

Characterising pigments on 30 000-year-old portable art from Apollo 11 Cave, Karas Region, southern Namibia

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Abstract

As an unambiguous indication of complex cognitive capacity, representational art presents explicit evidence for modern and symbolic human behaviour. The only examples of African figurative art dating to the Late Pleistocene comprise seven stone plaques recovered from Apollo 11 Cave in the Huns Mountains, southern Namibia. The plaques derive from a single anthropogenic layer dated by radiocarbon (¹⁴C) accelerator mass spectrometry (AMS) and optically simulated luminescence (OSL) methods to c. 30 000 years ago. We present the results of digital (CIE) L*a*b* colourimetric and portable energy dispersive X-ray fluorescence (ED-XRF), Raman spectroscopic and Fourier transform infrared reflectance (FT-IR) analyses of the pigments present on the plaques. These results provide the earliest direct evidence, in Africa, for the preparation of pigment-based paint-like mixtures and their application to create prehistoric art. Our research shows that in the creation of the depictions on the plaques, the artists used black pigments derived from manganese and charcoal, red pigments likely derived from ocherous shale and white pigments possibly derived from ostrich eggshell. Additionally, these plaques provide unique evidence for the combined use of mineral- and carbon-based pigment 'crayons' during the African Middle Stone Age.

Keywords: Middle Stone Age; Apollo 11; Figurative art; Pigments; (CIE) L*a*b*; ED-XRF;

FT-IR; Raman

1. Introduction

The southern African Middle Stone Age (MSA) has long provided significant information concerning the cultural, behavioural and cognitive evolution of *Homo sapiens*. Of the

indicators of cognitive complexity that become prevalent during the MSA, the capacity for symbolic thought, and the use of symbolism to mediate social behaviour, provide definitive indications of behavioural modernity (Henshilwood and d'Errico, 2011). Given the evidence supporting the model that fully modern human behaviour originated in sub-Saharan Africa (Henshilwood *et al.*, 2009; Henshilwood *et al.*, 2011; Lombard, 2012 ; Rito *et al.*, 2013), the c. 30 ka age reported for the earliest southern African portable figurative art is not unusual. Early examples of scored ochre come from Pinnacle Point Cave 13B dated to 164 ka (Marean *et al.*, 2007), and Klasies River Cave 1 at 100 ka (d'Errico *et al.*, 2012a). At Blombos Cave, engraved designs on 17 pieces of ochre, some of which are arguably the earliest abstract 'art', have been recovered from levels dated at 100 ka to 72 ka (Henshilwood *et al.*, 2009). A single example of an ochre piece engraved with a cross hatched design was also found at Klein Kliphuis Shelter and is dated at 50 ka (Mackay and Welz, 2008). Geometrically engraved ostrich eggshell fragments have also been recovered from Diepkloof Rock Shelter (Texier *et al.*, 2013), Klipdrift Shelter (Henshilwood *et al.*, 2014) and Apollo 11 Cave (Vogelsang *et al.*, 2010) in levels dated to between 85 ka and 52 ka. The apparent final use of abstract geometric decoration occurs during the terminal phases of the MSA Howiesons Poort (HP), at c. 50 ka.

Following a hiatus of nearly 20 000 years in the occurrence of abstract decoration in southern Africa, the next known occurrence comes from Apollo 11 Cave in southern Namibia where seven portable stone plaques were recovered, four of which bear figurative depictions. These are the earliest known examples of figurative art in Africa, recently re-dated by radiocarbon (¹⁴C) accelerator mass spectrometry (AMS) and optically stimulated luminescence (OSL) methods to c. 30 ka (Jacobs *et al.*, 2008; Vogelsang *et al.*, 2010; Wendt, 1972; Wendt, 1974 ; Wendt, 1976). Apart from these examples, figurative portable art only reappears in southern Africa after the Last Glacial Maximum (LGM) to become a recurring feature in Later Stone Age (LSA) contexts (see Morris and Beaumont (1994); Pearce (2010); Thackeray *et al.* (1981)). Given the cultural and evolutionary significance of these artefacts, we present the results of digital (CIE) L*a*b* colourimetric and portable energy dispersive X-ray fluorescence (ED-XRF), Raman spectroscopic and Fourier transform infrared reflectance (FT-IR) analyses of the pigments present on the plaques.

2. Apollo 11 Cave

Apollo 11 is a small (~ 150 m²) cave situated in a limestone cliff face above the upper Nuob River in the Huns Mountains (Succulent Karoo biome), Karas Region, southern Namibia (Fig. 1). Wolfgang Erich Wendt (University of Cologne, Germany) commenced excavations at the site in 1969, dividing the stratigraphic sequence into 5 major units labelled 0 to IV, including several sub-units labelled A to H (Wendt, 1974). In 2007, Vogelsang *et al.* (2010) divided the sequence into 24 units labelled from A to Z. Trench A was excavated by Wendt in 1969 and includes eleven 1 m² squares labelled A2 to A12. Four of the plaques, here referred to as AP1, AP3, AP4 and AP5, were recovered from square A9 in August 1969. In 1972, two extensions in square A9 (A9x1 and A9x2) were excavated to clarify stratigraphic concerns and to establish the positions of the plaques recovered in 1969. Consequently, three further plaques (AP2, AP6 and AP7), one fitting to a fragment from the excavation in August 1969, were recovered in October 1972 (Wendt, 1974 ; Wendt, 1976) (Table 1).

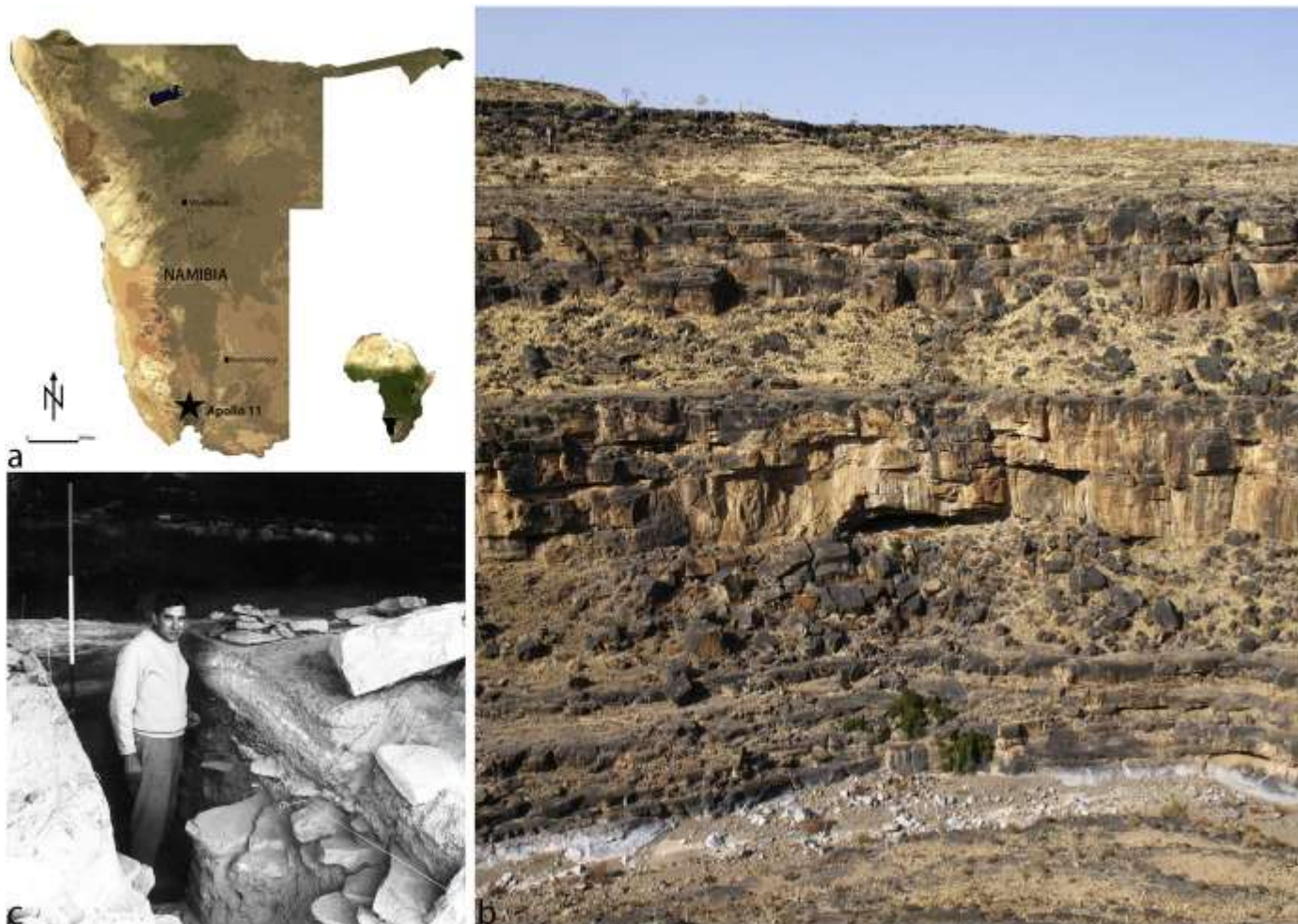


Fig. 1. Apollo 11 Cave is located in the Karas Region of southern Namibia (a) and situated against a limestone cliff on the eastern slope of the Nuob River (b). The site was excavated by W. Erich Wendt (University of Cologne, Germany) in 1969 and 1972 (c) (Photograph courtesy Antje Otto).

Table 1. Inventory of the plaques recovered from Apollo 11 in 1969 and 1972 (dimensions are indicated in mm for breadth, height and width, respectively).

Plaque number	Provenance	Recovered	Geology	Dimensions	Pigment
AP1 (NMN CN 2000-2500) <i>Hindquarters</i>	A9 + 10 cm	1969	Micaceous shale schist	64.07, 92.73, 9.11	Black red yellow white
AP2 (NMN CN 2000-2500) <i>Head and torso</i>	A9 Extension 2 + 5 cm	1972	Micaceous shale schist	48.96, 89.41, 9.12	Black white yellow
AP3 (NMN CN 2000-2500) <i>Zebra</i>	A9 + 2 cm	1969	Micaceous shale schist	80.77, 131.24, 9.75	Black white yellow
AP4 (NMN CN 2000-2500) <i>Rhinoceros</i>	A9 + 2 cm	1969	Micaceous shale schist	134.89, 107.93, 17.33	Black white orange
AP5 (NMN CN 2000-2500) <i>Unidentified</i>	A9 + 2 cm	1969	Micaceous shale schist	129.77, 96.11, 10.09	Red black grey white
AP6 (NMN CN 2000-2500) <i>Zebra</i>	A9 Extension 1– 5 cm	1972	Micaceous shale schist	132.05, 67.15, 7.92	Black white red
AP7 (NMN CN 2000-2500) <i>Unidentified</i>	A9 Extension 2 + 8 cm	1972	Micaceous shale schist	140.42, 115.49, 23.27	Black white red orange

The seven plaques derive from the uppermost horizon of Layer E, subsequently labelled Unit M by Vogelsang et al. (2010), at the interface between the final MSA and the earliest LSA levels (Wendt, 1974). Radiocarbon ages of 28 ka to 26 ka were initially reported for these levels (Wendt, 1974 ; Wendt, 1976), with four dates (PTA-1040, KN-I 813, KN-2056 and KN-2115) indicating a mean age of 28.5 ± 0.59 ka for the deposits surrounding the plaques. In 2007, a team from the University of Cologne collected eight samples for OSL dating (Jacobs et al., 2008) and subjected 44 samples from the original excavations to AMS dating (Vogelsang et al., 2010). It was determined that the lower LSA has a weighted mean age of 22.3 ± 0.4 ka and the uppermost MSA a mean age of 29.8 ± 1.1 ka. An AMS date of 29.0 ± 0.4 BP (KIA-35,917) and an OSL age of 29.4 ± 1.4 ka were obtained for the same deposits. Given these ages, and the fact that the uppermost MSA has a weighted mean age of 29.8 ± 1.1 years BP, we consider the plaques to be reliably dated to c. 30 ka.

Based largely on the perceived combination of animal and human physical characteristics on AP 1 and AP2 (Fig. 2), the Apollo 11 plaques have provided considerable inspiration for discussions concerning prehistoric symbolism and ideology. The depiction of anthropomorphic figures displaying animal physical characteristics (therianthropes) is an attribute widely associated with shamanistic cosmology (Lewis-Williams, 1981) and is pervasive in southern African San rock art. As the primary interpretation of therianthropes relates to shamans and their experiences of altered states of consciousness (Lewis-Williams, 2006), the image has been construed as reminiscent of the ability to both experience and communicate such altered states. The analyses of early LSA artefacts from Border Cave places the emergence of modern hunter-gatherer adaptation at 44 ka (d'Errico *et al.*, 2012b ; Villa *et al.*, 2012), suggesting that, in some regions at least, there might be a degree of technological and cultural continuity between the MSA and the LSA. It is, in light of the interpretation of the AP1 and AP2 imagery as shamanistic, therefore possible that a degree of ideological and cosmological continuity may exist between the MSA and the LSA.



Fig. 2. AP1 (on left) and AP2 (right) were recovered from Apollo 11 in 1969 and 1972, respectively. Colour images of AP3 to AP7 are provided in Online Resource 1. Photographic details of specific features are presented in Online Resource 2.

3. Materials and methods

The plaques are curated by the National Heritage Council of Namibia and the National Museum of Namibia. Analyses were carried out at the National Museum of Namibia in Windhoek. Prior to analyses, dust particles were dislodged with gentle air pressure. Visual observations were made with an Aven ZipScope digital USB microscope at 10 × to 25 × magnification. The plaques were photographed with a 16.2 MP Nikon D4 digital SLR camera fitted with a Nikon 60 mm AF-S Micro-Nikkor lens. Photo-mosaic images were imported into Nikon Camera Raw 8.6 and their exposure, white balance, sharpness and contrast settings adjusted (Rifkin et al., 2015). Image rendering therefore approximates the plaques to the extent that viable colourimetric (CIEL*a*b*) measurements can be made from the digital photographs.

In addition to obtaining measurements from the respective rock surface backgrounds of each plaque, we selected 37 pigment-rich points or ‘spots’ on the 7 plaques for analyses (Fig. 3). Our analytical protocol did not entail the sampling of pigments and we used only non-invasive portable ED-XRF, Raman spectroscopic and FT-IR analytical equipment. Since at least some pigmented areas could derive from post-depositional contact with artefacts and other materials, or as a result of organic taphonomic processes, we selected areas which appeared to a) host the most substantial concentrations of pigment and b) formed part of pigmented regions comprising discernible images or parts of images. Non-pigmented areas were also analysed to facilitate the comparison of pigments and the underlying rock surfaces. The differential results obtained for pigmented and non-pigmented surfaces are central to the identification of deliberately applied pigments.



Fig. 3. Areas analysed by digital (CIEL*a*b*) colourimetry and ED-XRF. Selections of the same areas were targeted for FT-IR and Raman analyses. Images shown were digitally enhanced with the DStretch® decorrelation stretch algorithm program (Harman, 2008 ; Rifkin et al., 2015).

3.1. Digital colourimetric analyses

Trichromatic colourimetric coordinates were determined within the Commission International de l'Eclairage (CIE) $L^*a^*b^*$ space (<http://www.cie.co.at/index.php/Publications/Standards>) in which L^* represents brightness (0 = black and 100 = white), a^* the red-green chromatic axis (+ a^* = red and - a^* = green) and b^* the yellow-blue axis (+ b^* = yellow and - b^* = blue) (Xu et al., 2012). We used the Adobe Photoshop CS6 'Lab slider option' to obtain $L^*a^*b^*$ values for each area analysed by ED-XRF. This method is gaining increasing applicability in scientific research (Frausto-Reyes *et al.*, 2009 ; León *et al.*, 2006). Colour measurements were obtained from 0.25 cm² areas which approximate the measurement apertures of most commercially available colourimeters. The CIE-2000 colour difference formula (ΔE^*_{2000}) (<http://www.brucelindbloom.com/index.html?ColorDifferenceCalcHelp.html>) was used to determine the colour differences between pigments and rock backgrounds.

3.2. ED-XRF

We used a Bruker Tracer III SD portable ED-XRF analyser to determine the elemental contents of the pigments on the plaques. The device was fitted with a Peltier cooled Silicon Drift Detector (SDD) with a resolution of 145 keV at 5.9 keV, resulting in a rate of 100 000 counts per second (cps). Two instrument settings were used. The instrument was configured to operate at 15 keV and 55 μ A without a filter. A vacuum system was included to remove air from the instrument and to facilitate the detection of low atomic number elements such as Al, Si and Ca. The device was also operated at 40 keV and 10 μ A. Two beam filters, comprising 1 mm Al and 1 mm Ti filters, were positioned in the filter slot between the sample and the X-Ray tube. These settings allowed the X-rays from 12 keV to 40 keV to excite elements with higher atomic numbers, specifically Fe and Mn. In both configurations, individual point analyses were performed for 120 s. Acquired data was processed with the Bruker PXRF and ARTAX software programs. We made control measurements on both experimentally pigmented and the actual archaeological rock support surfaces to establish the integrity of the equipment and to develop the optimal analytical procedure.







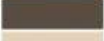
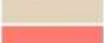


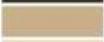










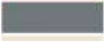








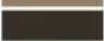













The XRF data were not obtained in controlled and reproducible conditions. The distance of the instrument from the samples, the surface roughness of the samples and ambient temperature and humidity could, among other factors, not be controlled. In this respect we cannot consider the data as accurate enough for true quantitative treatment. We therefore use the spectra and peak areas as qualitative and semi-quantitative information for a comparison between the rock support surfaces and pigmented surfaces.

3.3. FT-IR spectroscopy

Mid-infrared spectra were recorded using a portable Bruker Alpha-R spectrometer with a footprint of 22 \times 33 cm² and approximate weight of 7 kg. Infrared spectra could be collected without touching the sample by using a QuickSnap Reflection Module with a long working distance of about 2 cm. A video camera made it possible to observe the measured spot, which is circular and which has a diameter of roughly 6 mm. The spectral resolution of the

instrument is 4 cm^{-1} and 32 scans were averaged for each interferogram. Spectra were recorded in the range $700\text{--}4000 \text{ cm}^{-1}$ and are presented as reflectance spectra.

Table 2. Results for (CIE) $L^*a^*b^*$ colourimetric measurements of 0.25 cm^2 on the Apollo 11 plaques. Digitally recorded $L^*a^*b^*$ colour coordinate swatches and results for CIE-2000 colour difference (ΔE^*_{2000}) analyses are indicated. For all plaques, the respective rock supports were selected as the reference.

Plaque	Spot	Description	L^*	a^*	b^*	Colour	ΔE^*_{2000}
AP1	5	Rock support	72	9	28		–
	1	Black pigment	39	1	5		34.09
	2	Red pigment	68	40	35		17.69
	3	White pigment	95	1	18		17.34
	4	Black pigment	40	1	6		32.77
	6	Recent residue	47	8	20		22.49
AP2	6	Rock support	67	10	23		–
	1	Black pigment	34	4	9		34.13
	2	White pigment	85	3	14		14.94
	3	Red pigment	68	52	31		20.77
	4	Black pigment	23	2	4		43.61
AP3	5	Black pigment	20	1	3		45.74
	1	Rock support	73	6	22		–
	2	White pigment	88	2	15		11.54
	3	Black pigment	41	2	5		31.33
	4	White pigment	95	2	10		16.45
AP4	5	White pigment	91	2	16		13.11
	2	Rock support	63	6	16		–
	1	Black pigment	8	3	2		83.62
	3	Red pigment	49	24	44		18.83
	4	Black pigment	18	7	10		9.62
AP5	5	White pigment	93	1	11		85.31
	5	Rock support	61	8	20		–
	1	Red pigment	30	33	27		33.07
	2	Red pigment	44	44	39		24.46
	3	Grey pigment	50	2	3		21.81
	4	White pigment	91	2	8		23.37
	6	Black pigment	4	0	1		48.22
	7	Black pigment	20	2	2		38.48
	8	Grey pigment	52	1	4		16.23
	9	White pigment	96	1	6		26.80
AP6	10	Brown pigment	70	5	17		7.94
	5	Rock support	59	8	25		–
	1	Resin or mastic	67	5	14		9.02
	2	Resin or mastic	64	6	15		6.85
	3	Organic material	55	6	18		5.17
	4	Black pigment	21	2	6		35.43
AP7	6	White pigment	95	1	7		28.25
	4	Rock support	66	10	23		–
	1	Black pigment	25	1	2		41.83
	2	Orange residue	74	23	39		10.49
	3	Red pigment	50	41	21		23.8
	5	Red pigment	65	44	24		19.3
	6	Orange residue	65	19	39		7.35

3.4. Raman spectroscopy

Raman spectra were recorded with a portable BWTek iRaman Plus 785S instrument equipped with a triggered fibreoptic probe (BAC102). A 785 nm CleanLaze™ laser was used as excitation source, with the laser beam focused to an approximately 85 μm^2 spot. The Raman signal was detected with a 2048-pixel thermoelectrically cooled CCD camera (14 °C) with a spectral resolution of approximately 5 cm^{-1} . The instrument weighs \sim 3 kg and measures 17 \times 34 \times 23.4 cm. Several tests were conducted to establish the reliability of the equipment and to ascertain the optimal analytical procedure without touching the samples. We used non-archaeological red ochre and charcoal applied to a geologically similar type of stone for this purpose.

4. Results

4.1. Digital colourimetric analyses

Digital colour differentiation methods present an increasingly viable alternative to expensive and often unwieldy colour measurement devices (Frausto-Reyes *et al.*, 2009 ; León *et al.*, 2006). Colourimetric analyses indicate negligible to substantial differences (ΔE^*2000) between underlying rock support structures and pigments (Table 2). Minor differences in colour measurements do occur and are attributed to Adobe Photoshop rounding errors.

Colour measurement ($L^*a^*b^*$) values are most pronounced for the white pigments, followed by black and red pigments. The most significant differences in white pigments derive from AP4 (ΔE^*2000 85.31) and AP6 (28.25). Colour difference values (ΔE^*2000) of 83.62 were obtained for black pigment on AP4, 48.22 for black on AP5, and 41.83 for black pigments on AP7. Red is most distinguishable on AP5 (ΔE^*2000 33.07 and 24.06). The least distinct colour differences occur in the case of brown pigment on AP5 (ΔE^*2000 7.94) and orange residues on AP7 (7.35).

4.2. X-ray fluorescence (ED-XRF)

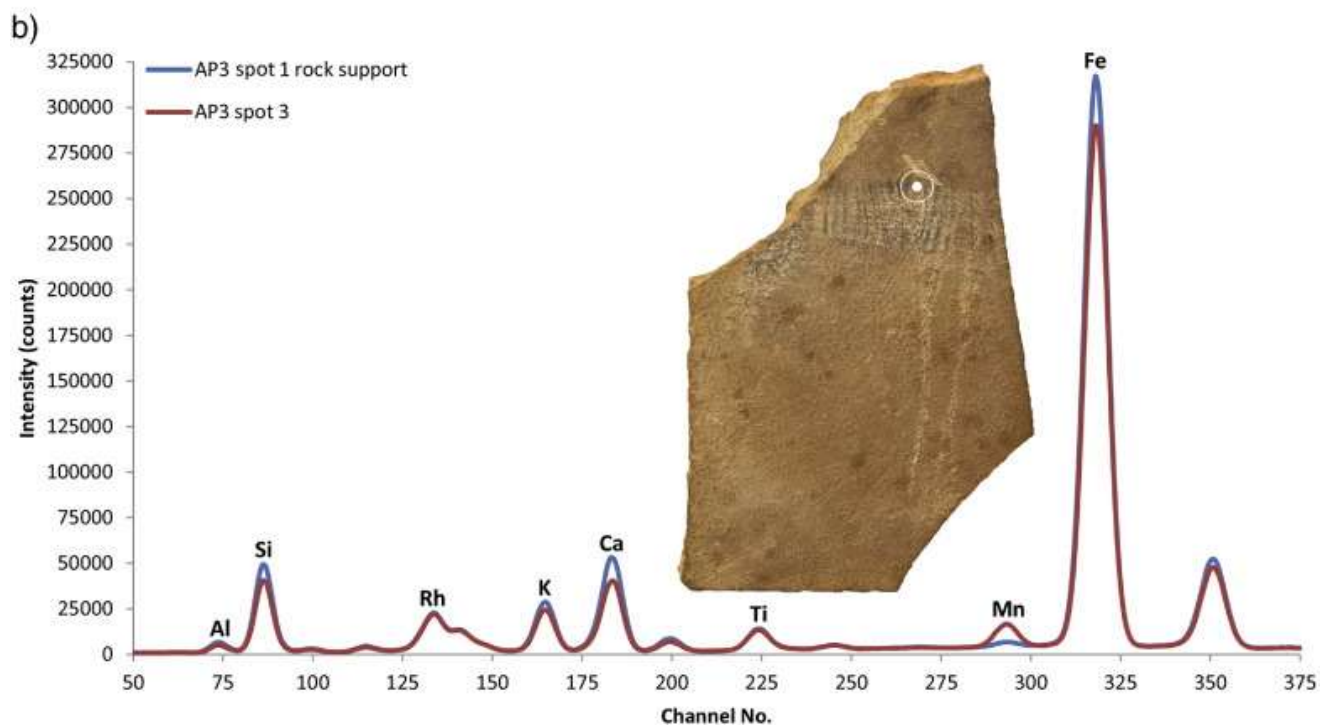
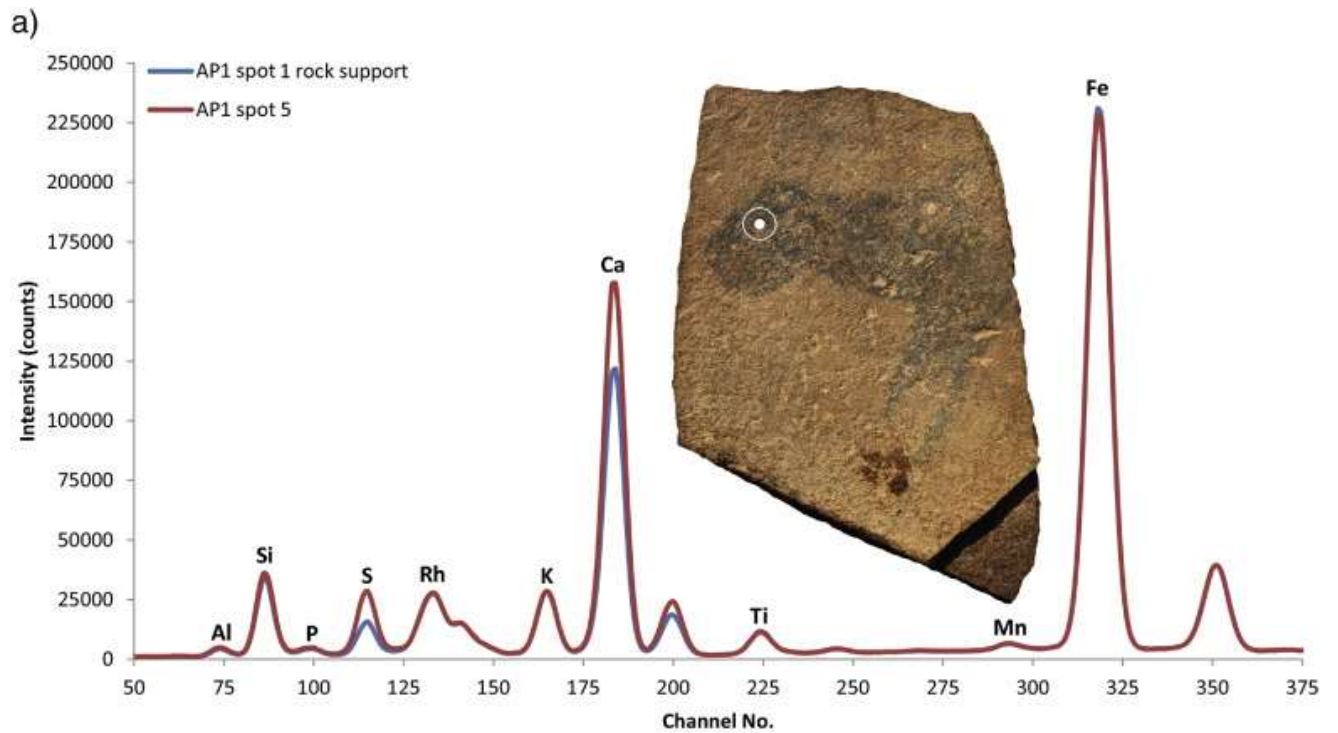
ED-XRF analyses indicate that the micaceous shale schist rock mediums comprise Al, Si, K, Ca and Fe as major elements and Ti and Mn as minor or trace elements (Table 3). Excitation at 15 keV revealed the presence of Al, Si, K, Ca, Ti, Mn and Fe and excitation at 40 keV the presence of heavier elements including Cu, Zn, Rb, Sr, Y and Zr. The incidence of these elements was not considered significant as no marked differences in signal intensities from rock supports and the pigmented areas were perceived.

Table 3. Normalised counts for ED-XRF analyses of AP1 to AP7. Measurements for the rock mediums are indicated in italics.

Plaque	Spot	Primary colour	<i>Elements detected at 15 keV</i>							<i>Elements detected at 40 keV</i>						
			Al	Si	K	Ca	Ti	Mn	Fe	Cl	Cu	Zn	Rb	Sr	Y	Zr
AP1	5	Support	<i>0.00890</i>	<i>0.06130</i>	<i>0.05540</i>	<i>0.31080</i>	<i>0.02470</i>	<i>0.01810</i>	<i>0.52080</i>	<i>0.00000</i>	<i>0.00646</i>	<i>0.00935</i>	<i>0.04249</i>	<i>0.06564</i>	<i>0.03867</i>	<i>0.09122</i>
	1	Black	0.00840	0.06120	0.05860	0.25750	0.02750	0.01880	0.56800	0.00567	0.00610	0.00913	0.04180	0.06733	0.03800	0.08989
	2	Red	0.00810	0.05970	0.05830	0.24260	0.02860	0.01930	0.58330	0.00549	0.00615	0.00914	0.04053	0.05672	0.03565	0.09479
	3	White	0.00770	0.05840	0.05540	0.23980	0.02840	0.01990	0.59040	0.00558	0.00633	0.00947	0.04135	0.06408	0.03713	0.08565
	4	Black	0.00760	0.05440	0.05390	0.26990	0.02630	0.02030	0.56770	0.00000	0.00632	0.00874	0.04222	0.06193	0.03846	0.08829
	6	Brown	0.00710	0.04070	0.05220	0.40640	0.02170	0.01880	0.45320	0.00000	0.00665	0.00943	0.04057	0.05983	0.03671	0.08808
AP2	6	Support	<i>0.01340</i>	<i>0.12420</i>	<i>0.06040</i>	<i>0.07440</i>	<i>0.03620</i>	<i>0.01960</i>	<i>0.67170</i>	<i>0.00486</i>	<i>0.00643</i>	<i>0.00901</i>	<i>0.04018</i>	<i>0.04741</i>	<i>0.03559</i>	<i>0.08696</i>
	1	Black	0.01250	0.11370	0.05990	0.06210	0.03600	0.01970	0.69610	0.00452	0.00664	0.00867	0.03910	0.04679	0.03640	0.08915
	2	White	0.01260	0.10060	0.06180	0.03980	0.03620	0.02040	0.72850	0.00863	0.00723	0.00894	0.04010	0.04845	0.03634	0.09292
	3	Red	0.01260	0.11410	0.06160	0.05910	0.03650	0.02020	0.69590	0.00506	0.00679	0.00883	0.04039	0.04522	0.03719	0.08766
	4	Black	0.01170	0.10270	0.06210	0.07530	0.03740	0.01990	0.69090	0.00484	0.00714	0.00919	0.04085	0.05013	0.03669	0.09049
AP3	5	Black	0.00980	0.08540	0.05880	0.11180	0.03350	0.02040	0.68030	0.00466	0.00667	0.00876	0.04018	0.05002	0.03689	0.08882
	1	Support	<i>0.01130</i>	<i>0.08090</i>	<i>0.05370</i>	<i>0.10340</i>	<i>0.03010</i>	<i>0.01820</i>	<i>0.70240</i>	<i>0.00484</i>	<i>0.00714</i>	<i>0.00919</i>	<i>0.04085</i>	<i>0.05013</i>	<i>0.03669</i>	<i>0.09049</i>
	2	White	0.01090	0.07990	0.05400	0.13460	0.03080	0.02090	0.66900	0.00427	0.00941	0.00746	0.03698	0.05212	0.03413	0.06543
	3	Black	0.00930	0.07280	0.05010	0.08660	0.03150	0.04270	0.70700	0.00408	0.00947	0.00752	0.03417	0.04612	0.03094	0.06102
	4	White	0.01110	0.09000	0.05460	0.11060	0.03310	0.01720	0.68330	0.00426	0.00947	0.00740	0.03530	0.05141	0.03269	0.06493
AP4	5	White	0.01080	0.08320	0.05360	0.10080	0.03280	0.01820	0.70060	0.00408	0.00944	0.00729	0.03481	0.04722	0.03133	0.06350
	2	Support	<i>0.00800</i>	<i>0.04830</i>	<i>0.04700</i>	<i>0.35350</i>	<i>0.02630</i>	<i>0.03550</i>	<i>0.48150</i>	<i>0.02061</i>	<i>0.00470</i>	<i>0.00761</i>	<i>0.04005</i>	<i>0.08472</i>	<i>0.04477</i>	<i>0.16744</i>
	1	Black	0.00790	0.04960	0.04580	0.33330	0.02780	0.03650	0.49900	0.01435	0.00445	0.00769	0.03844	0.08459	0.04112	0.15822
	3	Red	0.00790	0.05500	0.06070	0.33610	0.02800	0.03220	0.48010	0.00722	0.00498	0.00725	0.03716	0.09005	0.03877	0.16261
	4	Black	0.00820	0.05470	0.04700	0.31400	0.03000	0.03590	0.51020	0.01556	0.00492	0.00785	0.04082	0.09413	0.04364	0.16891
AP5	5	White	0.00880	0.04960	0.04660	0.34400	0.02660	0.03490	0.48960	0.04878	0.00555	0.00787	0.04638	0.10834	0.04976	0.19213
	5	Support	<i>0.00510</i>	<i>0.01940</i>	<i>0.05330</i>	<i>0.71250</i>	<i>0.00930</i>	<i>0.02970</i>	<i>0.17090</i>	<i>0.00970</i>	<i>0.00531</i>	<i>0.00754</i>	<i>0.03904</i>	<i>0.07627</i>	<i>0.03911</i>	<i>0.10914</i>
	1	Red	0.00980	0.06920	0.06360	0.15640	0.03310	0.09350	0.57430	0.00868	0.00497	0.00746	0.03535	0.07212	0.03502	0.09625
	2	Red	0.01180	0.08380	0.07310	0.13450	0.03360	0.10630	0.55690	0.00926	0.00503	0.00732	0.03730	0.07446	0.03574	0.10087
	3	Grey	0.01210	0.08220	0.06250	0.11930	0.04100	0.11410	0.56880	0.00990	0.00489	0.00783	0.03827	0.08652	0.03758	0.10537
	4	White	0.01600	0.09950	0.08640	0.12800	0.03810	0.05430	0.57770	0.01041	0.00563	0.00836	0.04229	0.07382	0.04016	0.10897
	6	Black	0.00850	0.06040	0.06150	0.31310	0.02560	0.05180	0.47910	0.01179	0.00523	0.00768	0.03958	0.08884	0.03750	0.10290
	7	Black	0.01300	0.10350	0.06700	0.08010	0.04240	0.07250	0.62150	0.01911	0.00507	0.00611	0.03675	0.12864	0.03728	0.10555
	8	Grey	0.01260	0.09240	0.06870	0.10530	0.03500	0.11080	0.57520	0.01175	0.00513	0.00766	0.03856	0.08754	0.03650	0.10220
	9	White	0.00810	0.04990	0.05900	0.35790	0.02900	0.09880	0.39730	0.00990	0.00486	0.00783	0.03826	0.07652	0.03758	0.10437
AP6	10	Brown	0.00910	0.06370	0.06700	0.15550	0.03490	0.14050	0.52930	0.01461	0.00508	0.00731	0.03949	0.12081	0.03880	0.11302
	5	Support	<i>0.01130</i>	<i>0.08470</i>	<i>0.05220</i>	<i>0.12040</i>	<i>0.03010</i>	<i>0.01800</i>	<i>0.68350</i>	<i>0.00428</i>	<i>0.00900</i>	<i>0.00731</i>	<i>0.03551</i>	<i>0.04831</i>	<i>0.03217</i>	<i>0.06503</i>
	1	Brown	0.00740	0.04440	0.05520	0.52960	0.01920	0.03170	0.31250	0.00629	0.00626	0.00805	0.04146	0.17437	0.04056	0.10987
	2	Brown	0.00770	0.04480	0.05760	0.52550	0.02010	0.02290	0.32150	0.00600	0.00667	0.00786	0.04015	0.17314	0.03921	0.09467
	3	Brown	0.00970	0.08350	0.04770	0.12670	0.02790	0.06430	0.64030	0.00415	0.00871	0.00747	0.03474	0.04791	0.03213	0.06314
	4	Black	0.01010	0.09240	0.04980	0.14200	0.03300	0.01950	0.65320	0.00436	0.00926	0.00807	0.03631	0.04868	0.03293	0.06740
AP7	6	White	0.00930	0.07930	0.04920	0.14560	0.03220	0.01970	0.66470	0.00432	0.00928	0.00774	0.03745	0.04839	0.03391	0.06583
	4	Support	<i>0.00880</i>	<i>0.06760</i>	<i>0.05670</i>	<i>0.17160</i>	<i>0.03520</i>	<i>0.03920</i>	<i>0.62080</i>	<i>0.00567</i>	<i>0.00439</i>	<i>0.00781</i>	<i>0.04185</i>	<i>0.06795</i>	<i>0.03672</i>	<i>0.12443</i>
	1	Black	0.00840	0.06800	0.05220	0.19510	0.03400	0.07260	0.56960	0.00514	0.00465	0.00801	0.04254	0.06056	0.03772	0.11718
	2	Orange	0.00950	0.07450	0.05620	0.22310	0.03210	0.06550	0.53910	0.00520	0.00450	0.00830	0.04110	0.07427	0.03637	0.12042
	3	Red	0.00830	0.06990	0.05390	0.21880	0.03090	0.04700	0.57120	0.00880	0.00419	0.00791	0.04455	0.05943	0.03718	0.11395
	5	Red	0.00760	0.06140	0.05190	0.20290	0.03270	0.07530	0.56810	0.00509	0.00395	0.00851	0.04154	0.05556	0.03852	0.18718
6	Orange	0.00780	0.06180	0.05170	0.19660	0.03400	0.05860	0.58950	0.00500	0.00420	0.00818	0.04100	0.07187	0.03457	0.11442	

We attribute the detection of significantly higher levels of Fe in relation to Si to the depth of analysis of the instrument as well the presence of a thin layer of air. For Si, the penetration depth is relatively shallow and results in the detection of this element within the first 27 μm of the surface of silicate-rich geological mediums. Conversely, Fe is detected at depths of up to 300 μm (Drake, 2015). Whereas Si has a fluorescence efficiency of 4% (which means that only 4 in 100 Si atoms fluoresce), Fe has a fluorescence efficiency of 35%. The presence of a thin layer of air between the instrument and the plaques might also have resulted in the

absorption of signals from elements such as Al, Si and Ca. Examples of ED-XRF spectra of the pigmented areas and underlying rock supports are indicated in Fig. 4. Several analysed points exhibit dissimilar spectral profiles when compared to the relevant rock supports, particularly in the case of Ca, Fe and Mn (Fig. 4b and c). Peaks arising from the rhodium target tube appear in all ED-XRF spectra and are indicated as such (Rh).



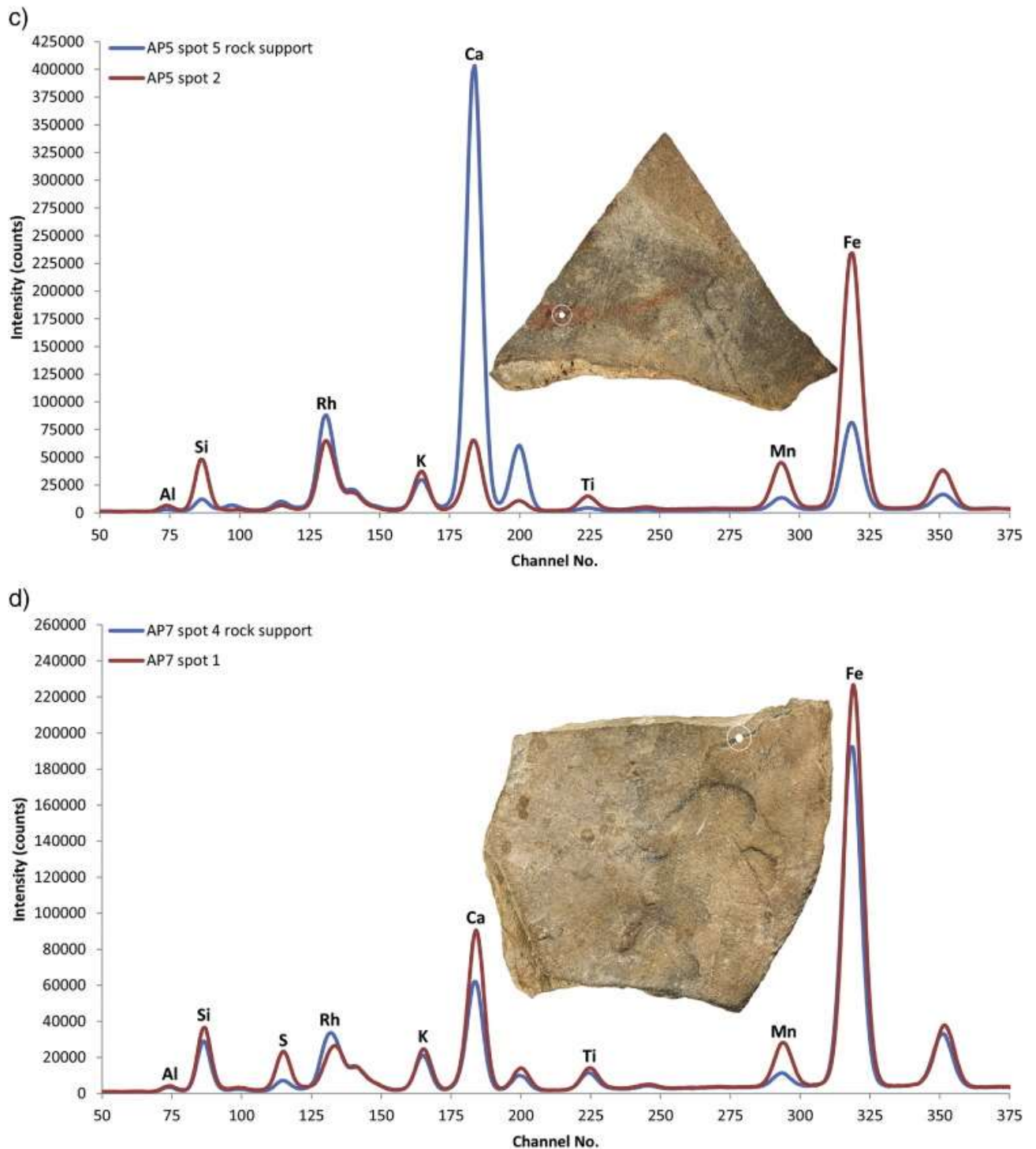


Fig. 4. ED-XRF spectra (15 keV) of traces of black pigment (spot 1) and the rock support surface (spot 5) on AP1 (a), of black pigment (spot 3) and the rock support (spot 1) on AP3 (b), of red pigment (spot 2) and the support (spot 5) on AP5 (c) and of the rock support surface (spot 4) and red pigment (spot 5) on AP7 (d).

To distinguish the elemental composition of pigments from that of the underlying rock surfaces, spectra of pigmented areas were compared with areas that were not covered with pigment. If the spectrum from a black pigmented area was similar to the ED-XRF spectrum, it was deemed probable that the black pigments used were carbon-based (López-Montalvo *et al.*, 2014 ; Roldán *et al.*, 2012). This is because organic black pigments generally comprise lighter elements that cannot be detected by ED-XRF. In instances where elements such as Mn and Fe or Ba, which generally form part of the oxides characteristic of inorganic

(geologically-derived) black pigments were absent, we conclude that carbon-based pigments were used. Fe is generally the primary component of red pigments, and in samples rich in clay minerals like ochre (red earth) or clayey rocks, Fe is usually mixed with Al, Si and sometimes also Ca. Ca is an intermittent component which is most often detected in white pigments. Calcite-based materials are the most commonly used white pigments and typically contain Ca but little to no P.

Most of the black pigments do not exhibit marked increases in Mn content, except for AP3 spot 3, in which case we interpret this to be indicative of the use of Mn-based pigments. Higher Ca contents are detected in AP1, AP5 and AP6 in resin- or mastic-like residues and a black and white pigmented areas, respectively. The presence of Ca in organic deposits could be due to the presence of calcium salts formed during post-depositional processes, or from a recent additive perhaps used during the curation of the plaques. A white deposit underlies the black pigment analysed on AP5. This white pigment, as well as white pigment from other areas on the same plaque, is probably composed of a calcium mineral. The high Ca content observed in the rock support for AP5 is unusual. Although it has been carefully selected to avoid pigment-rich deposits, the light colour of the area analysed suggests that it is in all probability covered by a white deposit. Its composition appears to be similar to that of the white pigment analysed on the support surface of AP5.

No significant differences in Fe content are apparent between either the red pigments or the underlying rock surfaces, nor is this the case for the other detected elements. This is most likely due to the high staining and surface covering capacity, but nevertheless superficial nature, of the red pigmented surfaces. In most instances, the red areas have faded substantially, making it difficult to determine whether these pigments were applied by painting or by drawing. Even the most prominent red layer, present on AP5, is in fact thin enough to expose an underlying black pigment layer. As a result of these ED-XRF related complications, we used FT-IR and Raman spectrometry as complementary analytical techniques for the identification of these pigments.

4.3. FT-IR spectroscopy

Since the high fluorescence backgrounds typically encountered during Raman analyses often limits the detection of unknown components (Prinsloo *et al.*, 2013) it was decided to use infrared spectroscopy as a complementary technique. Laboratory and synchrotron-based FT-IR spectroscopic instruments have been used successfully to analyse sampled rock art pigments (Hernanz *et al.*, 2006; Prinsloo *et al.*, 2008; Prinsloo *et al.*, 2013; Prinsloo *et al.*, 2014 ; Goodall *et al.*, 2009). In these studies, the pigment samples were either powdered and mixed with KBr and pressed into pellets for transmission measurements, or directly analysed using an ATR (attenuated total reflection) cell inside the instrument or attached to a microscope. In all of these instances, the data generated were transmission/absorbance spectra.

Although Raman and XRF portable instruments have been on the market for quite some time, and while numerous studies have reported their successful use in rock art studies (Bonneau *et al.*, 2012; Hernanz *et al.*, 2008; Hernanz *et al.*, 2012; Lufromento *et al.*, 2012 ; Prinsloo *et al.*, 2008), this is not the case for portable reflectance FTIR instruments.

This is because FTIR reflectance results in a combination of reflection and transmission/absorbance FTIR spectra, which is not always as easy to interpret. However, the increasing availability of portable spectrometers, and the possibility to use reflectance FTIR spectroscopy as non-invasive tool in heritage studies, have resulted in a number of in situ FT-IR reflectance studies on heritage building material, pigments in wall paintings and synthetic alteration of artefacts (Arrizabalaga *et al.*, 2015; Miliani *et al.*, 2012 ; Conti *et al.*, 2013). As far as we can ascertain, this is the first reported use of a portable reflectance FTIR instrument used for the analyses of rock art.

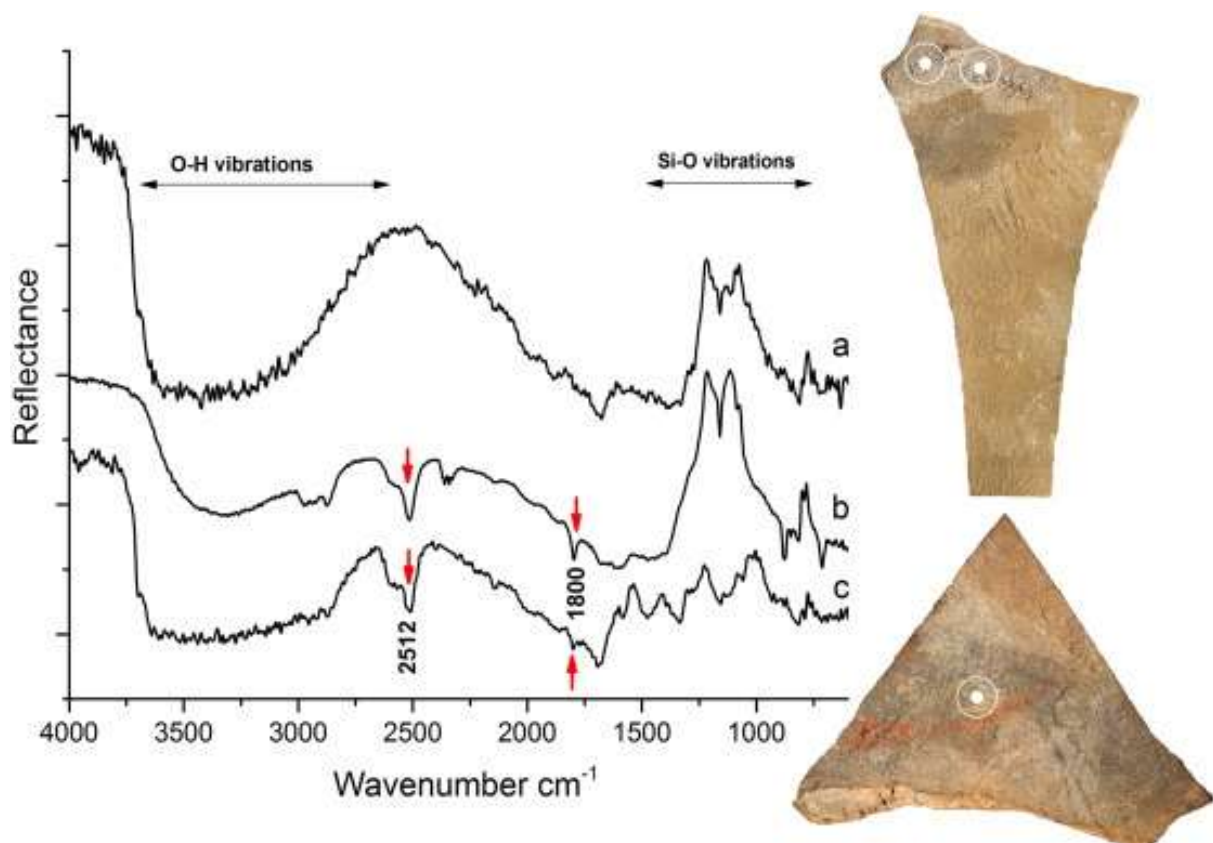


Fig. 5. A reflectance FT-IR spectrum representative of most spectra recorded on the Apollo 11 plaques (a) and a spectrum recorded for ostrich eggshell used as crayon to scribble on a quartzite surface (b). Comparable spectra were recorded for the residual crust on AP6 (spots 1 and 2 indicated at top right) and for the white pigment on AP5 (spot 9 at bottom right) (c).

In Fig. 5, the top spectrum (Fig. 5a) is representative of nearly all the spectra recorded on the samples, with the strong bands in the area of Si—O vibrations ($820\text{--}1290\text{ cm}^{-1}$) originating from the *reststrahlen* (residual rays) bands of the rock substrate consisting mostly of α -quartz. Although occurring at similar positions as true absorption/transmission bands, the origin of the *reststrahlen* bands are slightly different. Light is selectively reflected from the surface of a solid when the frequency of the light is nearly equal to the vibration frequency of the electrically charged atoms or ions, constituting the crystalline solid. The resulting reflection is known as *reststrahlen* (Prinsloo *et al.*, 2014). It should be noted that, when recording spectra in reflectance mode, absorptions appear as transmission spectra with the peak maxima in the opposite direction as the reflectance peaks. In some instances, we detected additional spectral bands (opposite direction of the reflectance bands), such as

in spectrum c (Fig. 5c) recorded for white pigment on AP6 (spots 1 and 2) and AP5 (spot 9). Prehistoric white pigments generally comprise white clay, hyena dung, raptor faeces or the combustion products of ostrich eggshell or bone (Prinsloo et al., 2013). Since calcite (CaCO_3) has been detected in South African rock art pigments (Tournié et al., 2010), and given that

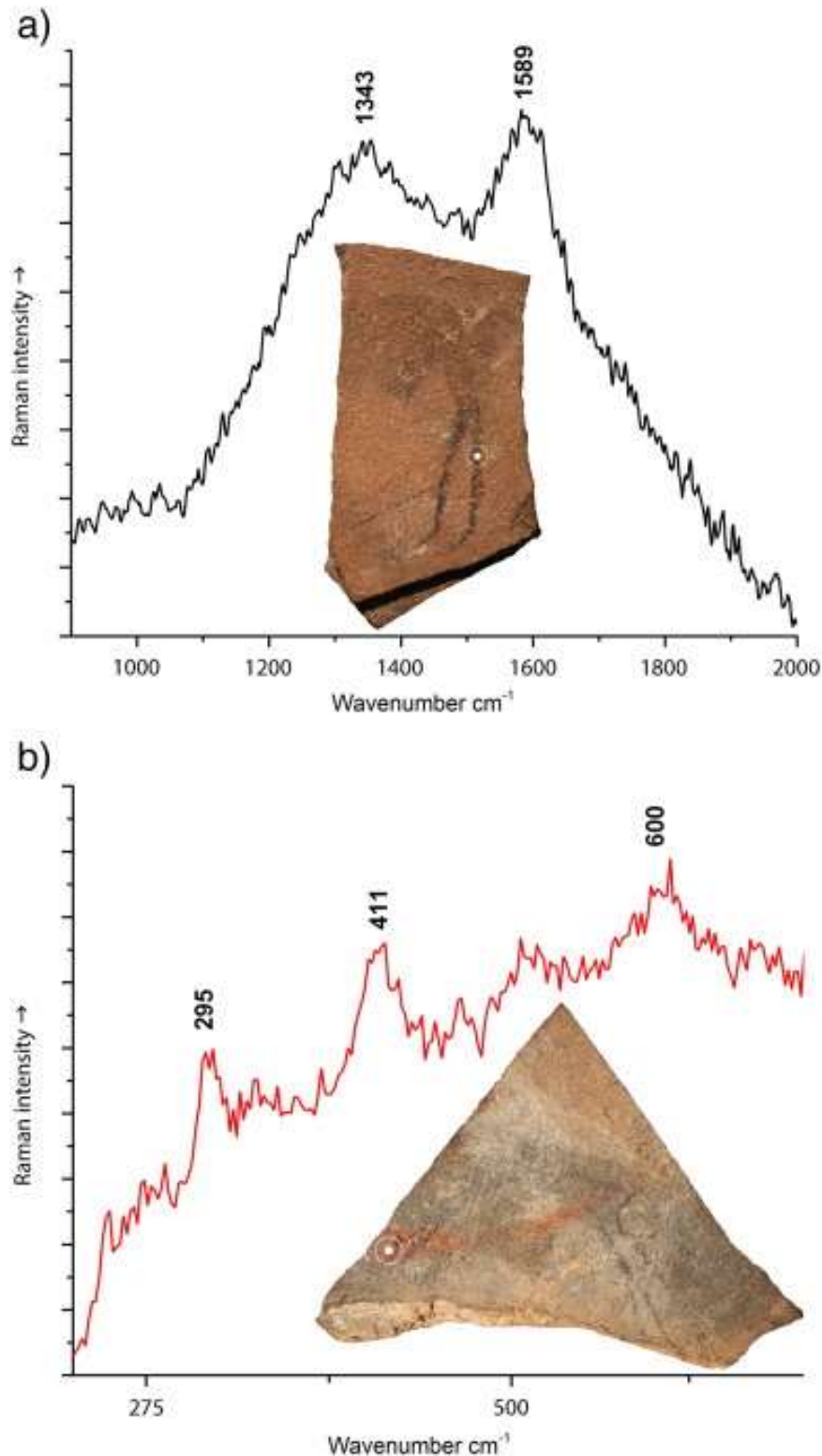


Fig. 6. Raman spectra provides a clear carbon signature for the black pigment on the 'leg' of AP2 (spot 4). Peaks are assigned as bands D and G typical of carbon structures (a). The presence of hematite on AP5 (spot 2) is indicated by the peaks at 295 cm^{-1} , 411 cm^{-1} and 611 cm^{-1} (b).

an abundance of ostrich eggshell fragments were excavated at the site, we produced a reference sample in reflectance mode of calcite by scratching ostrich eggshell (98% calcite) fragments onto a piece of quartzite. A spectrum of this sample (Fig. 5b) was recorded with a Hyperion microscope attached to a Vertex 70v (Bruker Optics) spectrometer and comprise strong reflectance bands of the quartz substrate and the same two bands at 1800 and 2512 cm^{-1} in spectrum c.

Calcite is identified in absorbance/transmission infrared spectra by a strong band at 1410 cm^{-1} and a characteristic sharp band at 872 cm^{-1} . In reflection spectra, this region is dominated by the *reststrahlen* band of the rock substrate and the calcite bands are distorted. The combination bands (sum or difference of fundamental bands) are enhanced in reflection spectra. It has been shown that the $\nu_1 + \nu_4$ and $\nu_1 + \nu_3$ bands can be used to identify carbonates and for calcite the bands occur at 1800 cm^{-1} and 2512 cm^{-1} , respectively (Arrizabalaga *et al.*, 2015 ; Miliani *et al.*, 2012). It is clear in Fig. 6 that both these diagnostic bands are present in the spectrum recorded on the rock art sample. Although calcite was positively identified, and in all probability does derive from using ostrich eggshell as a crayon, it is not possible to distinguish between ostrich eggshell and calcite derived from other sources. Ostrich eggshell generally does not produce a great pigment, but if heated until combustion occurs, the resultant product (CaO, generally known as lime and used as building material since pre-historic times) produces a superb pigment (Prinsloo *et al.*, 2013) which can be mixed with a binder to form a paint. Over time, CaO combines with atmospheric CO_2 to produce CaCO_3 which presents a similar spectrum as unprocessed ostrich eggshell.

As expected, we did not obtain spectra from the black, red and yellow pigments, primarily because black pigments tend to absorb infrared radiation and because the vibrational bands associated with red and yellow iron oxides fall outside the region measured with a portable FT-IR instrument.

4.4. Raman spectroscopy

Raman spectroscopy has been used effectively to analyse pigments both in the laboratory (Bonneau *et al.*, 2012; Hernanz *et al.*, 2008; Hernanz *et al.*, 2012; Lufromento *et al.*, 2012 ; Prinsloo *et al.*, 2008) and on-site (Lahlil *et al.*, 2012; Olivares *et al.*, 2013 ; Tournié *et al.*, 2010) and the use of portable instruments are well documented. In spite of using the red 785 nm laser line as excitation source, selected to be outside the wavenumber region of fluorescence originating from most organic materials (Prinsloo *et al.*, 2013), most spectra exhibited strong fluorescence. The spectra quality was also poor, which may be due to the fact that it was difficult to focus the spectra and to select optimally pigmented spots (the microscope attached to the instrument could not be used due to sample thickness). Another reason might be the accumulation of dust and other atmospheric pollution on samples in storage (Tournié *et al.*, 2010). It was nevertheless possible to identify some pigments, including the presence of carbon-derived black pigments, most likely charcoal, on several areas on AP2 (Fig. 6a) and also hematite (Fe-rich red pigment) on AP5 (Fig. 6b).

Table 4 provides a summary of the colourimetric ($L^*a^*b^*$), elemental (ED-XRF) and organic and inorganic (FT-IR and Raman) composition of the pigments present on the plaques.

Based on our observations under low magnification, we also provide information concerning the range of techniques most probably used to apply pigments to the plaques (see Online Resource 2).

Table 4. ED-XRF, FT-IR and Raman results and colourimetric coordinates (CIE-L*a*b*) for the pigments present on AP1 to AP7.

Plaque	Spot	Description	ED-XRF	FT-IR	Raman	ΔE^*_{2000}	Technique	
AP1	5	Rock support	Al Si K Ca Ti Fe Mn		TiO?			
	1	Black pigment	?		Carbon	34.09	Painted, drawn, rubbed and pecked	
	2	Red deposit/pigment	?			17.69		
	3	White deposit/pigment	?		CaCO ₃	17.34		
	4	Black pigment	?		Carbon	32.77		
	6	Brown residue/deposit	Ca			22.49		
AP2	6	Rock support	Al Si K Ca Ti Fe Mn					
	1	Black pigment	?		Carbon	34.13	Drawn and painted? Redrawn	
	2	White deposit/pigment	?			14.94		
	3	Red deposit/pigment	?			20.77		
	4	Black pigment	?		Carbon	43.61		
	5	Black pigment	Ca		Carbon	45.74		
AP3	1	Rock support	Al Si K Ca Ti Fe Mn		Carbon			
	2	White pigment (some black pigment)	Ca			11.54	Painted and drawn. Engraved?	
	3	Black pigment (some white pigment)	Mn			31.33		
	4	White pigment	?			16.45		
	5	White pigment	?			13.11		
	AP4	2	Rock support	Si Al K Ca Ti Fe Mn				
1		Black pigment	?		Carbon	83.62		Painted and drawn. Redrawn?
3		Red deposit/pigment	Si K			18.83		
4		Black pigment	Si Fe		Carbon	9.62		
5		White deposit/pigment	Cl			85.31		
AP5	5	Rock support/white deposit	Al Si K Ca Ti Fe Mn					
	1	Red and black pigments	Al Si K Ti Fe Mn		Hematite	33.07		
	2	Red pigment	Al Si K Ti Fe Mn		Hematite	24.46		
	3	Black and white pigments (grey)	Al Si K Ti Fe Mn			21.81	Painted, drawn, rubbed and redrawn	
	4	White pigment (some black pigment)	Al Si K Ti Fe Mn			23.37		
	6	Black pigment (some white pigment)	Al Si K Ca Ti Fe Mn			48.22		
	7	Black pigment	Al Si K Ti			38.48		

Plaque	Spot	Description	ED-XRF	FT-IR	Raman	ΔE^*_{2000}	Technique
			Fe Mn				
	8	Black and white pigments (grey)	Al Si K Ti Fe Mn			16.23	
	9	White pigment	Al Si K Ca Ti Fe Mn	CaCO ₃		26.80	
	10	Red and black pigments (brown)	Al Si K Ti Fe Mn			7.94	
AP6	5	Rock support	Al Si K Ca Ti Fe Mn				
	1	Brown residue/deposit	Ca K Mn Sr	CaCO ₃		9.02	
	2	Brown residue/deposit	Ca K Sr	CaCO ₃		6.85	Painted and drawn. Used as a palette?
	3	Brown residue/deposit	Mn		Carbon?	5.17	
	4	Black pigment	?			35.43	
	6	White pigment	?			28.25	
AP7	4	Rock support	Al Si K Ca Ti Fe Mn				
	1	Black pigment	Mn Ca		Carbon?	41.83	
	2	Orange residue (some black pigment)	Ca Mn Sr			10.49	Drawn and painted? Redrawn
	3	Red deposit/pigment	Ca			23.8	
	5	Red deposit/pigment	Fe Mn Ca			19.3	
	6	Brown residue/pigment (orange)	Mn Ca Sr			7.35	

5. Discussion

Our analyses indicate that a combination of mineral- and organically-derived pigments and techniques was used to create the figurative imagery on the Apollo 11 plaques (Table 4). The absence of P in the ED-XRF spectra excludes the use of black pigments obtained from bone and suggests the use of wood-derived charcoal black pigments. The black pigments on AP3, and to a lesser extent on AP7, exhibit elevated concentrations of Mn, suggesting that black or brown Mn-rich mineral pigment was used to create at least some parts of the images. The lack of notable changes in the elemental composition of black pigments and rock supports for AP1, AP2, AP4 and AP6 indicate that these pigments are in all probability carbon-based. Charcoal-derived pigments have been identified on southern African San rock art (Bonneau *et al.*, 2012 ; Prinsloo *et al.*, 2008) and also in French Palaeolithic caves (Chalmin *et al.*, 2003). Raman spectra provide a strong carbon signature for the pigment on AP2 (spot 4) (Fig. 6a). The tentative identification of carbon traces in the pigments of AP7 (spot 1) (1300 cm^{-1} to 1600 cm^{-1}) suggests that the artists may have used a combination of black pigments derived from both charcoal and manganese to create the imagery (Fig. 4b). CIE ($L^*a^*b^*$) colourimetric values do not facilitate the discrimination between carbon-based and Mn-based black pigments identified by ED-XRF and Raman. The presence of hematite derived from red mineral pigment on AP5 (spot 2) is not indicated particularly well by the ED-XRF results, but is confirmed by the Raman peaks recorded (Fig. 6b).

The differential elemental composition of pigments and supports for AP 6 (spots 1 and 2) confirms the presence of Ca. The probable use of ostrich eggshell as a white pigment is indicated by the spectral bands recorded on AP6 (spots 1 and 2) and AP5 (spot 9) (Fig. 5). Colour measurement values are also most pronounced for the white pigments, specifically those on AP5 (spot 9). White pigment on AP1 (spot 3) was identified to comprise CaCO_3 . AP3, AP4, AP6 and AP7 contain visible traces of a flaky, white translucent substance which may be the remains of a white pigment or a preparatory base. San artists possibly prepared rock surfaces for painting by washing with acidic liquids such as plant or fruit juice (Bonneau et al., 2012).

The images were created by way of a combination of different techniques (Table 4 and Online Resource 2). In Chauvet Cave (Ardèche, France), a combination of charcoal drawing, including modification by using fingers to blur or smudge lines, and fine engravings, was used to create figurative images at 30 ka (Fritz and Tosello, 2007). At 17 ka, the Great Bull in Lascaux Cave (Dordogne, France), was drawn with manganese oxides, charcoal and red ochre (Chalmin et al., 2007), and at Rouffignac Cave (13.5 ka), mammoths were engraved and drawn with manganese crayons (Lahlil et al., 2012). The use of both mineral- and carbon-based 'crayons' is therefore not unusual, but this study presents the first evidence for its combined use in the southern African MSA. Based on our visual macroscopic examination of the plaques, and on previous experimental work (Rifkin, 2012), the granular appearance of black pigments on AP2 and on AP5 suggests that these lines were in all probability applied in dry form, likely with a charcoal crayon. In contrast, some of the plaques bear residual traces of former liquid and possibly paint-like mixtures. The occurrence of a thick white residue rich in red and black particles on AP6 suggests that it may have been used as a palette on which to mix pigmentous ingredients. AP7 exhibits traces of black, white, orange and red pigments and several semi-circular remnants of what may represent dried-out traces of liquid paint drops (Rifkin et al., 2015). The homogenous distribution of black pigment on AP1 and white pigment on AP3 and AP5 suggests that these were applied to the plaques in liquid form. The white translucent substance on AP3, AP4 and AP6, thought to represent a white pigment or preparatory base, was also applied in liquid form. AP3 and AP6 furthermore provide evidence that black pigments were applied, by drawing, over white painted background surfaces. The temporal succession of pigments present on the plaques also point towards their social circulation and curation. Instances of superpositioning on AP3 and AP6, and on AP5 in particular, demonstrate that different types of pigments were repeatedly applied to this plaque, perhaps to restore older faded images and to create new depictions.

6. Conclusion

The greatest limitation of this study was the inability to subject the Apollo 11 plaques to laboratory-based analytical procedures. It was therefore essential to consider the application of suitable portable analytical instruments that could be used to analyse the artefacts on the premises of the National Museum of Namibia in Windhoek. Despite the fact that these portable instruments are not as precise as their laboratory-based versions, the ED-XRF, FT-IR and Raman equipment used in this study did facilitate the identification of at least four types of pigments (including red ochre-, black manganese- and charcoal- and white ostrich eggshell-derived pigments) used in the production of the imagery on the

plaques. Most studies aimed at ascertaining the composition of pigments used in parietal southern African San hunter–gatherer art confirm that red pigments most frequently derive from red ochre or hematite, that yellow pigments derive from yellow ochre or goethite, white generally comprise calcite, gypsum or, less frequently, white clay and that black pigments most often consist of amorphous carbon or manganese oxides (Rudner, 1983; Prinsloo *et al.*, 2008 ; Prinsloo *et al.*, 2013; Bonneau *et al.*, 2012).

Taking into account the increasing affordability of portable analytical instruments in relation to costly and stationary bench-top laboratory equipment, portable ED-XRF, Raman and FT-IR devices do provide viable alternatives to destructive sampling techniques and the problems associated with immobile analytical instruments. The results reported here therefore demonstrate the importance and advantages, and also the shortcomings, of using portable analytical techniques in studying the composition prehistoric pigments. It is known that MSA humans possessed the capacity to produce composite red ochre-based pigment mixtures at 100 ka (Henshilwood *et al.*, 2011) and also at 50 ka (Villa *et al.*, 2015), but the Apollo 11 art mobilier provides the first indication, in Africa, of the uses to which such paint-like compounds may have been put. In addition, and besides the incidence of red ochre particles in perforated shell beads from African and Near Eastern MSA and Levantine Mousterian sites dated from 92 ka to 60 ka (Bouzouggar *et al.*, 2007; d'Errico *et al.*, 2005 ; Vanhaeren *et al.*, 2006), this study provides direct evidence for the use of pigment-rich compounds to create figurative depictions.

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