layer is observed, which alternates with dense arene-ring-filled layers. A view of the unit-cell contents is shown in Fig. 2.

Synthesis and crystallization

Diphenyl-*p*-tolylphosphine (2 mmol) and silver nitrate (1 mmol) were dissolved separately in acetonitrile (10 ml). The solutions were carefully mixed together and heated to 353 K for approximately 2 h. The solution was left to crystallize, and small clear crystals were obtained.

Refinement

For full experimental details including crystal data, data collection and structure refinement details, refer to Table 1. The maximum and minimum residual electron density peaks of 0.60 and 1.14 e Å⁻³ are located 1.29 and 0.83 Å, respectively, from the Ag1 atom, features ascribed to the presence of the strong absorber.

Acknowledgements

We would like to acknowledge the National Research Foundation (NRF, SA), the University of Pretoria and the University of Johannesburg for funding provided.

Funding information

Funding for this research was provided by: National Research Foundation (grant No. 138280).

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full crystallographic data

IUCrData (2022). 7, x221045 [https://doi.org/10.1107/S2414314622010458]

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Bis[(4-methylphenyl)diphenylphosphane- κP](nitrate- $\kappa^2 O, O'$)silver(I)*Crystal data*

[Ag(NO₃){C₁₉H₁₇P}₂]

$M_r = 722.47$

Monoclinic, $C2/c$

$a = 18.1290$ (6) Å

$b = 11.0936$ (5) Å

$c = 17.9971$ (7) Å

$\beta = 104.849$ (4)°

$V = 3498.6$ (2) Å³

$Z = 4$

$F(000) = 1480$

$D_x = 1.372$ Mg m⁻³

Cu $K\alpha$ radiation, $\lambda = 1.54184$ Å

Cell parameters from 14857 reflections

$\theta = 4.7\text{--}78.5^\circ$

$\mu = 5.77$ mm⁻¹

$T = 150$ K

Blade, colourless

0.21 × 0.18 × 0.10 mm

Data collection

XtaLAB Synergy R, DW system, HyPix
diffractometer

Radiation source: Rotating-anode X-ray tube,
Rigaku (Cu) X-ray Source

Mirror monochromator

Detector resolution: 10.0000 pixels mm⁻¹

ω scans

Absorption correction: multi-scan
(CrysAlisPro; Rigaku OD, 2022)

$T_{\min} = 0.344$, $T_{\max} = 1.000$

20162 measured reflections

3668 independent reflections

3383 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.082$

$\theta_{\max} = 79.2^\circ$, $\theta_{\min} = 4.7^\circ$

$h = -22 \rightarrow 23$

$k = -13 \rightarrow 12$

$l = -22 \rightarrow 22$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.048$

$wR(F^2) = 0.109$

$S = 1.06$

3668 reflections

206 parameters

0 restraints

Primary atom site location: dual

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.029P)^2 + 20.5034P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.60$ e Å⁻³

$\Delta\rho_{\min} = -1.13$ e Å⁻³

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Ag1	0.5000	0.42530 (4)	0.7500	0.03274 (12)
P1	0.39526 (5)	0.37440 (9)	0.64210 (5)	0.0295 (2)
O1	0.55106 (15)	0.6308 (3)	0.72975 (18)	0.0428 (7)
O2	0.5000	0.7999 (4)	0.7500	0.0573 (12)
N1	0.5000	0.6886 (4)	0.7500	0.0361 (10)
C8	0.4038 (2)	0.2263 (4)	0.6014 (2)	0.0332 (8)
C14	0.3048 (2)	0.3676 (3)	0.6690 (2)	0.0317 (8)
C1	0.3777 (2)	0.4809 (4)	0.5629 (2)	0.0329 (8)
C13	0.3896 (2)	0.2044 (4)	0.5231 (2)	0.0378 (9)
H13	0.3746	0.2688	0.4877	0.045*
C18	0.1918 (2)	0.2650 (4)	0.6813 (2)	0.0405 (9)
H18	0.1604	0.1952	0.6736	0.049*
C19	0.2580 (2)	0.2661 (4)	0.6569 (2)	0.0349 (8)
H19	0.2717	0.1975	0.6318	0.042*
C15	0.2837 (2)	0.4685 (4)	0.7050 (2)	0.0407 (9)
H15	0.3152	0.5381	0.7136	0.049*
C17	0.1711 (2)	0.3650 (4)	0.7169 (2)	0.0448 (10)
H17	0.1254	0.3639	0.7336	0.054*
C2	0.3067 (2)	0.4939 (4)	0.5108 (2)	0.0426 (9)
H2	0.2650	0.4456	0.5159	0.051*
C6	0.4374 (2)	0.5548 (4)	0.5552 (3)	0.0449 (10)
H6	0.4863	0.5470	0.5902	0.054*
C3	0.2965 (3)	0.5765 (5)	0.4519 (3)	0.0538 (12)
H3	0.2481	0.5826	0.4159	0.065*
C9	0.4264 (3)	0.1314 (4)	0.6525 (3)	0.0466 (10)
H9	0.4370	0.1457	0.7062	0.056*
C12	0.3973 (3)	0.0894 (4)	0.4965 (3)	0.0496 (11)
H12	0.3874	0.0754	0.4428	0.060*
C16	0.2163 (2)	0.4664 (4)	0.7282 (3)	0.0456 (10)
H16	0.2014	0.5353	0.7521	0.055*
C4	0.3559 (3)	0.6513 (5)	0.4439 (3)	0.0533 (11)
C11	0.4192 (3)	-0.0056 (4)	0.5472 (3)	0.0552 (12)
H11	0.4242	-0.0845	0.5285	0.066*
C5	0.4258 (3)	0.6397 (5)	0.4970 (3)	0.0555 (12)
H5	0.4667	0.6909	0.4934	0.067*
C10	0.4338 (3)	0.0157 (4)	0.6257 (3)	0.0540 (11)
H10	0.4488	-0.0489	0.6610	0.065*
C7	0.3448 (4)	0.7468 (6)	0.3819 (3)	0.0817 (19)
H7A	0.3270	0.7088	0.3313	0.123*
H7B	0.3934	0.7878	0.3852	0.123*
H7C	0.3069	0.8055	0.3891	0.123*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Ag1	0.02296 (18)	0.0399 (2)	0.0335 (2)	0.000	0.00388 (14)	0.000
P1	0.0214 (4)	0.0375 (5)	0.0295 (4)	-0.0018 (3)	0.0064 (3)	0.0016 (4)
O1	0.0279 (13)	0.0427 (16)	0.0639 (18)	-0.0022 (12)	0.0233 (13)	-0.0038 (14)
O2	0.045 (3)	0.033 (2)	0.095 (4)	0.000	0.020 (2)	0.000
N1	0.026 (2)	0.035 (2)	0.045 (3)	0.000	0.0062 (19)	0.000
C8	0.0216 (16)	0.041 (2)	0.0380 (19)	-0.0008 (14)	0.0088 (14)	0.0016 (16)
C14	0.0285 (17)	0.040 (2)	0.0271 (16)	-0.0037 (15)	0.0079 (14)	0.0028 (15)
C1	0.0276 (17)	0.0384 (19)	0.0331 (18)	-0.0005 (15)	0.0084 (14)	0.0022 (15)
C13	0.0292 (18)	0.045 (2)	0.038 (2)	0.0015 (16)	0.0074 (16)	-0.0025 (17)
C18	0.0287 (19)	0.052 (2)	0.042 (2)	-0.0087 (17)	0.0118 (16)	0.0056 (18)
C19	0.0301 (18)	0.042 (2)	0.0329 (18)	-0.0043 (16)	0.0082 (15)	0.0011 (16)
C15	0.0294 (19)	0.048 (2)	0.047 (2)	-0.0053 (17)	0.0130 (17)	-0.0059 (19)
C17	0.031 (2)	0.062 (3)	0.046 (2)	-0.0012 (19)	0.0179 (18)	0.004 (2)
C2	0.033 (2)	0.050 (2)	0.041 (2)	-0.0052 (18)	0.0032 (17)	0.0079 (18)
C6	0.0280 (19)	0.056 (3)	0.051 (2)	0.0009 (18)	0.0112 (17)	0.013 (2)
C3	0.046 (3)	0.064 (3)	0.045 (2)	0.002 (2)	0.000 (2)	0.013 (2)
C9	0.042 (2)	0.049 (2)	0.047 (2)	0.0025 (19)	0.0088 (19)	0.0061 (19)
C12	0.043 (2)	0.053 (3)	0.051 (2)	0.004 (2)	0.009 (2)	-0.009 (2)
C16	0.036 (2)	0.054 (3)	0.051 (2)	-0.0006 (19)	0.0193 (19)	-0.005 (2)
C4	0.056 (3)	0.062 (3)	0.043 (2)	0.002 (2)	0.014 (2)	0.015 (2)
C11	0.045 (3)	0.047 (3)	0.072 (3)	0.003 (2)	0.013 (2)	-0.009 (2)
C5	0.046 (3)	0.065 (3)	0.058 (3)	-0.006 (2)	0.018 (2)	0.018 (2)
C10	0.050 (3)	0.046 (3)	0.065 (3)	0.007 (2)	0.013 (2)	0.012 (2)
C7	0.096 (5)	0.083 (4)	0.058 (3)	-0.010 (4)	0.003 (3)	0.025 (3)

Geometric parameters (Å, °)

Ag1—P1 ⁱ	2.4095 (9)	C1—C6	1.393 (5)
Ag1—P1	2.4095 (9)	C13—C12	1.383 (6)
Ag1—O1 ⁱ	2.522 (3)	C18—C19	1.380 (5)
Ag1—O1	2.522 (3)	C18—C17	1.380 (6)
P1—C8	1.822 (4)	C15—C16	1.388 (5)
P1—C14	1.827 (4)	C17—C16	1.375 (6)
P1—C1	1.815 (4)	C2—C3	1.377 (6)
O1—N1	1.255 (3)	C6—C5	1.384 (6)
O2—N1	1.235 (6)	C3—C4	1.395 (7)
N1—O1 ⁱ	1.255 (3)	C9—C10	1.390 (7)
C8—C13	1.386 (5)	C12—C11	1.383 (7)
C8—C9	1.389 (6)	C4—C5	1.385 (7)
C14—C19	1.393 (5)	C4—C7	1.514 (7)
C14—C15	1.395 (6)	C11—C10	1.389 (7)
C1—C2	1.392 (5)		
P1 ⁱ —Ag1—P1	152.89 (5)	C15—C14—P1	117.5 (3)
P1—Ag1—O1	109.76 (7)	C2—C1—P1	122.9 (3)

P1 ⁱ —Ag1—O1 ⁱ	109.76 (7)	C2—C1—C6	118.6 (4)
P1 ⁱ —Ag1—O1	94.92 (7)	C6—C1—P1	118.5 (3)
P1—Ag1—O1 ⁱ	94.92 (7)	C12—C13—C8	120.3 (4)
O1—Ag1—O1 ⁱ	50.63 (12)	C19—C18—C17	120.2 (4)
C8—P1—Ag1	113.90 (12)	C18—C19—C14	120.1 (4)
C8—P1—C14	104.11 (17)	C16—C15—C14	119.6 (4)
C14—P1—Ag1	111.84 (12)	C16—C17—C18	120.2 (4)
C1—P1—Ag1	115.07 (12)	C3—C2—C1	120.4 (4)
C1—P1—C8	106.68 (18)	C5—C6—C1	120.4 (4)
C1—P1—C14	104.22 (17)	C2—C3—C4	121.3 (4)
N1—O1—Ag1	95.4 (2)	C8—C9—C10	120.5 (4)
O1—N1—O1 ⁱ	118.5 (4)	C13—C12—C11	120.8 (4)
O2—N1—O1	120.8 (2)	C17—C16—C15	120.4 (4)
O2—N1—O1 ⁱ	120.8 (2)	C3—C4—C7	122.2 (5)
C13—C8—P1	123.6 (3)	C5—C4—C3	117.9 (4)
C13—C8—C9	119.1 (4)	C5—C4—C7	119.9 (5)
C9—C8—P1	117.3 (3)	C12—C11—C10	119.3 (4)
C19—C14—P1	122.9 (3)	C6—C5—C4	121.3 (4)
C19—C14—C15	119.5 (3)	C11—C10—C9	120.0 (4)

Symmetry code: (i) $-x+1, y, -z+3/2$.