Enantioanalysis of pharmaceutical compounds

by

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SYNOPSIS

Due to the differences in pharmacokinetics and pharmacodynamics of the enantiomers of the

same chiral pharmaceutical substance, there is a high need of reliable analytical methods for

enantiopurity tests of them in the raw material as well as in its pharmaceutical formulations.

If some of the chiral pharmaceutical compounds can be delivered as racemates, there are

many others for which the enantiopurity is essential. Enantioanalysis using enantioselective,

potentiometric membrane electrodes became a good alternative of the chromatographic

methods due to its high reliability.

To have reliable analytical information it is necessary to use reliable analytical methods and

electrodes. The most reliable design for the enantioselective, potentiometric membrane

electrodes proved to be the one based on carbon paste. This electrodes are made by mixing

graphite powder with paraffin oil to give carbon paste, which is modified by the addition of a

chiral selector (e.g., cylodextrins, maltodextrins, macrocyclic antibiotics and fullerenes).

The high sensitivity, selectivity, enantioselectivity, accuracy and precision made the enantioselective, potentiometric membrane electrodes suitable to be used for the enantioanalysis of different pharmaceutical compounds such as S- and R-deprenyl and S-ibuprofen as raw materials and in their pharmaceutical formulations.

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Introduction

Chirality is a very important issue for different compounds of pharmaceutical and biological importance. In the modern pharmaceutical industry chirality has become a major concern. This is attributed largely to a heightened awarness that enantiomers of a racemic drug may have different pharmacological activities, as well as different pharmacokinetics and pharmacodynamic effects. The importance of chiral drugs, specifically single isomer drugs in the pharmaceutical market has grown at an exponential rate each year, due to recent developments in the area of chiral drug technologies. The gains in potency, efficacy and selectivity obtained by treatment with single isomer drugs are undeniable. Therefore, enantiopurity is now imperative in the production of most pharmaceutical products with a chiral moiety.

The challenges imposed on the pharmaceutical companies by the problems emanating from the side effects that could be caused by the presence of an undesirable component in racemic drugs, or problems arising from the different pharmacokinetics and pharmacodynamic effects that may be triggered by each of the enantiomer, has further stimulated the imperativeness to find an analytical method that can discriminate between the two enantiomers. These methods will be very helpful and should exhibit reliable analytical information, fast analysis, and could be employed for the enantiopurity tests of pharmaceutical compounds.

In molecular recognition of enantiomers, electrochemical sensors are a very good alternative for chromatographic and structural analysis techniques. Electrochemical sensors have superceded and out classed the other techniques with respect to many analytical aspects, including the high reliability that is given by high precision, high reproducibility, rapidity, and due to the fact that electrochemical sensors can be used directly for measurement of compounds in solution, without any prior separation of the substances that has to be analysed.

The primary aim of this dissertation is to construct reliable enantioselective, potentiometric membrane electrodes to be applied in enantiomeric analysis of pharmaceutical compounds: deprenyl and ibuprofen. Carbon paste are proposed as matrix for the sensors design, since it has proved to be the most reliable design, and due to the economic convenience of using carbon which can be easily obtained from the abundantly available and affordable graphite powder. For the selection of the best chiral selector, chiral recognition that is based on selective binding is considered.

Direct potentiometry was employed for the assay of enantiomers in pharmaceutical tablets and raw materials. The selection of the type of electrode and matrix of its membrane was done in accordance with the complexity of the structure of the enantiomer to be determined.

DECLARATIONS

I declare that the dissertation, which I hereby submit for the degree MSc at the University of Pretoria, is my own work and has not previously been submitted by me for a degree at this or any other tertiary institution.

Tumelo R. Mashile

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- Determination of S-(+)-ibuprofen using enantioselective, potentiometric membrane electrodes based on macrocyclic antibiotics
 - R.I. Stefan-van Staden, T.R. Mashile, B. Lal, K.C. Mathabate, J.F. van Staden **J.Pharm.Biomed.Anal.**, Submitted.
- 2. Enantioselective assay of S-(+)-ibuprofen using enantioselective, potentiometric membrane electrodes based on maltodextrins
 - R.I. Stefan-van Staden, T.R. Mashile

Sens.Actuators B, In Press (SNB 9191).

- 3. Determination of R-deprenyl using a maltodextrin based enantioselective, potentiometric membrane electrode
 - R.I. Stefan-van Staden, T.R. Mashile

Instrum.Sci.Technol., Submitted

4. Enantioselective, potentiometric membrane electrodes based on α -, β - and γ cyclodextrins as chiral selectors for the assay of S-deprenyl.

RI Stefan-van Staden, T.R. Mashile, J.F. van Staden, H.Y. Aboul-Enein **Anal.Chim.Acta**, Submitted.

- 5. Enantioselective, potentiometric membrane electrodes based on C_{60} fullerene derivatives for the assay of deprenyl
 - R.I. Stefan-van Staden, T.R. Mashile, B. Lal

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