

Interactions of Coated-Gold Engineered Nanoparticles with Aquatic Higher Plant *Salvinia minima* Baker

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The ζ potentials were calculated using Smoluchowski equation:

$$\nu E = 4\pi\epsilon_0\epsilon r + \frac{\zeta}{6\pi\mu} + (1 + \kappa r) \quad (\text{Eq. S1})$$

where ϵ_0 and ϵr are the relative dielectric constant and the electrical permittivity of a vacuum respectively, μ is the solution viscosity, r is the particle radius and $\kappa = (2n_0z^2e^2/\epsilon r\epsilon_0k_B T)^{1/2}$ is the Debye-Hückel parameter, n_0 is the bulk ionic concentration, z is the valence of the ion, e is the charge of an electron, k_B is the Boltzmann constant, and T is the absolute temperature.

The ionic strength (IS) of the medium was calculated as follows:

$$\text{IS} = \frac{1}{2} \sum_i C_i Z_i^2 \quad (\text{Eq. S2})$$

where IS is the ionic strength in mM, C_i is the concentration of the i^{th} species in mM, and Z_i is the charge of the i^{th} species.

Table S1: Composition of Hoagland's medium

Macronutrient	Per litre of nutrient solution
KNO ₃	5 ml of 1 M
Ca(NO ₃) ₂ ·4H ₂ O	5 ml of 1 M
MgSO ₄ ·7(H ₂ O)	2 ml of 1 M
KH ₂ PO ₄	2 ml of 1 M
Micronutrients	Per litre (g)
H ₃ BO ₃	2.86
MnCl ₂ ·4(H ₂ O)	1.81
ZnSO ₄ ·7H ₂ O	0.22
CuSO ₄ ·5H ₂ O	0.08
MoO ₃	0.02
Fe-EDTA	1-5 ml of 1000 mg/L
Minus nitrogen	Per litre of nutrient solution
Ca(NO ₃) ₂ ·4H ₂ O	0.75 ml of 1 M
Ca(H ₂ PO ₄) ₂ ·H ₂ O	10 ml of 0.05 M
CaSO ₄ ·2H ₂ O	200 ml of 0.01 M
K ₂ SO ₄	5 ml of 0.5 M
MgSO ₄ ·7H ₂ O	2 ml of 1 M

Source: Hoagland and Arnon [1]

Table S2: Mean sizes (nm) of nAu obtained using TEM

Sample	Size (nm) (Manufacturer)	Size (nm) (Current study)
5 nm cit-nAu	4.8±0.7	5.6±1.4
20 nm cit-nAu	20.0±2.5	20.7±2.5
40 nm cit-nAu	39.9±4.1	41.3±4.0
5 nm BPEI-nAu	4.3±0.6	5.0±0.6
20 nm BPEI-nAu	18.2±1.6	19.0±2.7
40 nm BPEI-nAu	41.3±4.2	41.3±2.8

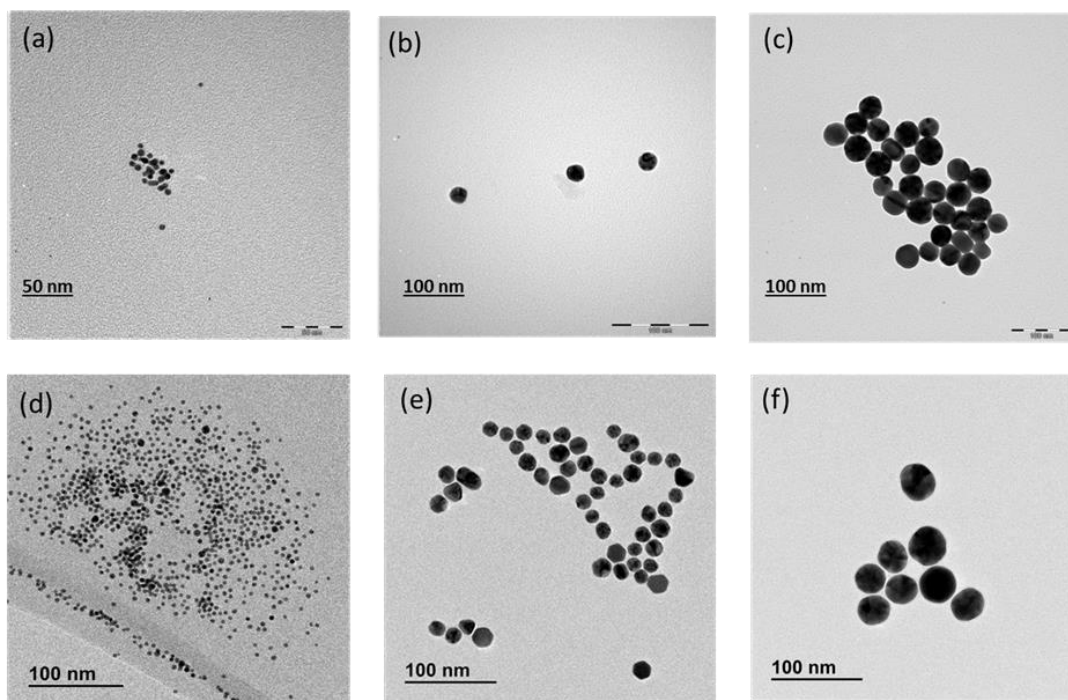


Figure S1. TEM images of nAu (a) 5 nm-Cit, (b) 20 nm-Cit, (c) 40 nm-Cit, (d) 5 nm-BPEI, (e) 20 nm-BPEI, (f) and 40 nm-BPEI. Source: Mahaye [2].

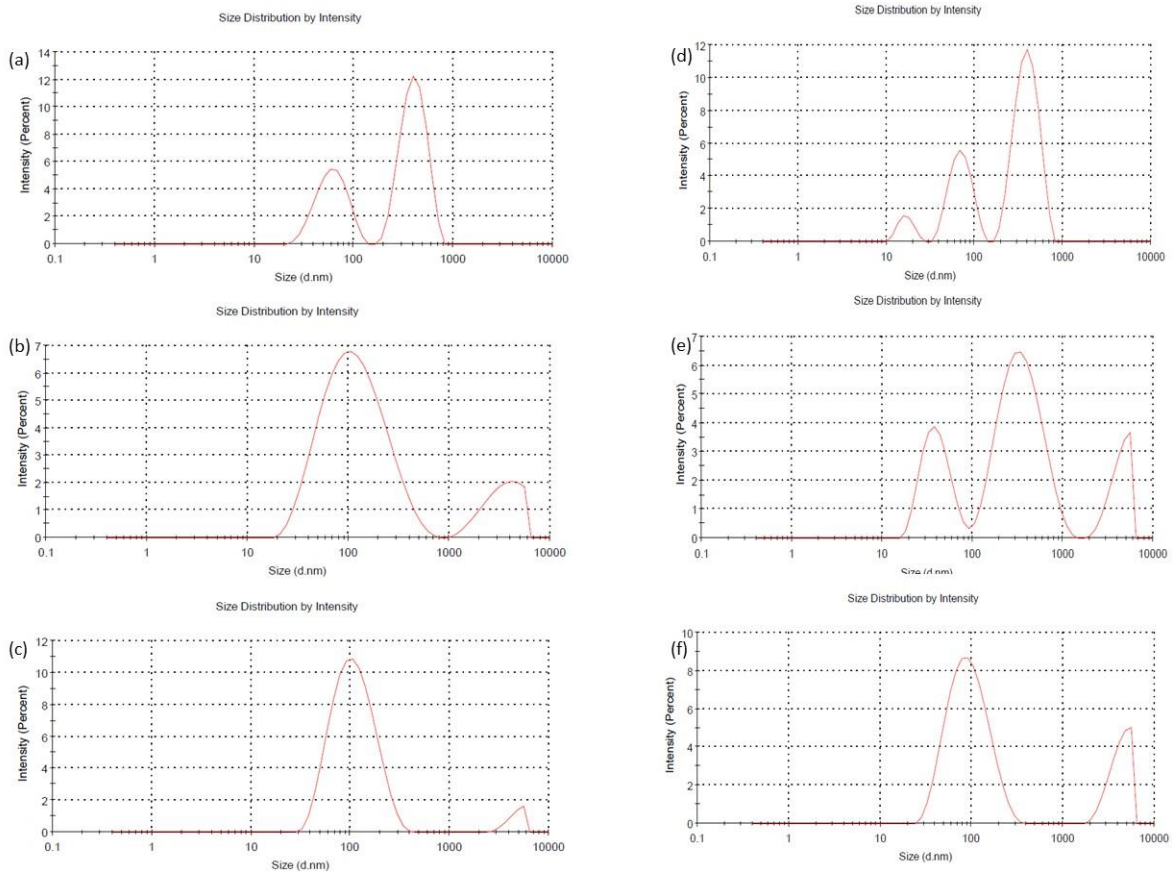


Figure S2: Particle size distribution of gold nanoparticles at 1000 $\mu\text{g/L}$ in 10% Hoagland's medium measured using Dynamic Light Scattering technique at 0 h (a) 5 nm Cit-nAu, (b) 20 nm Cit-nAu, (c) 40 nm Cit-nAu, (d) 5 nm BPEI-nAu, (e) 20 nm

BPEI-nAu, and (f) 40 nm BPEI-nAu.

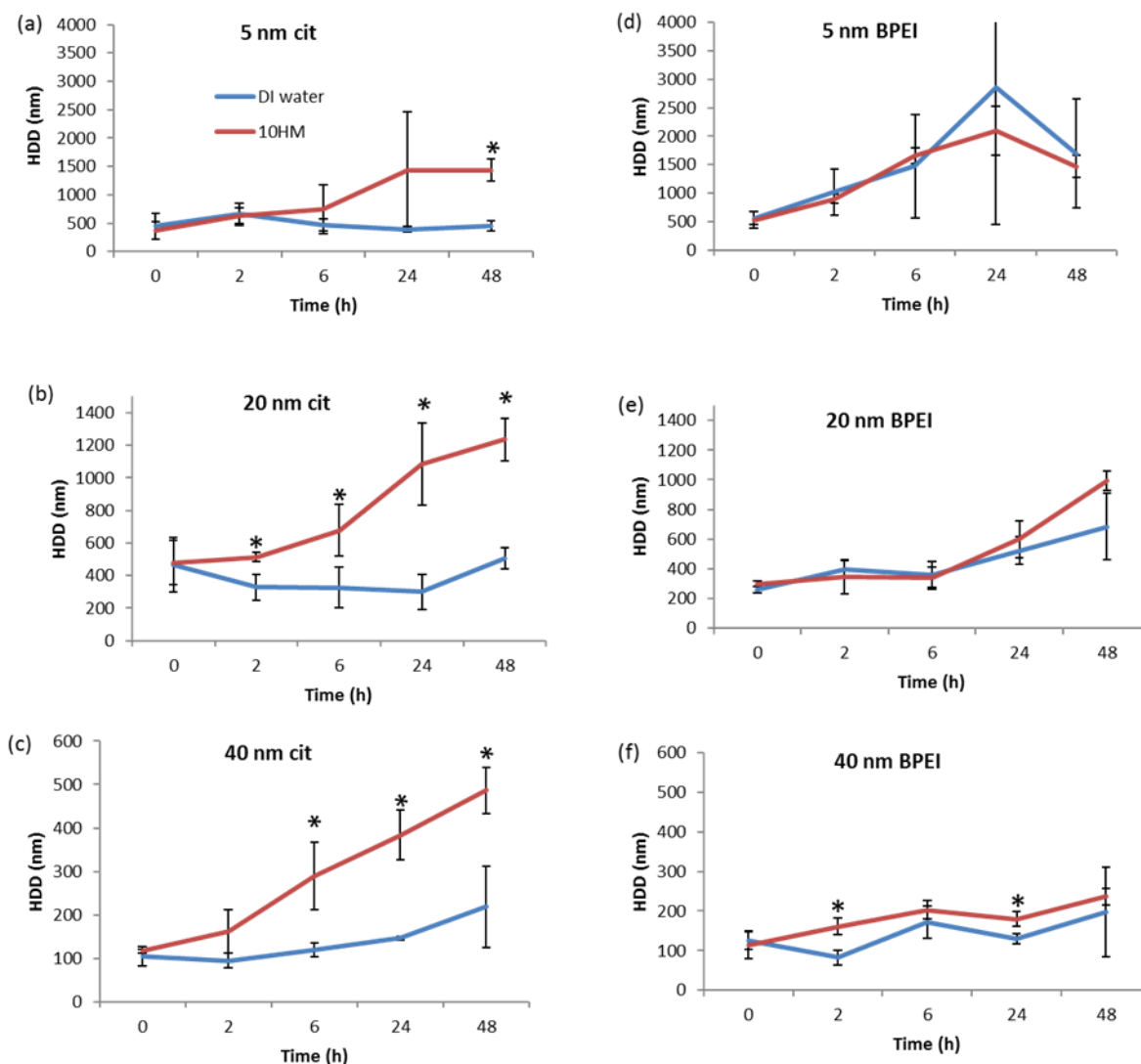


Figure S3: Hydrodynamic diameters of nAu in de-ionized water and 10% Hoagland's medium tracked using Dynamic Light Scattering technique over 48 h. (a) 5 nm Cit-nAu, (b) 20 nm Cit-nAu, (c) 40 nm Cit-nAu, (d) 5 nm BPEI-nAu, (e) 20 nm BPEI-nAu, and (f) 40 nm BPEI-nAu. Bars denote standard deviations, and statistical difference between the two media per time period is denoted by * where $n = 3$.

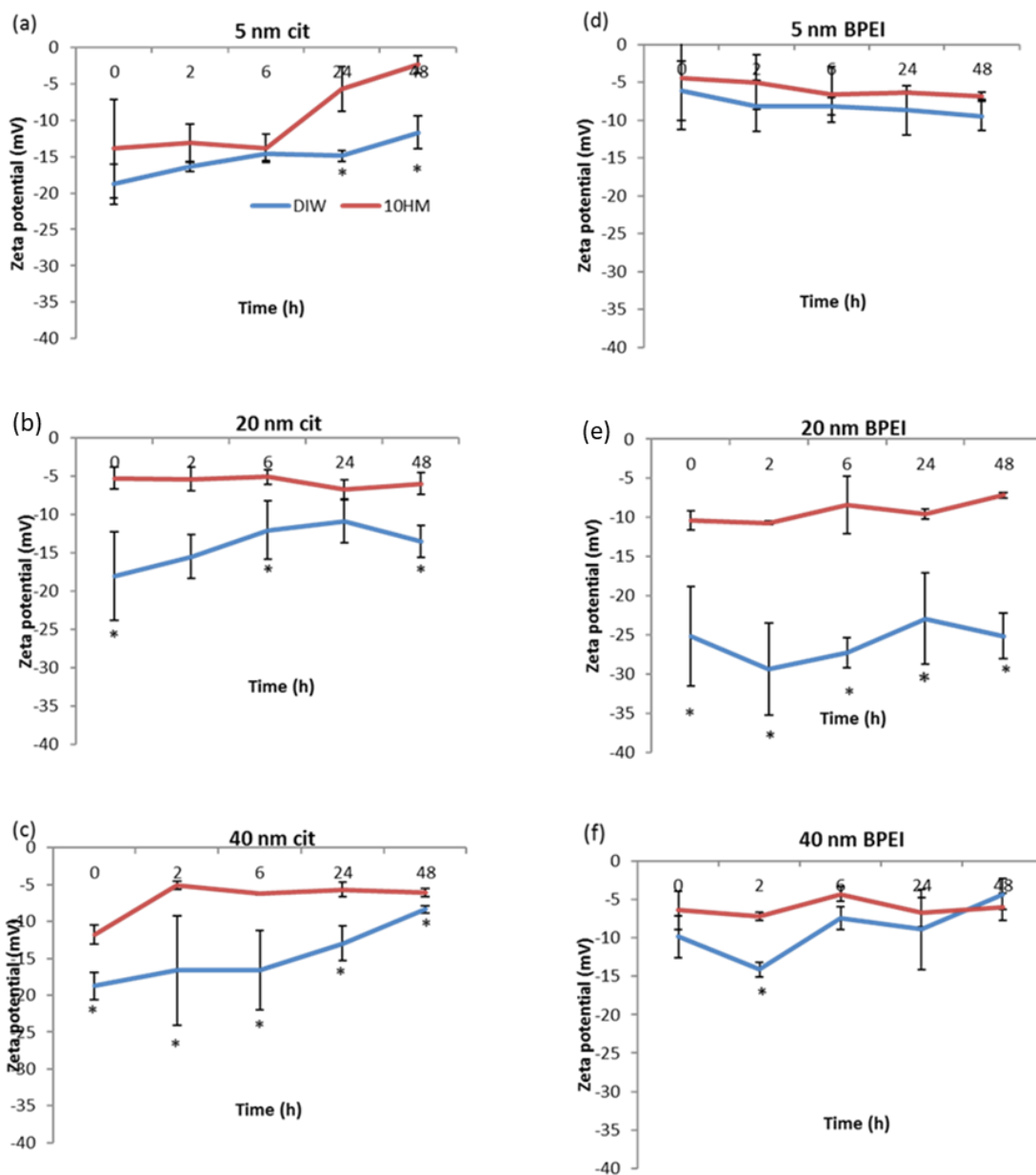


Figure S4: Zeta potentials of nAu in de-ionized water and 10% Hoagland's medium obtained using Dynamic Light Scattering technique over 48 h. (a) 5 nm Cit-nAu, (b) 20 nm Cit-nAu, (c) 40 nm Cit-nAu, (d) 5 nm BPEI-nAu, (e) 20 nm BPEI-nAu, and (f) 40 nm BPEI-nAu. Bars denote standard deviations, and statistical difference between the two media per time period is denoted by * where $n = 3$.

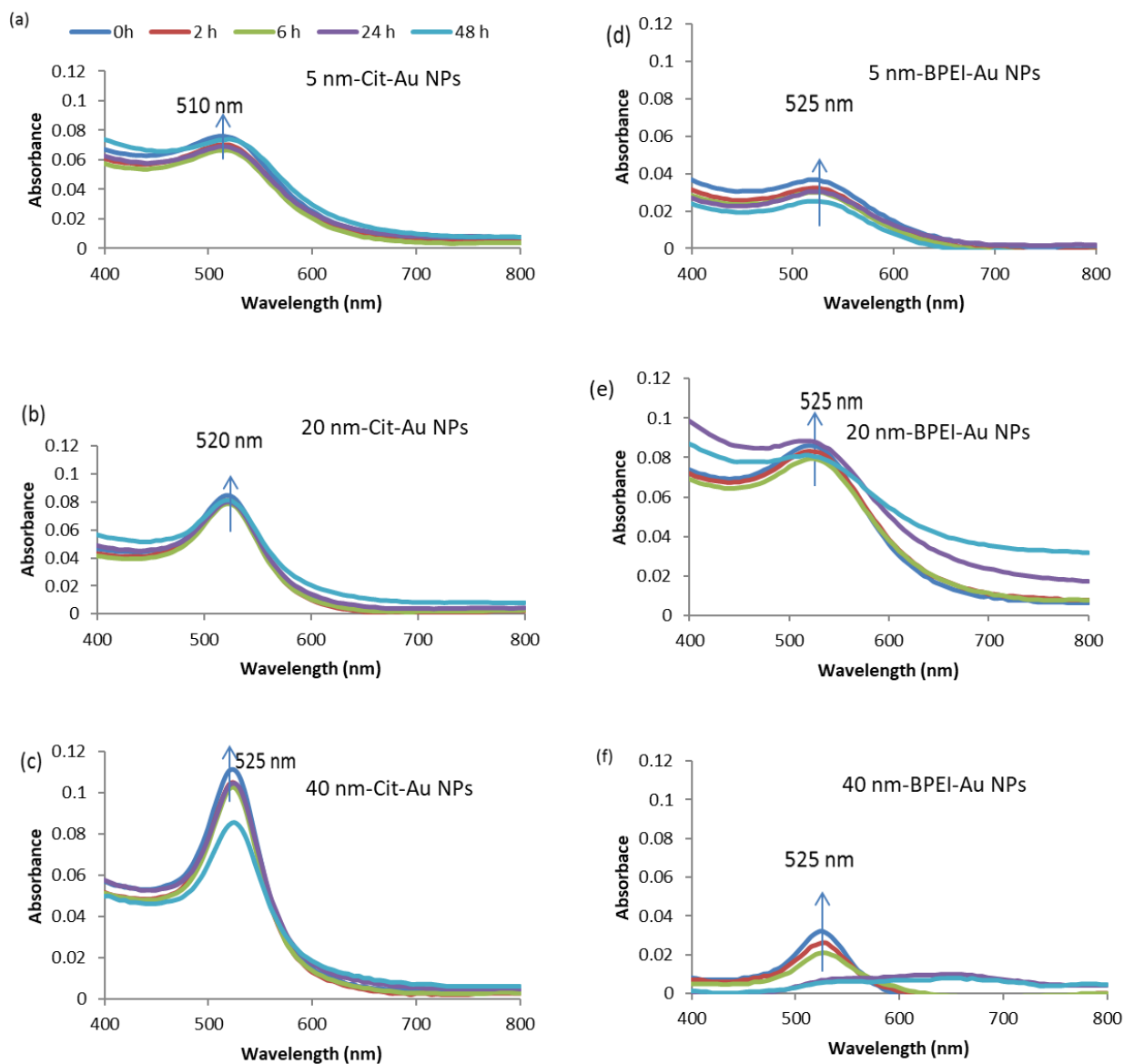


Figure S5: UV-vis spectrum of nAu in de-ionized water as a function of time. (a) 5 nm Cit-nAu, (b) 20 nm Cit-nAu, (c) 40 nm Cit-nAu, (d) 5 nm BPEI-nAu, (e) 20 nm BPEI-nAu, and (f) 40 nm BPEI-nAu. Data are presented as means ($n = 3$). Arrows show the position of the main peak. Source: Mahaye [2].

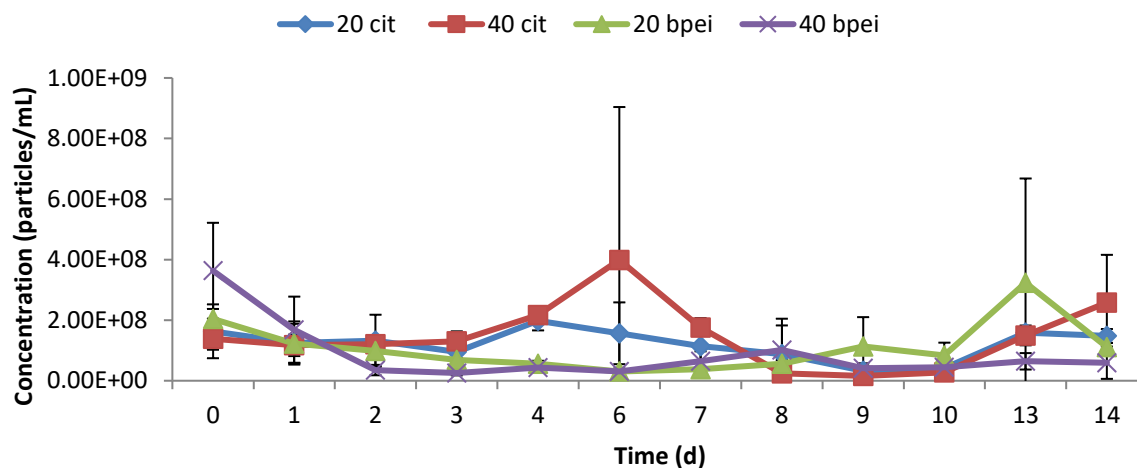


Figure S6: *in situ* nAu concentration (particles/mL) examined using Nanoparticle Tracking Analysis (NanoSight NS500, NTA 3.0 Software, Amesbury, UK). Results are represented as mean ($n=3$) and bars denote standard deviation. The 5 nm sized-nAu samples were excluded from NTA analysis as the technique cannot measure sizes below 10 nm accurately (NanoSight NS500 operating manual, Version P553S).

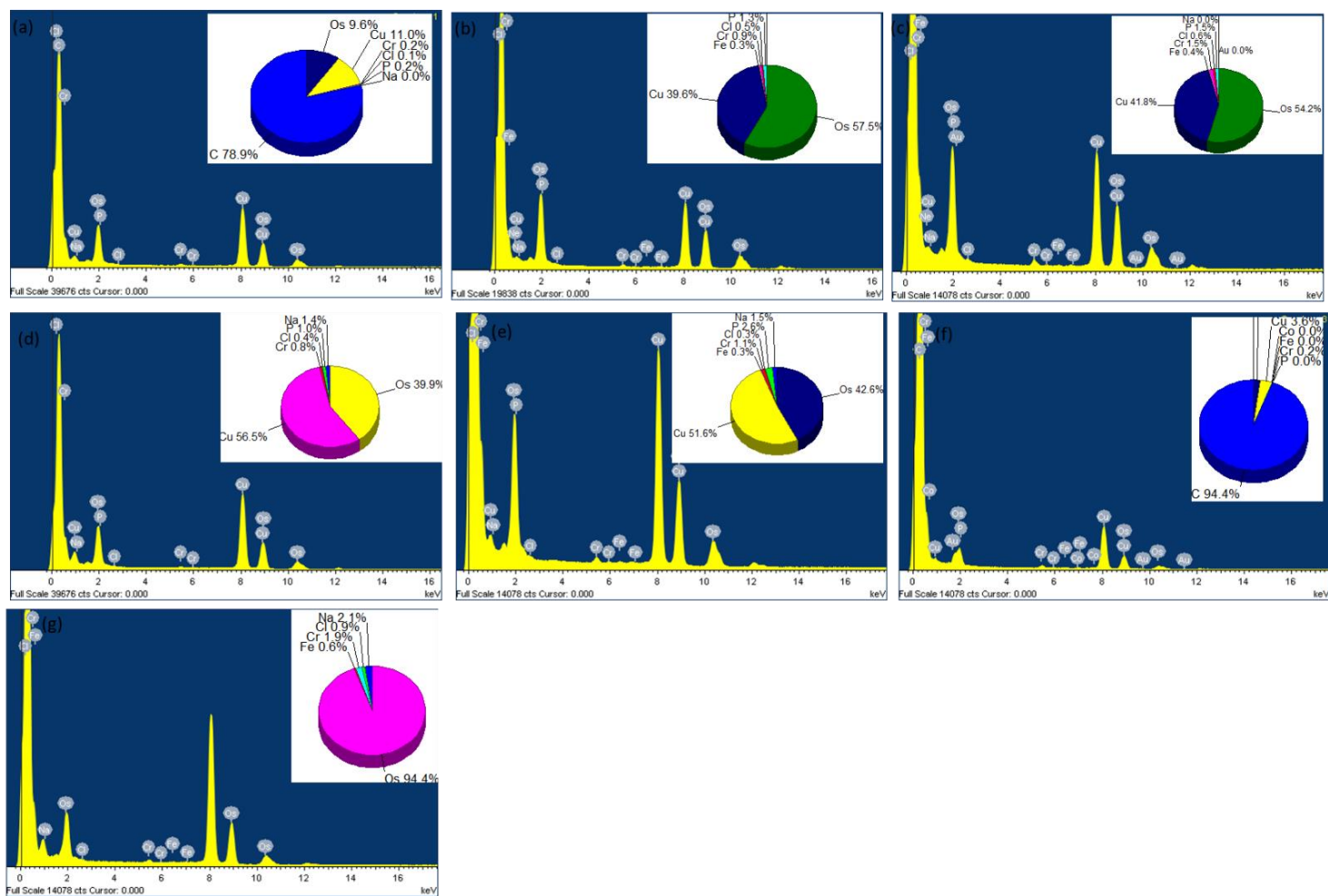


Figure S7: TEM-EDX spectra confirming the absence of nAu internalization on plant roots: (a) control, (b) 5 nm cit-nAu, (c) 20 nm-cit nAu, (d) 40 nm cit-nAu, (e) 5 nm BPEI-nAu, (f) 20 nm BPEI-nAu, and (g) 40 nm BPEI. Peaks indicate that only iron, copper, sodium, osmium, chromium, chlorine, and phosphorus were identified. The inserts on Figures indicate percentages of the detected elements.

References

1. Hoagland DR; Arnon DI. The Water-Culture Method for Growing Plants without Soil. *California Agricultural Experiment Station Circular* **1950**, 347, 1–32.
2. Mahaye, N. Stability of Gold and Cerium Oxide Nanoparticles in Aqueous Environments, and Their Effects on *Pseudokirchneriella Subcapitata* and *Salvinia Minima*. PhD, University of Pretoria: Pretoria, **2019**. <https://repository.up.ac.za/handle/2263/72778>.

