

Electrochemical performance of two-dimensional $\text{Ti}_3\text{C}_2\text{-Mn}_3\text{O}_4$ nanocomposites and carbonized iron cations for hybrid supercapacitor electrodes

Kabir O. Oyedotun¹, Damilola Y. Momodu¹, Michael Naguib², Abdulmajid A. Mirghni¹, Tshifhiwa M. Masikhwa¹, Abubakar A. Khaleed¹, Mesfin Kebede³, and Ncholu Manyala^{1*}

¹Department of Physics, Institute of Applied Materials, SARCHI Chair in Carbon Technology and Materials, University of Pretoria, Pretoria 0028, South Africa.

²Department of Physics and Engineering Physics, Tulane University, New Orleans, LA 70118, USA.

³Energy Materials, Materials Science and Manufacturing, Council for Scientific and Industrial Research (CSIR), Pretoria, 0001, South Africa

*Corresponding author's email: ncholu.manyala@up.ac.za, Tel.: + (27)12 420 3549.

SUPPORTING INFORMATION

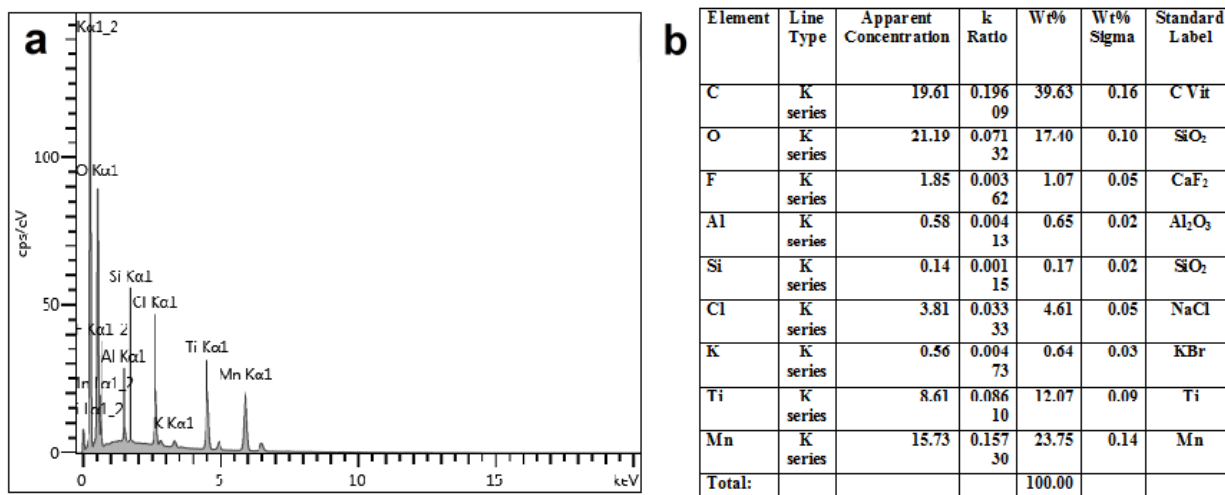


Fig. S1. (a) EDX spectrum, and (b) table of chemical analysis of $\text{Ti}_3\text{C}_2\text{-Mn}_3\text{O}_4$ nanocomposite respectively.

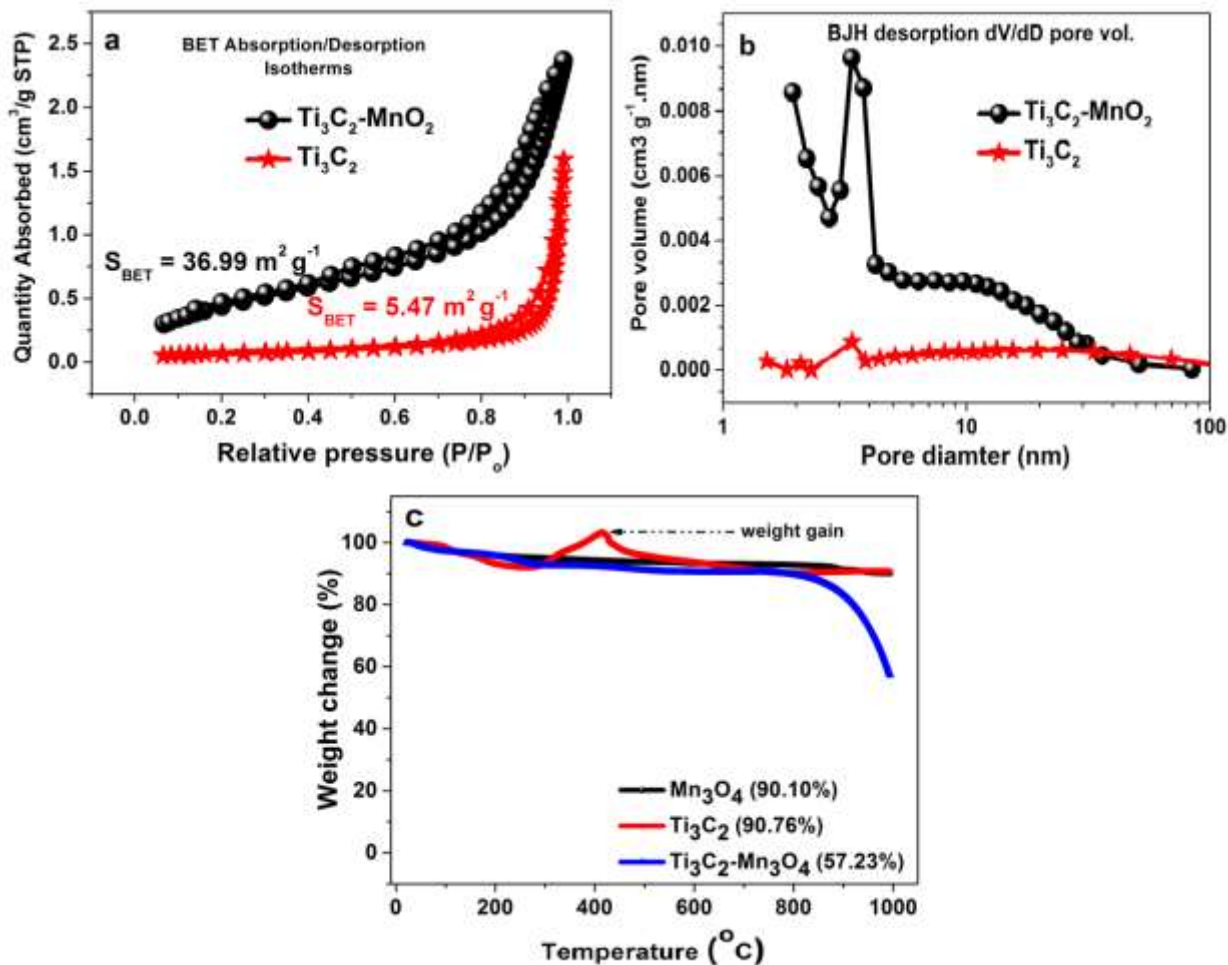


Fig. S2. (a) N_2 isotherms and (b) pore size distribution of $\text{Ti}_3\text{C}_2\text{-Mn}_3\text{O}_4$, and (c) Thermogravimetric analysis curves of Ti_3C_2 , Mn_3O_4 and $\text{Ti}_3\text{C}_2\text{-Mn}_3\text{O}_4$ nanocomposite.

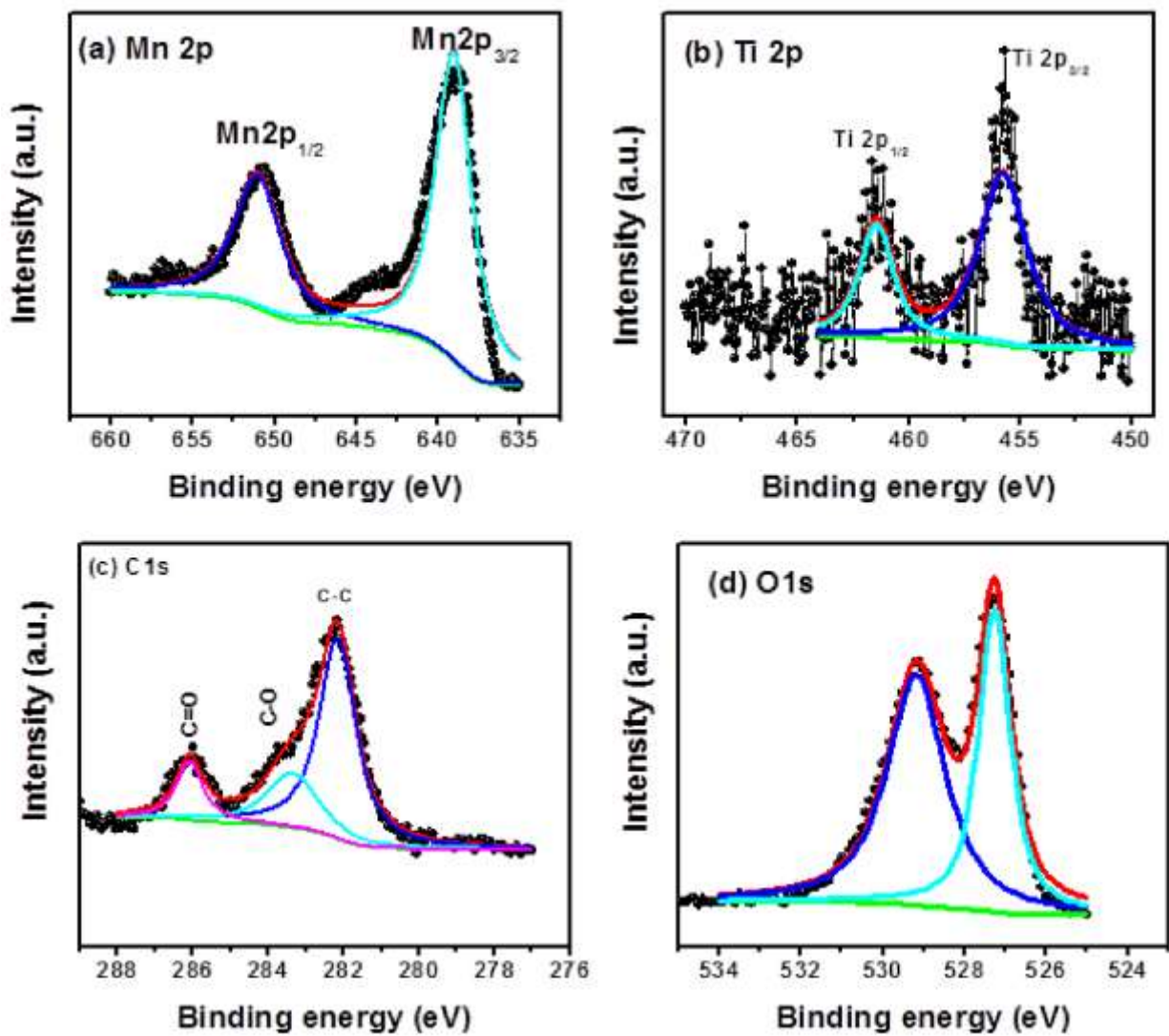


Fig. S3. The core level spectrum of (a) Mn 2p, (b) Ti 2p, (c) C 1s and (d) O 1s of a Ti₃C₂-Mn₃O₄ nanocomposite, respectively.

Table S1: Showing XRF analysis of as-synthesized $\text{Ti}_2\text{C}_3\text{-Mn}_3\text{O}_4$ nanocomposite.

Compound	Weight %
Al_2O_3	2.41
K_2O	0.80
Na_2O	0.67
MgO	0.23
Sm_2O_3	0.13
Tb_4O_7	0.10
SiO_2	0.07
V_2O_5	0.05
Gd_2O_3	0.03
Cl	0.03
Co_3O_4	0.03
CaO	0.02
P	0.02
Au	0.01
MoO_3	0.01
La_2O_3	0.01
PtO_2	0.01
% Mn_3O_4	79.83
Remaining % Ti_3C_2	15.56
TOTAL	100.00