## Composition-dependent structure evolution of FeVO<sub>4</sub> nano-oxide and its visible-light photocatalytic activity for degradation of methylene blue Supplementary material

Kgabo P. Thaba,<sup>1,2</sup> Mabel M. Mphahlele-Makgwane,<sup>1,2</sup> \* Pannan I. Kyesmen,<sup>3</sup> Mmantsae Diale,<sup>3</sup> Priscilla G. Baker, <sup>5</sup> Peter R. Makgwane <sup>4,5</sup> \*

<sup>1</sup> Department of Water and Sanitation, Private Bag X1106, University of Limpopo, Sovenga, 0728, South Africa.

<sup>2</sup> Department of Chemistry, Private Bag X1106, University of Limpopo, Sovenga, 0728, South Africa.

<sup>3</sup> Department of Physics, University of Pretoria, Private Bag X20, Hatfield 0028, South Africa <sup>4</sup>Centre for Nanostructures and Advanced Materials (CeNAM), Council for Scientific and Industrial Research (CSIR), Pretoria, 0001, South Africa.

<sup>5</sup> Department of Chemistry, University of the Western Cape, Bag X17, Robert Sobukwe Drive, Bellville, 7535, South Africa.

\*Correspondence authors:

Email: mabel.mphahlele-makgwane@ul.ac.za; makgwane.peter@gmail.com

Tel: +27152033498; Fax: +27152035979



Figure S1. XRD patterns of the  $V_2O_5$  powder catalyst.



Figure S2. Elemental mapping and EDX spectra of  $V_2O_5$ .

		-	-	
Element Line	Weight (%)	Weight (%) Error	Atom (%)	Atom (%) Error
O K	31.73	± 2.64	59.67	$\pm 4.97$
V K	68.27	$\pm 0.37$	40.33	$\pm 0.22$
V L				
Total	100.00		100.00	
Total	100.00	1	100.00	

Table S1. Elemental compositions of V<sub>2</sub>O<sub>5</sub> catalyst.





Figure S3. Elemental mapping and EDX spectra of Fe<sub>2</sub>O<sub>3</sub>.

Element Line	Weight (%)	Weight (%) Error	Atom (%)	Atom (%) Error
O K	27.15	$\pm 0.17$	56.54	± 0.34
Fe K	72.85	$\pm 0.35$	43.46	± 0.21
Fe L				
Total	100.00		100.00	

**Table S2.** Elemental compositions of Fe2O3 catalyst.



Figure S4. Elemental mapping and EDX spectra of FeV-4.

Elamont Lina	$W_{ai} = h t (0/)$	Waisht (0/) Emer	$\Lambda$ to $(0/)$	$\Lambda$ to $(0/)$ Eulon
Element Line	weight (%)	weight (%) Error	Aiom(%)	Atom (%) Error
O K	28 94	+0.32	57 59	+0.64
0 11	20.91	- 0.52	01.09	- 0.01
VK	34 72	+0.22	21.70	+0.14
V IX	54.72	$\pm 0.22$	21.70	$\pm 0.14$
V/ I				
V L				
	26.22	<u> </u>	<b>a</b> a <b>a</b> 1	<u> </u>
Fe K	36.33	$\pm 0.34$	20.71	$\pm 0.20$
Fe L				
Total	100.00		100.00	
1 Juli	100.00		100.00	
Total	100.00		100.00	

 Table S3. Elemental compositions of FeV-4 catalyst.



Figure S4. Elemental mapping and EDX spectra of FeV-3.

Element Line	Weight (%)	Weight (%) Error	Atom (%)	Atom (%) Error
ОК	23.24	± 0.18	50.55	± 0.39
V K	27.07	± 0.19	18.49	± 0.13
V L				
Fe K	49.69	± 0.37	30.96	± 0.23
Fe L				
Total	100.00		100.00	

 Table S4. Elemental compositions of FeV-3 catalyst.



Figure S5. Elemental mapping and EDX spectra of FeV-2.

Element Line	Weight (%)	Weight (%) Error	Atom (%)	Atom (%) Error
O K	33.21	± 0.19	62.96	$\pm 0.37$
V K	14.70	± 0.16	8.75	± 0.09
V L				
Fe K	52.08	± 0.28	28.29	± 0.15
Fe L				
Total	100.00		100.00	

**Table S5**. Elemental compositions of FeV-2 catalyst.



Figure S6. Elemental mapping and EDX spectra of FeV-1.

Element Line	Weight (%)	Weight (%) Error	Atom (%)	Atom (%) Error
O K	29.90	$\pm 0.33$	59.44	$\pm 0.65$
Al K	0.42	$\pm 0.05$	0.50	$\pm 0.06$
V K	7.02	± 0.21	4.39	± 0.13
V L				
Fe K	62.65	$\pm 0.72$	35.68	± 0.41
Fe L				
Total	100.00		100.00	

 Table S6. Elemental compositions of FeV-1 catalyst.



Fig. S8. XPS survey of the FeVO<sub>4</sub> heterostructure catalysts.



Figure S9. XPS O 1s profiles of Fe<sub>2</sub>O<sub>3</sub> and FeVO<sub>4</sub> heterostructure catalysts.



**Figure S10**. Effect of methylene blue (MB) concentration of photo-degradation activity of FeV-3 catalyst.



Figure S11. Effect adsorption-equilibration (induction time) on photodegradation activity of FeV-3 catalyst.