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Supporting Information

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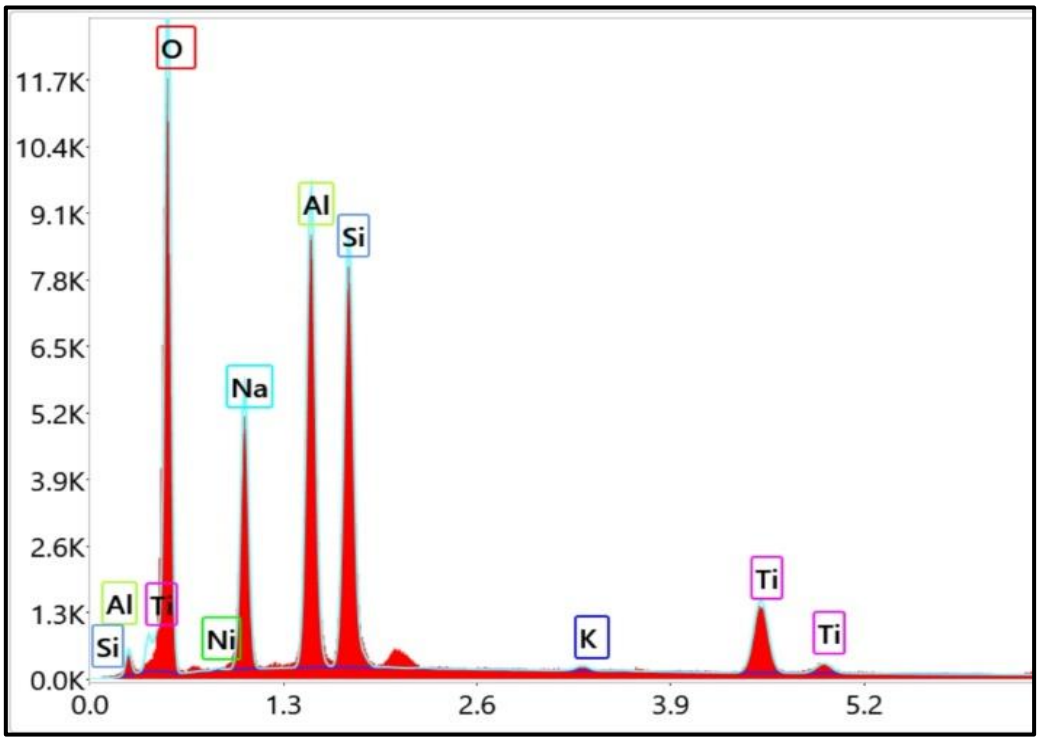

Fig.S1. EDX spectrum of SOD

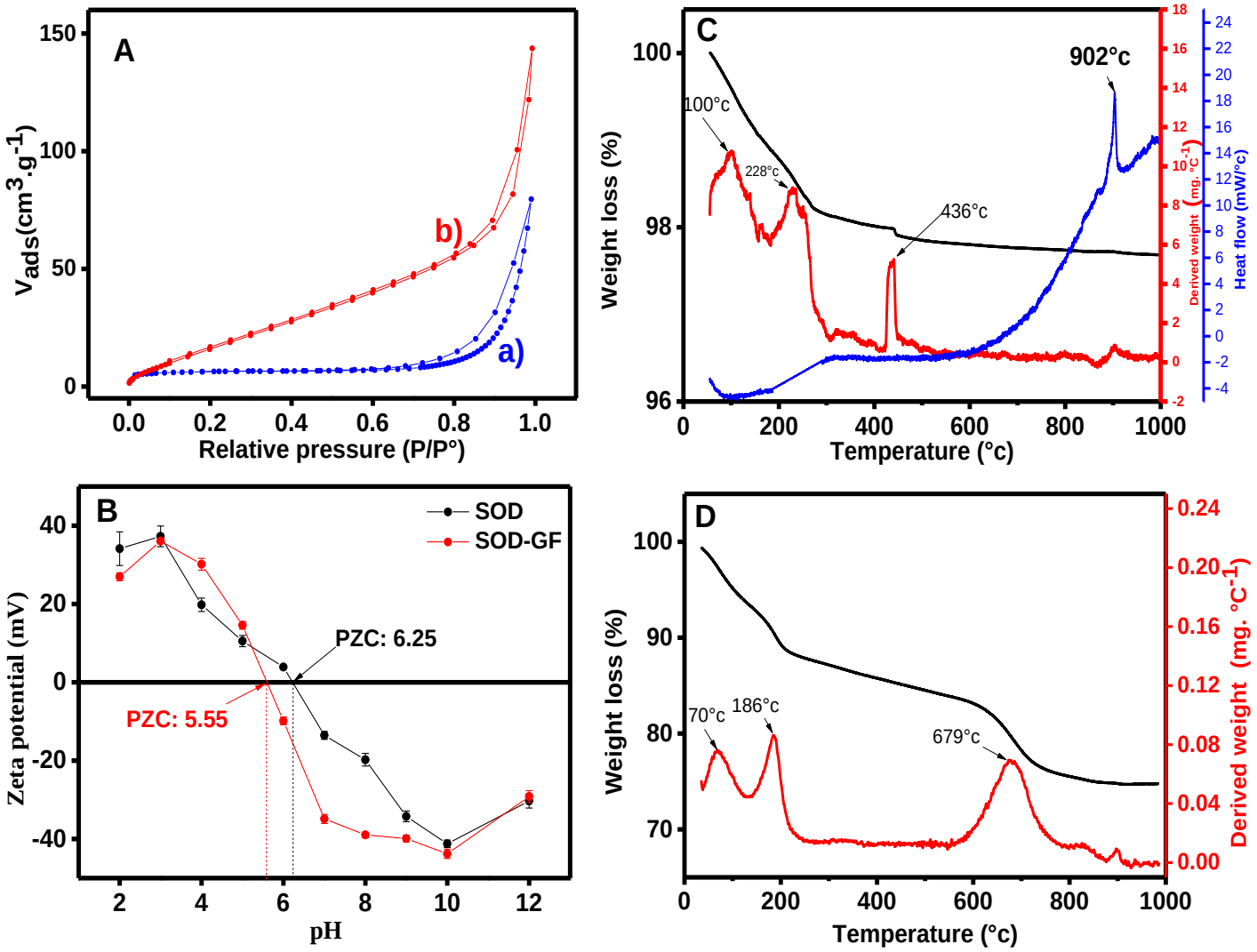

Fig. S2. **A)** Adsorption-desorption isotherms of **a)** SOD and **b)** SOD-GF; **B)** Zeta potential curves as function of pH; TGA curves of **C)** nanosodalite and **D)** composite

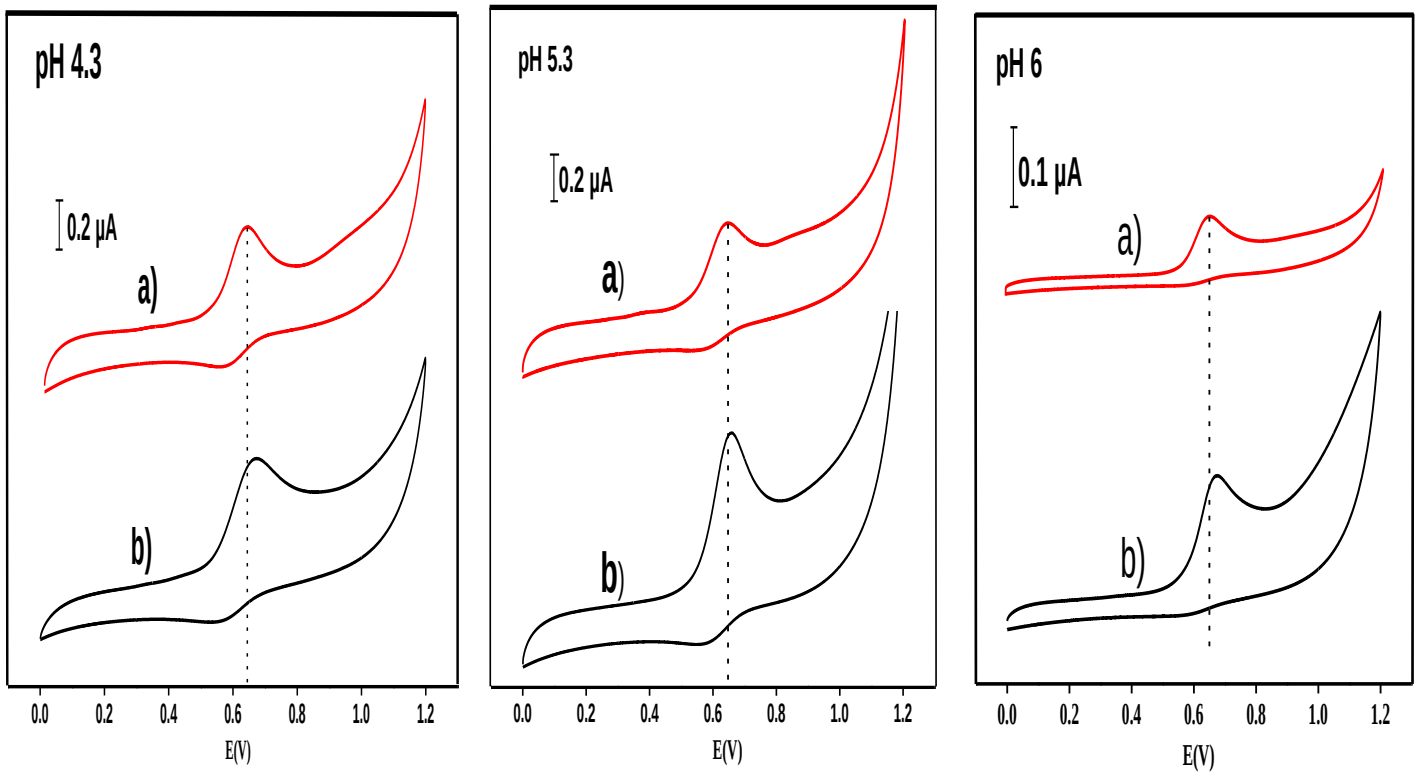

Fig S3. CVs recorded at 50 mV/s in RB at different pH containing 29.5 µM of CPZ at **a)** SOD-GF/GCE and **b)** SOD-GF/GCE

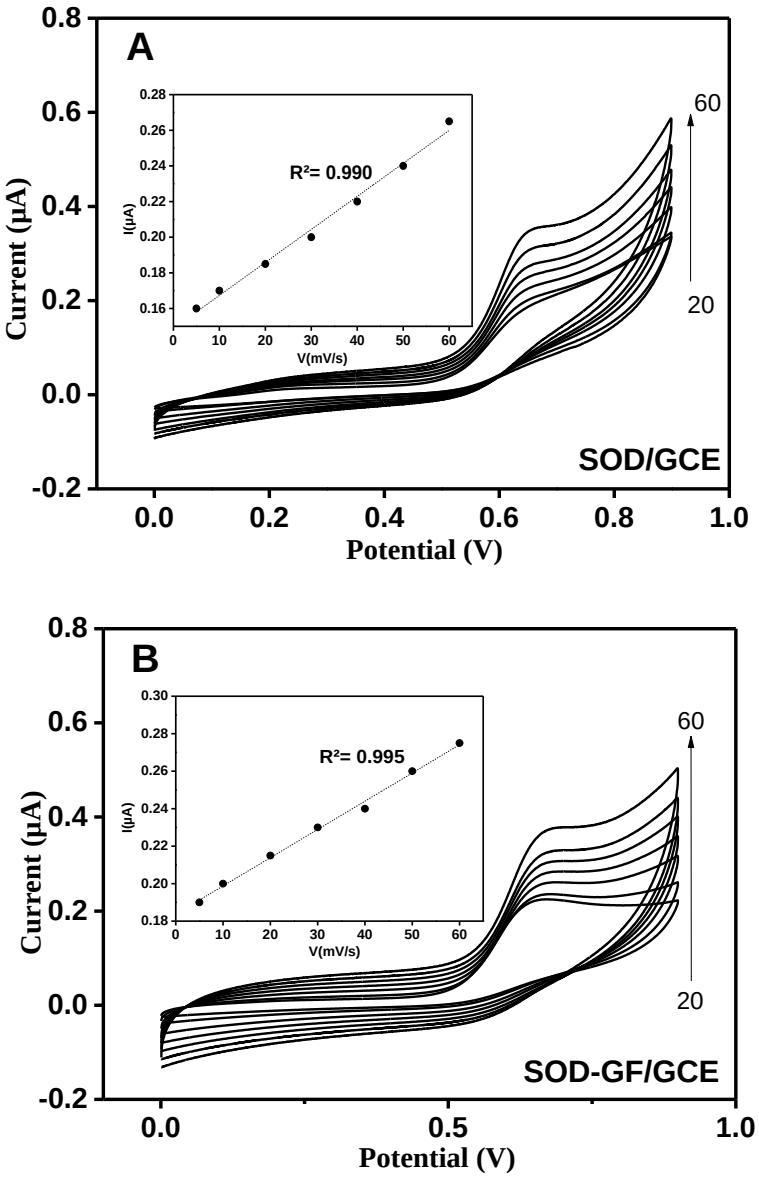

Fig. S4. Cyclic voltammograms recorded in RB (pH 10) at different scan rates containing 29.5 μM of CPZ for **A)** SOD/GCE and **B**) SOD-GF/GCE; inset show the plot of peak currents *vs* scan rate

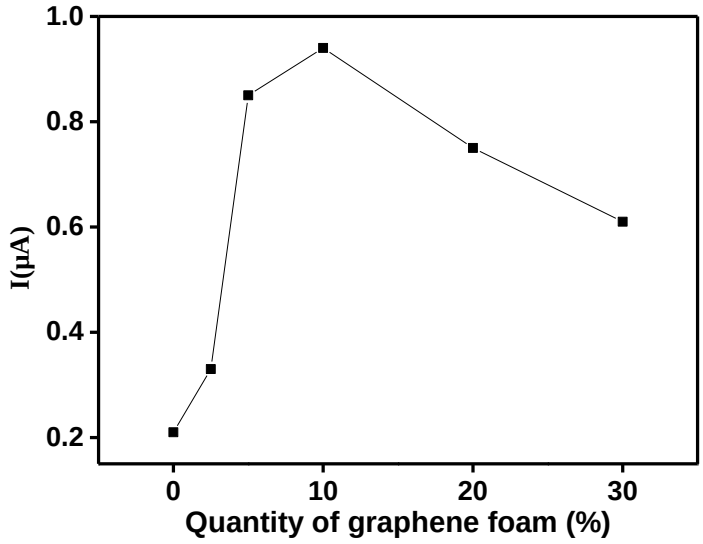

Fig. S5. Effect of various amounts of GF within SOD-GF composite film modified GCE on DPV currents peak recorded in RB pH 7.12 containing 5.2 μM

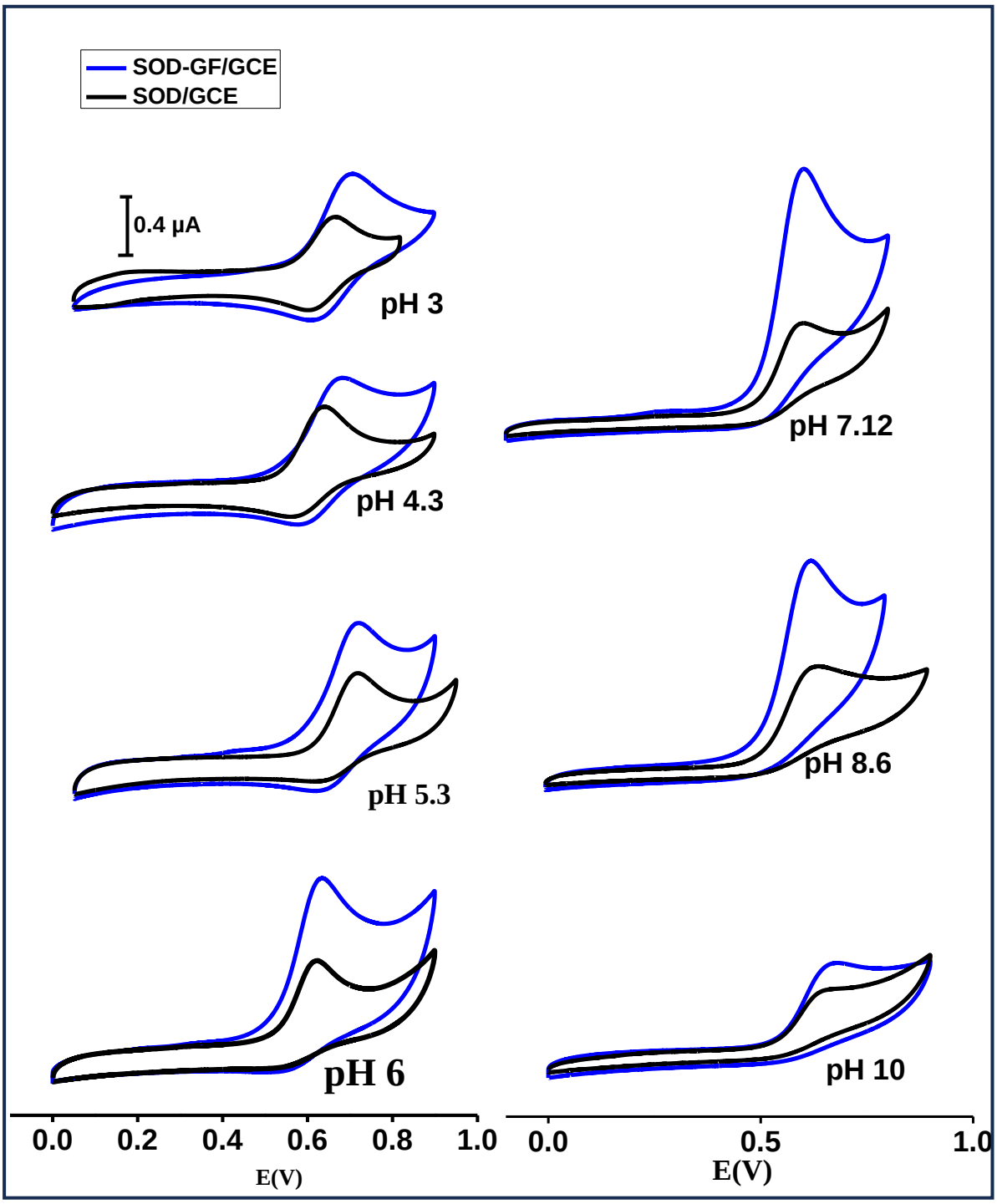

Fig S6. CVs recorded at 50 mV/s on SOD and SOD-GF in RB solution at different pH and containing 29.5 µM of CPZ

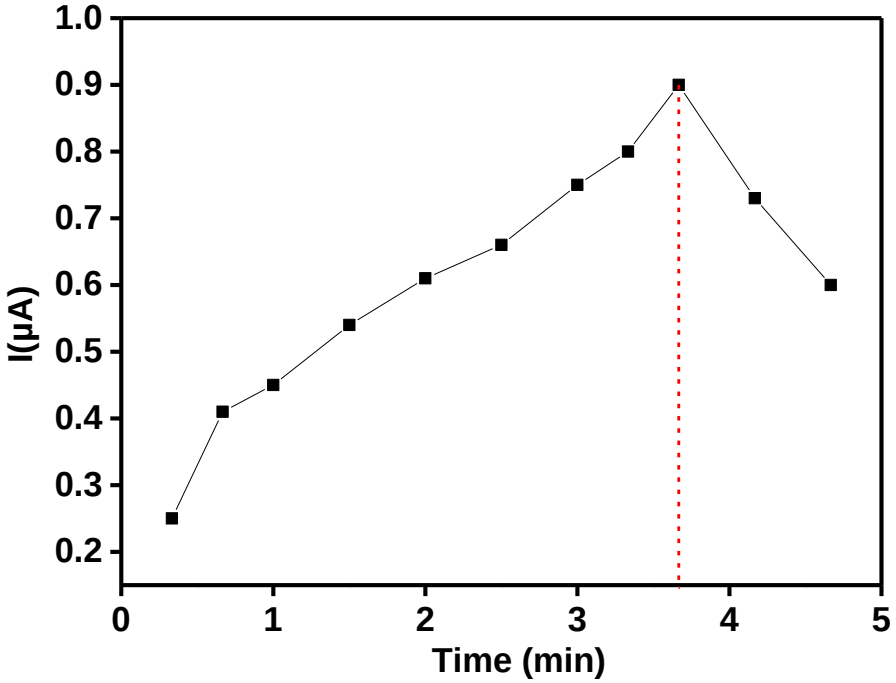

Fig. S7. Plot of DPVs response versus accumulation time after immersion of SOD-GF/GCE in RB (pH 7.12) containing 4.5 μM of CPZ.

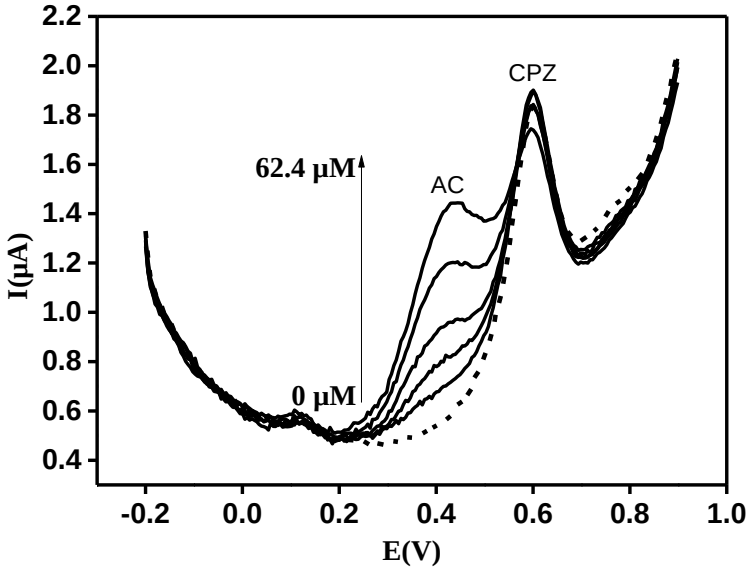

Fig. S8. DPVs recorded at SOD-GF/GCE in RB 7.12 containing fix concentration of CPZ in which gradual addition of AC from 0 – 62.4 µM was made

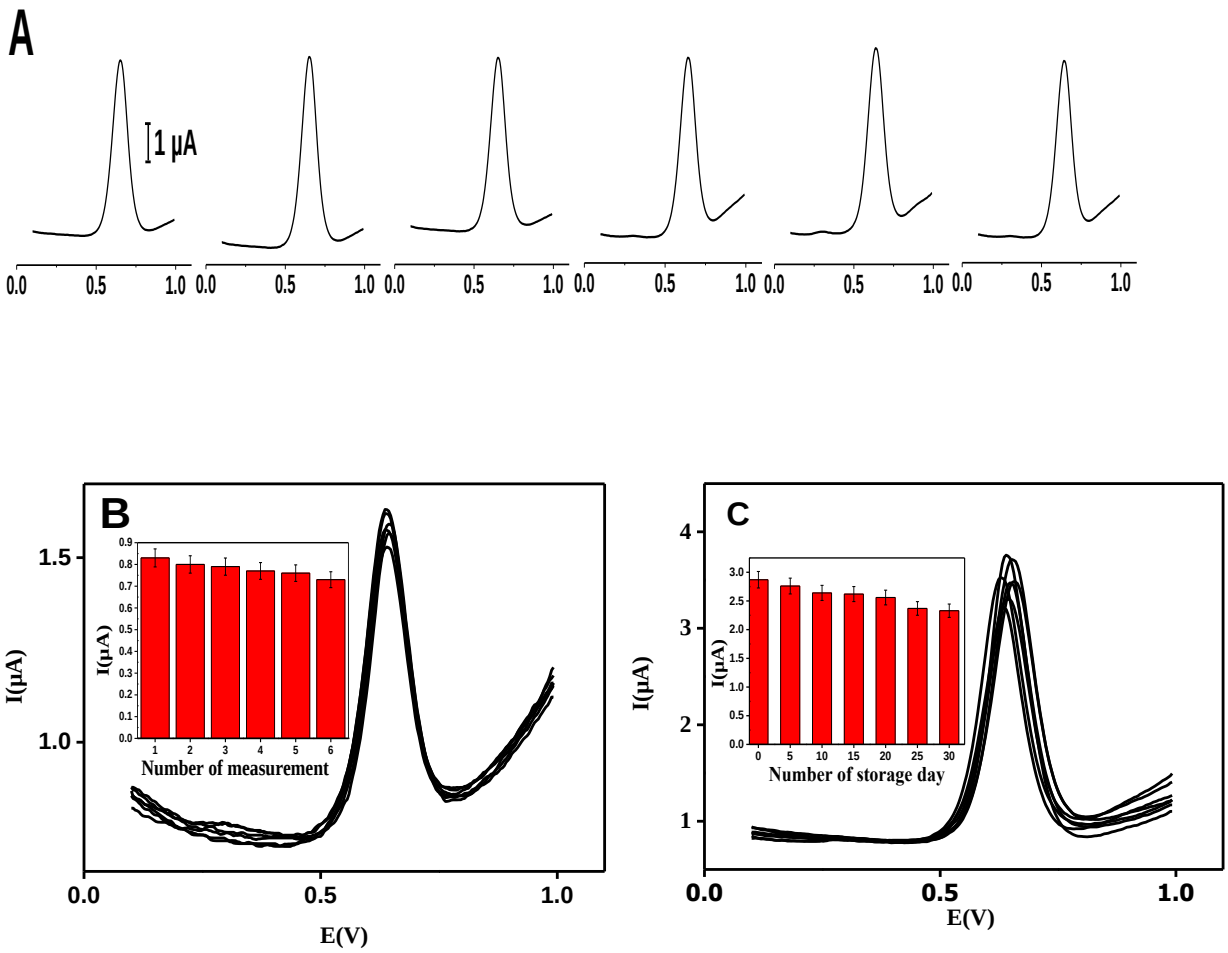

Fig. S9. A) DPVs recorded in RB pH 7.12 containing **20** µM of CPZ at SOD-GF/GCE under the same experimental conditions. Superposition DPV responses recorded in RB pH 7.12 containing **B)** 5 µM and **C)** 18 µM. Insets represents the corresponding DPV responses.