

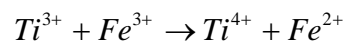
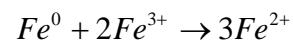
## 5 Appendices

### 5.1 Determination of tri-valent titanium ( $Ti^{3+}$ ) in titania slag

#### Calculations

The principle behind the determination of the trivalent titanium is the formation of reduced forms of titanium and iron when leached with ferric sulphate. The ferrous sulphate formed during the reduction of ferric ammonium sulphate is determined by titration with 0.1N potassium permanganate using 1.10-phenanthroline as an indicator. Iron will form iron (II) and therefore a correction must be made in the determination of  $Ti^{3+}$ .

#### Equations



$$\%Ti_2O_3 \text{ (expression in terms of } TiO_2) = \left[ 0.00799(A - B) \frac{100}{W} \right] - [1.43(C + 2D)]$$

Where  $A$  = Titrant value of potassium permanganate solution,  $cm^3$

$B$  = Blank titrant value of potassium permanganate solution,  $cm^3$

$C$  = Fe (tot) (from XRF)

$D$  = Metallic iron content (Determined as per Standard Task Procedure for **Determination of metallic iron ( $Fe^0$ ) in Titanium Slag**)

$W$  = Mass of the sample (mg)

1.43 = Conversion factor for Fe to  $Fe_2O_3$

79.9 = Molar mass of  $TiO_2$

Allowable variance in  $Ti^{3+}$ : 0.5%



Steps	Operating/Quality Criteria		✓x	Hazard and Risk Exposure	Control Measures	✓x
Check equipment/ tools/chemicals	Analytical balance (4 decimal place)	Titrastol ampoule (0.1N KMnO <sub>4</sub> )		Do not have the correct equipment, tools or chemicals	Obtain new or safe replacement before commencing with the task.	
	Water purifier	Stop watch				
	Magnetic stirrer bars	Stainless sink				
	Stirrer rod	Funnel, glass				
	Indicator bottle with a medicine dropper	Measuring cylinders, glass (25ml, 100ml, 500ml, 1000ml)				
	Volumetric flask, (100ml, 1000ml)	Measuring cylinder, plastic (10ml)				
	Hotplate with sand bath	Erlenmeyer flask				
	Magnetic stirrer/hotplate	1.1 Phenanthroline				
	Wash bottle	Sulphuric Acid (95-99%)				
	Beaker, teflon (250ml)	Boric Acid crystals (AR)				
	Beaker, glass (50ml, 250ml, 1000ml)	Calcium Gluconate Solution				
	Hydrofluoric Acid, HF (48%)	Ammonium Ferric Sulphate				
	Burette (50ml)	Ferrous sulphate				



Steps	Operating/Quality Criteria			✓x	Hazard and Risk Exposure	Control Measures	✓x
	CRM (SARM 57)						
1. Check Personal Protective Clothing (PPC)	Overalls/Lab coat		Safety shoes		Do not have the correct PPC	Obtain the appropriate PPC before commencing with the task	
2. Check Personnel Protective Equipment (PPE)	Nitrile gloves		Safety glasses		Do not have the correct PPE	Obtain the appropriate PPE before commencing with the ask	
	Latex gloves		Heat resistant gloves				
4. Ensure following documentation is available	Standard Task Procedure for: Calibration of, and weighing on an Analytical Balance (Powerdocs: IHM_CPC# 26953).				Documentation is not accessible from the computerized system.	Place a breakdown log with AST Help Desk, Dial 5000	
	Wet Chemistry Daily Log sheet (Powerdocs: IHM_CPC#47800).						
	Standard Task Procedure for: Cleaning up of chemical spillages (Powerdocs: IHM_CPC#27097).				Documentation is not accessible from the computerized system.	Place a breakdown log with AST Help Desk, Dial 5000	
	Standard Task Procedure for: Waste Management-Hydrogen Fluoride (HF) (Powerdocs: IHM_CPC#60720)						
	Standard Task Procedure for: Waste Management-Altered, Unaltered and Slag waste (Powerdocs: IHM_CPC#60726)						



Steps	Operating/Quality Criteria	✓x	Hazard and Risk Exposure	Control Measures	✓x
	Standard Task Procedure for Waste management – Glassware (Powerdocs: IHM_CPC#60724)				
5 Procedure	Inspect all glassware for chips or cracks before and after use.		Injury to hands	Use nitrile gloves	
			Incorrect disposal	Dispose of broken glassware into the “Broken glassware bin”.	
5.1 Preparation of reagents	Ensure that the balance has been calibrated as per Standard Task Procedure for <b>Calibration of, and weighing on an analytical Balance.</b>		Not calibrated	Calibrate and verify the balance before commencing with task.	
	Reagents are prepared using only “A” grade calibrated glassware.				
a) <b>Potassium permanganate solution</b> (KMnO <sub>4</sub> ) 0.1N	Prepare a titrasol ampoule of 0.1N KMnO <sub>4</sub> in a 1000ml volumetric flask.				
	Carefully break off the neck of the ampoule using the blade supplied., or twist off the neck (depending on which container is supplied).		Injury to hands.	Wear nitrile gloves.	
	Quantitatively transfer the contents of the ampoule into a clean dry 1000ml volumetric flask using a funnel.		Spillage	Discard and reprepare the reagent.	
	Rinse the neck and ampoule into the flask, at least 3 times with distilled water from a wash bottle.		Incorrect normality	Ensure that every drop is washed into the flask. Rinse down the funnel as well.	
	Make up to the mark with distilled water.		Over shot the mark	Discard and remake the solution.	



Steps	Operating/Quality Criteria	✓x	Hazard and Risk Exposure	Control Measures	✓x	
	Homogenise and transfer the solution into a dark coloured reagent bottle.					
	Seal, and label the reagent bottle with the name of reagent, date, time and name of analyst.		Incorrect labelling.	Ensure that the correct labelling is done.		
b) <b>Sulphuric Acid Solution 50 % (1:1) (Stock Solution)</b>	Quantitatively transfer 450ml distilled water into a clean dry 1000ml glass beaker using a 500ml measuring cylinder.		Water purifier not working	Replace DI cartridges.		
	Place the 1000ml beaker into the stainless sink.					
	Plug the sink using the stopper and fill with cold water to create a water bath.		Beaker begins to float and tips over.	Fill sufficient water in the bath to allow the beaker to be immersed without floating.		
	Quantitatively transfer 500ml of Sulphuric acid (95-99%) into a clean dry 500ml measuring cylinder			Inhalation due to fumes	Work in a fume hood.	
				Injury to hands	Wear nitrile gloves.	
				Spillage	Adhere to the Standard Task Procedure for <b>Cleaning up of chemical spillages.</b>	
	Slowly transfer small quantities at a time of the 500ml Sulphuric acid into the 1000ml beaker containing the 450ml of distilled water using a glass funnel.			Solution heats rapidly.	Add the acid slowly. Keep the volumetric flask cool by swirling it in the water bath or stir with a stirrer rod.	
			Let the H <sub>2</sub> SO <sub>4</sub> run onto the side of the beaker.			
			Always add acid to water.			



Steps	Operating/Quality Criteria	✓x	Hazard and Risk Exposure	Control Measures	✓x
			Injury to hands	Use nitrile gloves	
	Allow the solution to cool to room temperature in the water bath.				
	Once cooled, transfer to a clean dry 1000ml volumetric flask using a funnel and make up to the mark with distilled water.		Mark is over shot	Discard and reprepare.	
	This is now the 50% sulphuric acid stock solution.				
	Transfer to a reagent bottle, seal, and label the reagent bottle with the name of reagent, date, time and name of the laboratory technician.		Incorrect labelling.	Ensure that the correct labelling is done.	
	Transfer the stock solution to the designated auto dispenser when required.		Contamination of reagent	Ensure that the designated dispenser is used.	
				Rinse the dispenser with the new stock solution before filling up.	
c) <b>Boric Acid Solution</b> ( $H_3BO_3$ )	Add ~500ml of distilled water into a clean 2000ml volumetric flask.				
	Weigh 48g $\pm$ 0.01g of dry Boric acid ( $H_3BO_3$ ) crystals into a clean dry 50ml beaker.		Incorrect sample weight.	Discard and reweigh.	
	Place a clean dry glass funnel into the 2000ml volumetric flask containing the distilled water.				
	Quantitatively transfer the boric acid crystals into the volumetric flask by rinsing out the beaker with ~1000ml distilled water over the glass funnel.		Substances sticking to sides of the glassware	Ensure the sides of the glassware are washed down with water.	



Steps	Operating/Quality Criteria	✓x	Hazard and Risk Exposure	Control Measures	✓x
	Quantitatively transfer 220ml of Sulphuric acid (95-99%) into a clean dry 500ml measuring cylinder.		Inhalation due to fumes	Work in a fume hood.	
			Injury to hands	Wear nitrile gloves.	
			Spillage	Adhere to the Standard Task Procedure for <b>Cleaning up of chemical spillages.</b>	
	Slowly transfer small quantities at a time of the 220ml sulphuric acid into the 2000ml volumetric flask containing the boric acid solution.		Solution heats rapidly.	Add the acid slowly. Continuously swirl the flask under running cold water.	
			Injury to hands	Wear nitrile gloves.	
	Allow the solution to cool.				
	Once cooled, make up to the mark with distilled water.				
Transfer to a reagent bottle, seal, and label the reagent bottle with the name of reagent, date, time and name of the laboratory technician.		Incorrect labelling.	Ensure that the correct labelling is done.		
Transfer to the designated auto dispenser when required.		Contamination of reagent	Ensure that the designated dispenser is used.		
			Ensure that the designated dispenser is cleaned and rinsed with the new stock solution.		
d) <b>Ammonium Ferric Sulphate</b> (Stock Solution)	Weigh out accurately 25g ± 0.01g of Ammonium Ferric Sulphate into a clean dry 250ml beaker.		Incorrect sample weight.	Discard and reweigh.	



Steps	Operating/Quality Criteria	✓x	Hazard and Risk Exposure	Control Measures	✓x
	Using a measuring cylinder, add 150mls of distilled water into the beaker.		Water purifier not working	Replace DI cartridges.	
	Quantitatively transfer 75ml concentrated Sulphuric Acid (95-99%) into a 100ml measuring cylinder		Injury to hands	Wear nitrile gloves.	
			Inhalation due to fumes	Work in the fume hood.	
	Carefully add the 75ml Sulphuric Acid to the beaker containing the Ammonium Ferric Sulphate.		Solution gets hot	Allow the cool water to flow under the beaker.	
	Mix the solution well using a stirring rod.		Injury to hands	Wear nitrile gloves	
	Allow the solution to cool to room temperature.		Cooling takes too long	Force cool in a water bath.	
	Transfer the solution to the designated dark coloured reagent bottle. This is now the ammonium ferric sulphate stock solution.		Contamination of reagent	Ensure that the designated reagent bottle is used.	
				Ensure that the designated reagent bottle is cleaned and rinsed with the new stock solution.	
			Reagent is sensitive to light.	Store in a dark bottle, inside a cupboard.	
	Seal, and label the reagent bottle with the name of reagent, date, time and name of analyst.		Incorrect identification	Ensure the correct labelling is done.	
e) Phenanthroline indicator	Weigh out accurately, $0.7g \pm 0.01g$ Ferrous sulphate into a clean dry weighing boat.		Incorrect sample weight.	Discard and reweigh the sample.	
	Weigh $1.0g \pm 0.01g$ of 1.1 Phenanthroline into a separate clean dry weighing boat.		Incorrect sample weight.	Discard and reweigh the sample.	





Steps	Operating/Quality Criteria	✓x	Hazard and Risk Exposure	Control Measures	✓x
	Place a clean dry glass funnel into a 100ml volumetric.				
	Quantitatively transfer the chemicals into the volumetric flask by rinsing out the beaker with distilled water over the glass funnel.		Substances sticking to sides of the weighing boats.	Ensure the sides of the weighing boats are washed down with water.	
	Make up to the mark with distilled water.				
	Homogenize the solution.		Spillages.	Hold the flask firmly. Ensure the cap is on tightly.	
	Transfer to a reagent bottle, seal, and label the reagent bottle with the name of reagent, date, time and name of the laboratory technician.		Incorrect labelling.	Ensure that the correct labelling is done.	
5.2 Quality Control	A Quality control sample is prepared and analysed as per section 5.3, in duplicate with every batch of unknown samples.		Quality sample is out of limits	Do not report the production results Repeat the analysis.	
	Perform the analysis on LIMS by selecting the appropriate sample ID from LIMS <b>SPC Ti<sub>2</sub>O<sub>3</sub> &amp; Fe (Metallic)</b> and the method <b>Trivalent (Ti<sup>3+</sup>) Sarm57</b>		LIMS not available	Record the sample ID on the Wet chemistry Daily Log sheet.	
	SARM 57 is analysed in duplicate as per section 5.3. The average is reported.				
5.3 Preparation of the blank and sample	Unknown samples are prepared and analysed in duplicate and the average is reported.				



Steps	Operating/Quality Criteria	✓x	Hazard and Risk Exposure	Control Measures	✓x
	A blank should be prepared when new reagents are prepared and analysed with each batch of production samples.				
	Accurately weigh out $0.5000g \pm 0.0010g$ of milled sample into a weighing boat.		Incorrect sample weight.	Discard and reweigh. For the preparations of a blank, no sample is added.	
	Quantitatively transfer the weighed sample into a teflon beaker.				
	Add 15ml of the ammonium ferric sulphate stock solution into a teflon beaker, using a 25ml measuring cylinder.				
	Swirl to cover the entire sample.		Injury to hands.	Wear nitrile gloves.	
			Injury to eyes.	Wear safety glasses.	
	Transfer 20ml of 50% sulphuric acid into the teflon beaker, using the dispenser.		Incorrect volume dispensed	Ensure that the desired volume setting on the dispenser is selected before dispensing the solution.	
	Place the teflon beaker on the sand bath of the hot plate.				
	Bring to boil. Allow boiling for 1 minute while swirling by timing with a stopwatch.		1 minute exceeded	Repeat the preparation from step 5.3	
	Quantitatively transfer 10ml HF into 10ml plastic measuring cylinder, or using the dispenser, dispense 10 ml HF directly in to Teflon beaker.		Injury to hands	Use latex/nitrile gloves.	
			Inhalation due to fumes	Work in the fume hood.	



Steps	Operating/Quality Criteria	✓x	Hazard and Risk Exposure	Control Measures	✓x
			Damage to glassware	Use a plastic measuring because HF attacks the silica in glass.	
			Skin Contact	Avoid contact with skin. Wash with copious amount of water and then apply calcium gluconate solution to the area. Calcium gluconate solution is kept in the fridge in the Wet Chemistry Laboratory.	
				Seek medical help.	
	Add the 10ml HF into the teflon beaker.				
	Place the teflon beaker on the hotplate and boil at a lower heat setting for 10 minutes while swirling.				
	After boiling for 10 minutes remove the teflon beaker from hot plate.		Injury to hands.	Use heat resistant/nitrile gloves.	
	Immediately add 100ml boric acid stock solution to the teflon beaker using the dispenser.		Inhalation due to fumes.	Work in the fumehood.	
	Cool to room temperature.				
	Add 10 drops of the Phenanthroline indicator into the teflon beaker, using the medicine dropper.				
	Place a magnetic stirrer bar into the teflon beaker and stir well on the magnetic stirrer.				



Steps	Operating/Quality Criteria	✓x	Hazard and Risk Exposure	Control Measures	✓x
5.4 Titration of the sample	Rinse a clean dry 50ml burette out with 0.1N KMnO <sub>4</sub> solution.		Contamination	Rinse the burette with the solution to ensure it is free from contaminants.	
	Fill the 50ml burette to the mark with 0.1N KMnO <sub>4</sub> .				
	Add drop wise KMnO <sub>4</sub> until the endpoint is reached.		Endpoint is exceeded	Repeat the analysis from step 5.3	
	Enter the titration value (volume of the 0.1N KMnO <sub>4</sub> used), on LIMS under the appropriate sample type and method		Incorrect LIMS method selected.	Select the appropriate sample ID and the method <b>Trivalent (Ti<sup>3+</sup>) Sarm57</b> for the control sample and <b>Trivalent titanium (Ti<sup>3+</sup>)</b> for production samples.	
			LIMS is not accessible	Record the titrant value (volume of the 0.1N KMnO <sub>4</sub> used) on the Wet Chemistry Daily Log sheet	
5 Reporting of results	All calculations are performed by LIMS				
	Quality control samples are analysed in duplicate and the average is reported.				
	Unknown samples are analysed in duplicate and the average is reported.				
7 Waste Management	Discard liquid mercury chloride waste and contaminated waste paper in the designated mercury chloride waste bins in the Wet Chemistry Laboratory.		Incorrect waste disposal.	Adhere to Standard Task Procedure for <b>Waste Management: Mercury Chloride.</b>	



Steps	Operating/Quality Criteria	✓x	Hazard and Risk Exposure	Control Measures	✓x
	Discard altered slag waste in the appropriate bins in the mill and pressroom.		Incorrect waste disposal	Adhere to Standard Task Procedure for <b>Waste Management: Altered, Unaltered and Slag Waste</b>	
	Discard chipped or broken glassware into the appropriate bin in the Wet Chemistry Laboratory.		Incorrect waste disposal	Adhere to Standard Task Procedure for <b>Waste Management: Glassware</b>	
8 Housekeeping	Wash all glassware and store away.		Damage to glassware.	Ensure that all damage to equipment/glassware is reported before the end of the shift.	
	Clean the workbench on completion of analysis.				
	Return all reagents to their respective storage cupboards				
	Clean up all spillages		Spillage	Adhere to the standard task procedure for <b>Cleaning up of chemical spillages.</b>	



## 5.2 List of parameters evaluated in the search for parameters correlating with the -106 $\mu\text{m}$ fraction.

initial block mass

block yield

yield mass

%FeO

%TiO<sub>2</sub>

%Al<sub>2</sub>O<sub>3</sub>

%CaO

%Cr<sub>2</sub>O<sub>3</sub>

%MgO

%MnO

%SiO<sub>2</sub>

%V<sub>2</sub>O<sub>5</sub>

primary cooling time (hrs)

Tapping rate (kg/min)

Tapping rate<sup>-0.5957</sup> (kg/min)

Equivalent %FeO

(Equivalent %FeO)<sup>2</sup>

%Ti<sub>2</sub>O<sub>3</sub>

%TiO<sub>2 true</sub>

(%TiO<sub>2 true</sub>)<sup>2</sup>

Equivalent %Ti<sub>2</sub>O<sub>3</sub>

(Equivalent %Ti<sub>2</sub>O<sub>3</sub>)<sup>2</sup>

SiO<sub>2</sub>+Al<sub>2</sub>O<sub>3</sub>(glass)+CaO

Specific surface area - initial

Ln(spec surface area - initial)

Specific surface area - after

Ln(spec surface area - after)

(Specific surface area - after)<sup>3.6321</sup>

T<sub>liquidus</sub> (calculated)

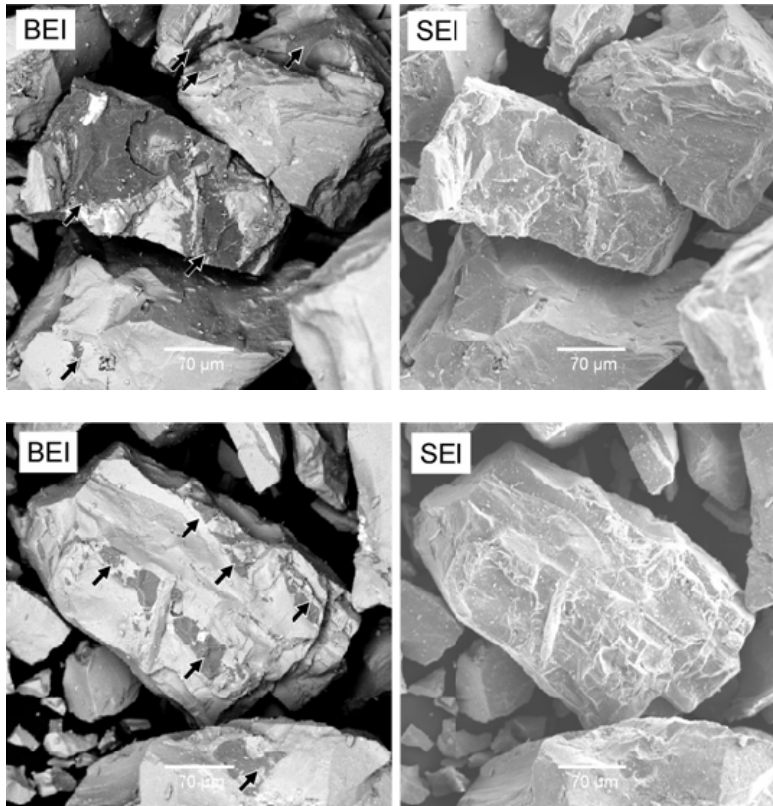
T<sub>solidus</sub> (calculated)

T<sub>tap</sub>

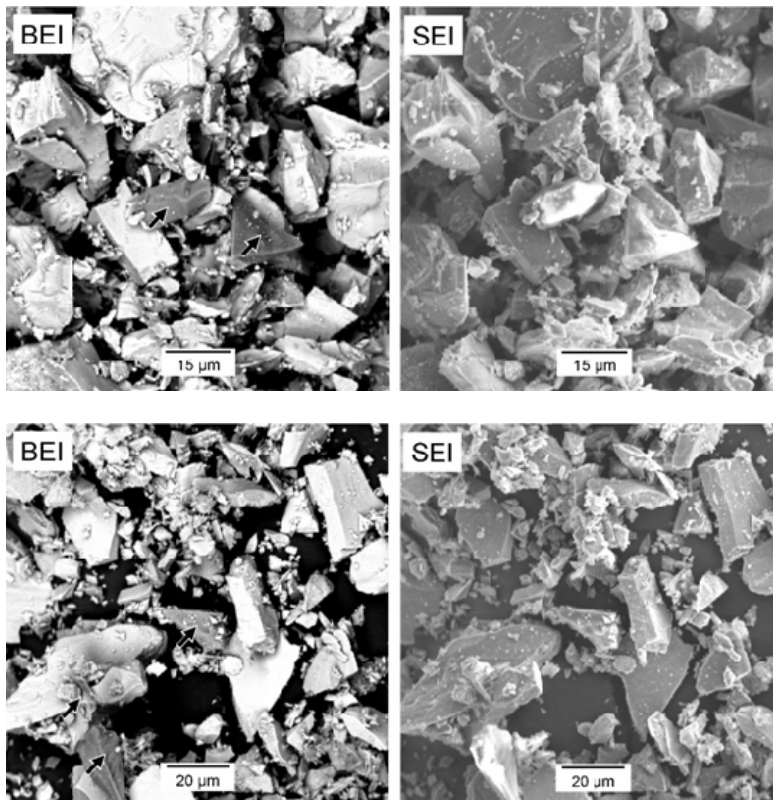
superheat

### 5.3 SEM examples of SiO<sub>2</sub> on particle surfaces

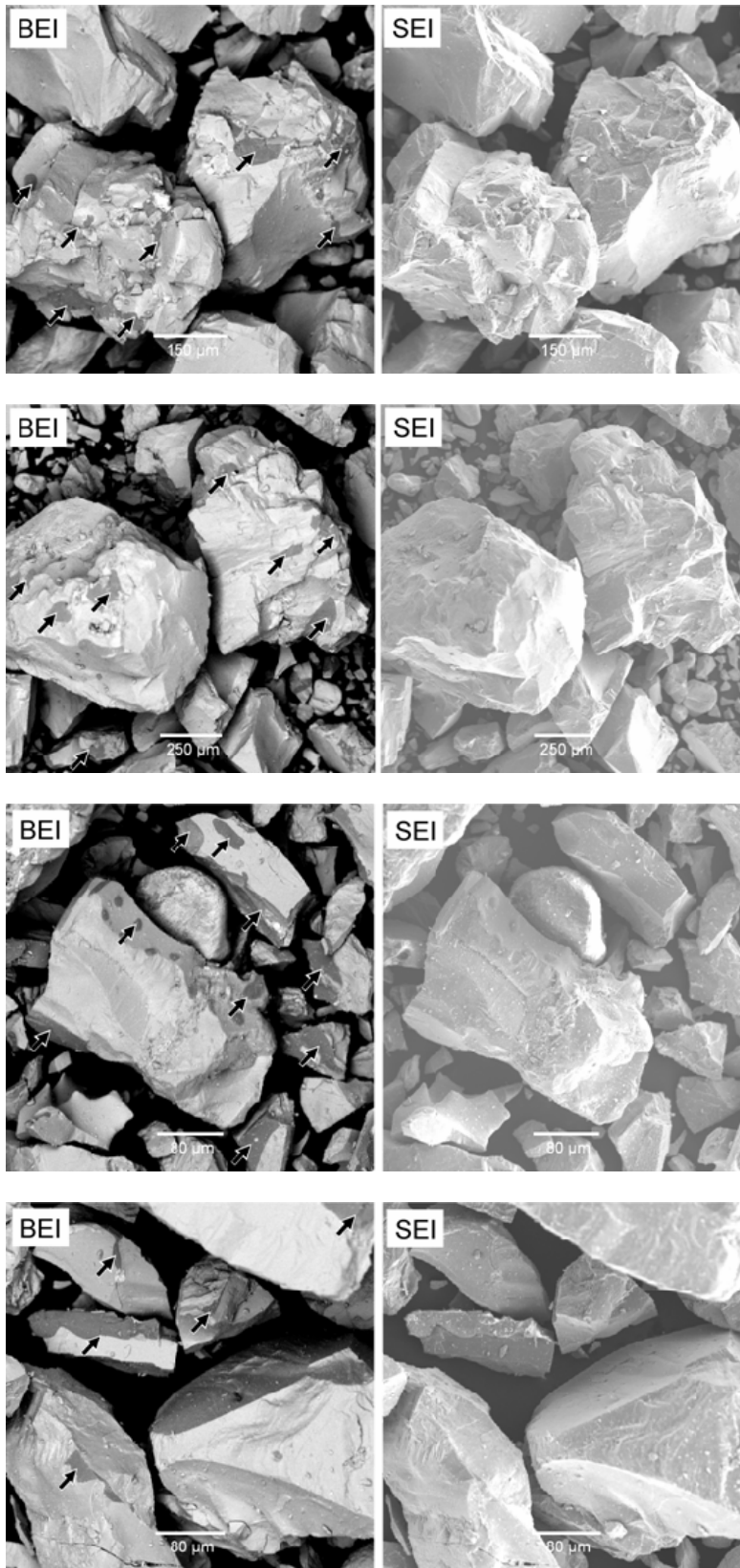
Coarse fraction (+850  $\mu\text{m}$  -106  $\mu\text{m}$ )



Fine fraction (-106  $\mu\text{m}$ )

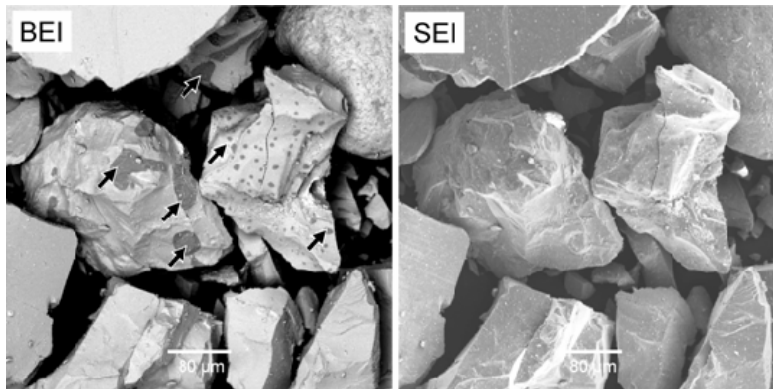


Coarse fraction

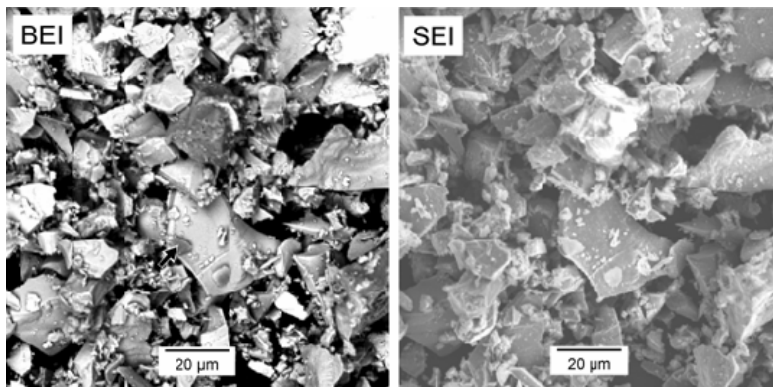
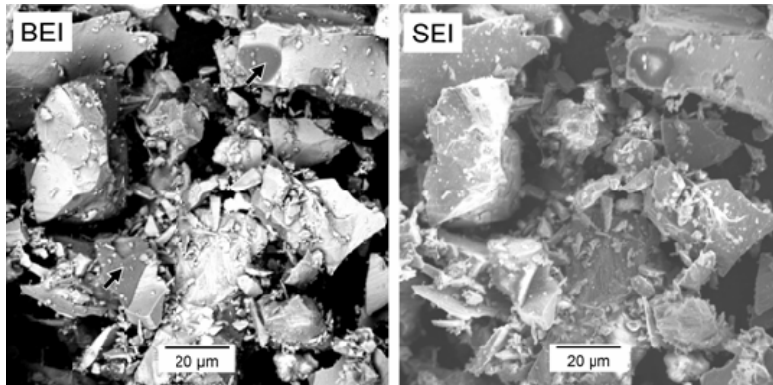
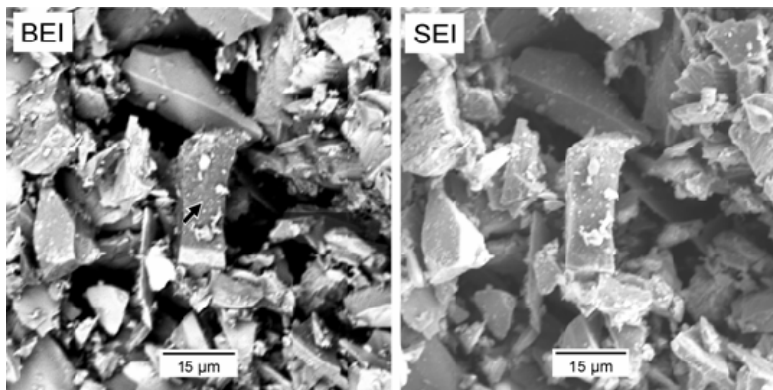




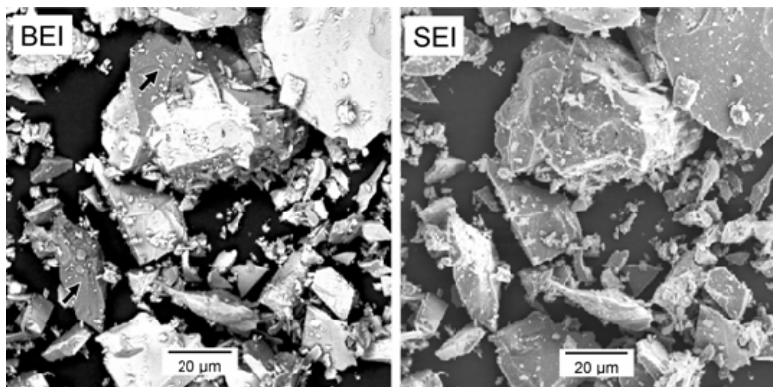
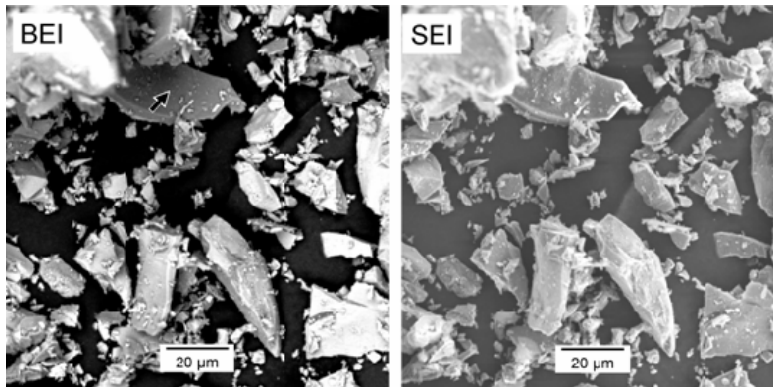
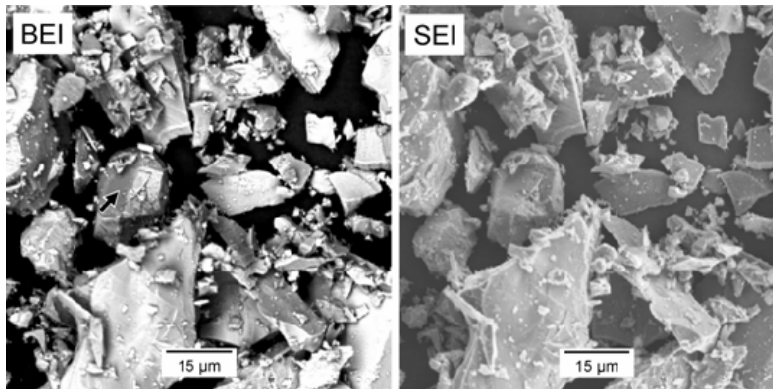
### Coarse fraction



### Fine fraction



Fine fraction



## 5.4 FlexPDE code for the one dimensional example

```

! #####
! Neumann problem.pde
!
!       Date:           2007-10-28
!       Developer:      Hanlie Kotze
!       Model Purpose:  Part of PhD studies. Objective is to have a numerical and analytical solutions of
! a semi-infinite volume which starts off at a temperature above it's liquidus and cools down to
! solidification.
!       The shell thickness as a function of time, and temperature profile at a given time, are compared
! for the two solutions. The analytical solution was done by Prof Chris Pistorius - "Neumann problem.xls".
! Equations are from Carslaw & Jaeger, 1959. The following is the numerical solution.
!       Units:           Temperature           [°C]
!                       Distance             [m]
!                       Time                 [s]
!                       Density              [kg/m3]
!                       Mass                 [kg]
!                       Heat Capacity        [J/(kg.°C)]
!                       Thermal Conductivity [W/(m.°C)]
!
!       References:
!       1       Carslaw & Jaeger, 1959
!       Assumptions:
!       1       Material properties are constant (not a function of temperature)
!       2
!
! #####

TITLE
    'Neumann problem'

COORDINATES
    CARTESIAN1

SELECT
    SMOOTHINIT
    ERRLIM = 1E-8
    NGRID = 10
    NODELIMIT = 100

VARIABLES
    temperature

DEFINITIONS
! #####
! Physical Properties
! #####
Cpsolid = 900                ! heat capacity of the solid [J/kgK]
Cpliquid = 1000              ! heat capacity of the liquid [J/kgK]
Hf = 650000                  ! heat of fusion [J/kg]
ksolid = 2                   ! thermal conductivity of the solid [W/mK]
kliquid = 4                  ! thermal conductivity of the liquid [W/mK]
density = 4000               ! material density [kg/m3]

! #####
! Boundary Conditions
! #####
Tsolidus = 1510
Tliquidus = 1530
Tinitial = 1550
Tsurface = 100
Maxtime = 10*24*3600
Domainlength = 2.0
Fliquid = max( min((temperature - Tsolidus) / (Tliquidus - Tsolidus),1.0), 0)
Fsolid = max( min((Tliquidus - temperature) / (Tliquidus - Tsolidus), 1), 0)

Cp = Cpsolid + SWAGE(temperature-Tsolidus,0,(Cpliquid-Cpsolid)/2+Hf/(Tliquidus-
Tsolidus),(Tliquidus - Tsolidus)/10) - SWAGE(temperature-Tliquidus,0,Hf/(Tliquidus-Tsolidus)+(Cpsolid-
Cpliquid)/2,(Tliquidus - Tsolidus)/10)

k = SWAGE(temperature-Tsolidus, ksolid, kliquid, (Tliquidus-Tsolidus))

INITIAL VALUES
    temperature = Tinitial

EQUATIONS
    temperature:    (density*Cp)*dt(temperature) - div(k*grad(temperature)) = 0.0

BOUNDARIES
    REGION 1
        START(0) POINT VALUE(temperature) = Tinitial
        LINE TO (Domainlength) POINT VALUE(temperature) = RAMP(t-50, Tinitial, Tsurface, 500)
    FEATURE 'Line'
        START(0) LINE TO (Domainlength)

TIME
    FROM 0 TO Maxtime

```



MONITORS

```
FOR t = 0,1,2,4,8,16,32,50 BY 100 TO ENDTIME
ELEVATION(temperature) ON 'Line'
ELEVATION(Fliquid, Fsolid) ON 'Line'
```

PLOTS

```
FOR t = 0, 3600, 7200, 14400, 28800, 57600, 115200, 230400, 460800, 921600
ELEVATION(temperature) ON 'Line' EXPORT FORMAT "#d, #1"
ELEVATION(Fliquid, Fsolid) ON 'Line' EXPORT FORMAT "#d, #1, #2"
```

HISTORIES

```
HISTORY(temperature) AT (0) AT (Domainlength*.2) AT (Domainlength*.4) AT (Domainlength*.6) AT
(Domainlength*.8) AT (Domainlength*.9) AT (Domainlength-0.1) AT (Domainlength-0.05) AT (Domainlength-0.02)
AT (Domainlength) EXPORT FORMAT "#t #r, #i"
```

```
HISTORY(Fliquid) AT (0) AT (Domainlength*.2) AT (Domainlength*.4) AT (Domainlength*.6) AT
(Domainlength*.8) AT (Domainlength*.9) AT (Domainlength-0.1) AT (Domainlength-0.05) AT (Domainlength-0.02)
AT (Domainlength) EXPORT FORMAT "#t #r, #i"
```

```
HISTORY(Cp) AT (0) AT (Domainlength*.2) AT (Domainlength*.4) AT (Domainlength*.6) AT
(Domainlength*.8) AT (Domainlength*.9) AT (Domainlength-0.1) AT (Domainlength-0.05) AT (Domainlength-0.02)
AT (Domainlength) EXPORT FORMAT "#t #r, #i"
```

```
HISTORY(k) AT (0) AT (Domainlength*.2) AT (Domainlength*.4) AT (Domainlength*.6) AT
(Domainlength*.8) AT (Domainlength*.9) AT (Domainlength-0.1) AT (Domainlength-0.05) AT (Domainlength-0.02)
AT (Domainlength) EXPORT FORMAT "#t #r, #i"
```

END

## 5.5 Flex PDE code for the slag block model

### File 1: slag block and pot system

```

! #####
! UP Slag Block Cooling 3a.pde
!
! Client:      Department of Material Science and Metallurgical Engineering
!              University of Pretoria
! Date:       2007-02-13
!
! Developer:   Johan Zietsman
!
! Model Type:  Transient
! Model Purpose: The model is part of Hanlie Kotze's Ph.D. study. Chris Pistorius requested Ex
Mente to assist with the development of a
!              model based on Hanlie's formulation. Hanlie would then use this model to
!              generate results required for the completion of her Ph.D.
!              The purpose of the model is to describe the cooling and solidification
!              of a high-TiO2 slag block as it moves through the various cooling stages
!              and cooling conditions on the plant.
!              The model must take into account the following factors:
!              - tap temperature,
!              - tap composition,
!              - cooling conditions.
!              The model must also be able to describe a small pilot-plant configuration
!              and the full-scale industrial configuration.
!
! Units:      Temperature      [°C]
!              Distance        [m]
!              Time             [s]
!              Fraction         []
!              Chemical Composition [%]
!              Density          [kg/m3]
!              Mass             [kg]
!              Heat Capacity    [J/(kg.°C)]
!              Thermal Conductivity [W/(m.°C)]
!
! #####

TITLE
      'UP Slag Block Cooling 6a'

COORDINATES
      YCYLINDER

SELECT
      ERRLIM = 1E-4
      SMOOTHINIT
      GRAY
      NODELIMIT = 1000

VARIABLES
      temperature (1700)

DEFINITIONS
! #####
! Constants
! #####

      g = 9.81          ! Gravitational acceleration      [m/s2]
      sigma = 5.669E-8 ! Stefan-Boltzmann constant      [W/(m2.K4)]
      slagDensity = 3800.0 ! Slag density      [kg/m3]
      steelDensity = 7600 ! Steel density      [kg/m3]

! #####
! Model Inputs
! #####

!%% TAP37
{
      slagFeOContent_Analysed = 13.25
      slagTemperature_AtTap = 1669.0
      tapMass = 1002.0}

!%% TAP38
{
      slagFeOContent_Analysed =13.92
      slagTemperature_AtTap = 1668.0
      tapMass = 1365.0}

!%% TAP42 C9 WATERcooling
{
      slagFeOContent_Analysed = 11.99
      slagTemperature_AtTap = 1709.0
      tapMass = 1017.0}

```



```
! %% Tap 64 C9 AIRcooling
{
  slagFeOContent_Analysed = 11.46
  slagTemperature_AtTap = 1683.0
  tapMass = 1420.0}

!
  Big tap
  slagFeOContent_Analysed = 10.10
  slagTemperature_AtTap = 1678.0
  tapMass = 18200

  potTemperature_BeforeTap = 25.0
  slagFeOContent = slagFeOContent_Analysed
  slagLiquidusTemperature = (0.2351*slagFeOContent^2 - 11.24*slagFeOContent + 1664.1)
  slagSolidusTemperature_0 = (0.0364*slagFeOContent^2 - 4.845*slagFeOContent + 1502.7)

  ! Domain Dimension Inputs (1.5 t)
  {
    potMassCapacity = 1500.0
    potR = 0.523
    potTheta_Degrees = 15.0
    potTheta_Radians = (potTheta_Degrees / 360.0) * 2.0 * PI
    potWallThickness = 0.04}

  ! Domain Dimension Inputs (20 t, dwg 615-BZ00 0002G)
  potMassCapacity = 20000.0
  potR = 1.175
  potTheta_Degrees = 15.0
  potTheta_Radians = (potTheta_Degrees / 360.0) * 2.0 * PI
  potWallThickness = 0.11
  ! #####
  ! Domain Dimensions
  ! #####

  ! Domain Dimension Calculations
  potVolume = potMassCapacity / slagDensity
  blockVolume = tapMass / slagDensity
  potPhi_Degrees = 90.0 - potTheta_Degrees
  potPhi_Radians = (potPhi_Degrees / 360.0) * 2.0 * PI
  blockL = ((potVolume - PI * potR ^ 3.0 / 3.0 / tan(potTheta_Radians)) / (-PI *
(cos(potTheta_Radians) ^ 3.0 / 3.0 / tan(potTheta_Radians) + PI / 3.0 * (2.0 + cos(potPhi_Radians)) * (1.0
- cos(potPhi_Radians) ^ 2.0)) ^ (1.0 / 3.0)
  blockrt = blockL*cos(potTheta_Radians)
  blockzt = blockL*(1.0 - sin(potTheta_Radians)) - blockL
  blockVolume_SphericalCap = (PI/3.0) * blockL^3.0 * (2.0 + cos(potPhi_Radians)) * (1.0 -
cos(potPhi_Radians))^2.0
  blockVolume_ConicalSection = (PI/3.0) / tan(potTheta_Radians) * (potR^3.0 - blockrt^3.0)
  blockR = ((3.0*tan(potTheta_Radians)/PI)*(blockVolume - blockVolume_SphericalCap) +
blockrt^3.0)^(1.0/3.0)
  blockZ = (blockR - blockrt)/tan(potTheta_Radians) + blockzt + blockL
  ! #####
  ! Conditions
  ! #####

  ambientTemperature = 25.0

  ! #####
  ! Physical Properties
  ! #####

  k = 0.0 ! Thermal conductivity
  Cp = 0.0 ! Heat capacity

  slagk = (max(0.00175*temperature+0.3,0.3))*1

  slagCp_Solid = - 0.0314*slagFeOContent^2 - 0.4042*slagFeOContent^2 + 908.51
  slagCp_Liquid = 0.0561*slagFeOContent^2 - 3.3668*slagFeOContent + 1044.3
  slagEnthalpy_At25 = - 139.51*slagFeOContent^2 - 1086.1*slagFeOContent + 515805
  CpSolidify = (slagEnthalpy_At25 + slagCp_Liquid*(slagLiquidusTemperature - 25) -
slagCp_Solid*(slagSolidusTemperature_0 - 25)) / (slagLiquidusTemperature-slagSolidusTemperature_0)

  slagCp= IF (temperature <= slagSolidusTemperature_0) THEN slagCp_Solid
  ELSE IF (temperature >= slagLiquidusTemperature) THEN slagCp_Liquid
  ELSE (slagEnthalpy_At25 + slagCp_Liquid*(slagLiquidusTemperature - 25) -
slagCp_Solid*(slagSolidusTemperature_0 - 25)) / (slagLiquidusTemperature-slagSolidusTemperature_0)
  slagEnthalpy = IF (temperature <= slagSolidusTemperature_0) THEN slagCp_Solid*(temperature-25)
  ELSE IF (temperature >= slagLiquidusTemperature) THEN
slagEnthalpy_At25+slagCp_Liquid*(temperature-25)
  ELSE slagCp_Solid*(slagSolidusTemperature_0-25)+CpSolidify*(temperature-
slagSolidusTemperature_0)
  ! Steel
  steelCp = 465.0/slagDensity*steelDensity ! Carbon Steel 0.5% (Holman)

  steelk = -0.03488*(temperature) + 59.1 ! cast steel, Holman
  ! #####
  ! Initial Conditions
  ! #####

  initialTemperature = 25.0
```



```
initialLiquidFraction = min(max((temperature - slagSolidusTemperature_0)/(slagLiquidusTemperature - slagSolidusTemperature_0),0),1)

!TRANSFER("transfer_1.dat", initialTemperature, initialLiquidFraction)
! #####
! Boundary Conditions
! #####

! Water cooling.
waterVapourDensity = 0.5863 ! At 107°C (Holman)
waterLatentHeat = 2256000.0 ! Latent heat of water to gas reaction, J/kg (internet data)

waterSurfaceTension = 0.0588 ! [N/m] (internet data)
waterDensity = 994.9 ! At 32°C (Holman)
waterCp = 4174.0 ! (Holman)
waterSaturationTemperature = 100.0
waterPr = 0.9 ! Prandtl number of water
waterk = 0.68 ! Thermal conductivity of water
waterViscosity = 0.003 ! Kinematic viscosity of water in m2/s

coolingwaterVolumetricFlowRate = 1.0 / 3600.0 ! m3/hr to m3/s
coolingwaterDropSpeed = 1.0 ! m/s
coolingwaterDropDiameter = 1.0 / 1000.0 ! mm to m
coolingwaterWaterTemperature = ambientTemperature
coolingwaterBlockArea = 2 * Pi * blockL ^ 2 * (1 - COS(potPhi_Radians)) + Pi * (blockrt + blockR) *
SQRT((blockZ - blockzt) ^ 2 + (blockR - blockrt) ^ 2)
coolingwaterVolumetricFlux = coolingwaterVolumetricFlowRate / coolingwaterBlockArea

coolingwaterRe = coolingwaterVolumetricFlux * coolingwaterDropDiameter / waterViscosity
coolingwaterNu = 2.512 * coolingwaterRe ^ 0.76 * waterPr ^ 0.56
coolingwaterTIncip = 13.43 * coolingwaterRe ^ 0.167 * waterPr ^ 0.123 * (waterk /
coolingwaterDropDiameter) ^ 0.22 + coolingwaterWaterTemperature
coolingwaterTcrit = 18 * ((waterVapourDensity * waterLatentHeat * coolingwaterVolumetricFlux) *
(waterSurfaceTension / (waterDensity * coolingwaterVolumetricFlux ^ 2 * coolingwaterDropDiameter)) ^ 0.198)
^ (1 / 5.55) + coolingwaterWaterTemperature
coolingwaterTmin = 204.895 * coolingwaterVolumetricFlux ^ 0.066 * coolingwaterDropSpeed ^ 0.138 *
coolingwaterDropDiameter ^ (-0.035) + coolingwaterWaterTemperature
coolingwaterTdfb = 280.762 * coolingwaterVolumetricFlux ^ 0.087 * coolingwaterDropSpeed ^ 0.11 *
coolingwaterDropDiameter ^ (-0.035) + coolingwaterWaterTemperature
coolingwaterHFdfb = 6100300.0 * coolingwaterVolumetricFlux ^ 0.588 * coolingwaterDropSpeed ^ 0.244
coolingwaterHFfmin = 3324400.0 * coolingwaterVolumetricFlux ^ 0.544 * coolingwaterDropSpeed ^ 0.324
coolingwaterHFfcrit = waterVapourDensity * waterLatentHeat * coolingwaterVolumetricFlux * 122.4 * (1
+ 0.0118 * (waterDensity / waterVapourDensity) ^ 0.25 * (waterDensity * waterCp *
(waterSaturationTemperature - coolingwaterWaterTemperature) / (waterVapourDensity * waterLatentHeat))) *
(waterSurfaceTension / (waterDensity * coolingwaterVolumetricFlux ^ 2 * coolingwaterDropDiameter)) ^
0.0198

coolingwaterN2 = (coolingwaterHFdfb - coolingwaterHFfmin) / (coolingwaterTdfb - coolingwaterTmin) ^
2
coolingwaterEN1 = -2 * coolingwaterN2 * coolingwaterTmin
coolingwaterN0 = coolingwaterHFfmin - coolingwaterEN1 * coolingwaterTmin - coolingwaterN2 *
coolingwaterTmin ^ 2

coolingwaterh = IF temperature > coolingwaterTdfb THEN
(63.25 * (temperature - coolingwaterWaterTemperature) ^ 1.691 *
coolingwaterVolumetricFlux ^ 0.264 / coolingwaterDropDiameter ^ 0.062) / (temperature -
coolingwaterWaterTemperature)
ELSE IF temperature = coolingwaterTdfb THEN
coolingwaterHFdfb / (temperature - coolingwaterWaterTemperature)
ELSE IF (temperature < coolingwaterTdfb) AND (temperature >
coolingwaterTmin) THEN
((coolingwaterN0 + coolingwaterEN1 * (temperature -
coolingwaterWaterTemperature) + coolingwaterN2 * (temperature - coolingwaterWaterTemperature) ^ 2)) /
(temperature - coolingwaterWaterTemperature)
ELSE IF temperature = coolingwaterTmin THEN
coolingwaterHFfmin / (temperature - coolingwaterWaterTemperature)
ELSE IF (temperature < coolingwaterTmin) AND (temperature >
coolingwaterTcrit) THEN
(coolingwaterHFfcrit - (coolingwaterHFfcrit - coolingwaterHFfmin) /
(coolingwaterTcrit - coolingwaterTmin) ^ 3 * (coolingwaterTcrit ^ 3 - 3 * coolingwaterTcrit ^ 2 *
coolingwaterTmin + 6 * coolingwaterTcrit * coolingwaterTmin * (temperature - coolingwaterWaterTemperature)
- 3 * (coolingwaterTcrit + coolingwaterTmin) * (temperature - coolingwaterWaterTemperature) ^ 2 + 2 *
(temperature - coolingwaterWaterTemperature) ^ 3)) / (temperature - coolingwaterWaterTemperature)
ELSE IF temperature = coolingwaterTcrit THEN
coolingwaterHFfcrit / (temperature - coolingwaterWaterTemperature)
ELSE IF (temperature < coolingwaterTcrit) AND (temperature >
coolingwaterTincip) THEN
(0.0000187 * (temperature - coolingwaterWaterTemperature) ^ 5.55)
/ (temperature - coolingwaterWaterTemperature)
ELSE coolingwaterNu * waterk / coolingwaterDropDiameter

! Air cooling.
solidsurfaceEmissivity = 0.8
coolingairTf = 0.5 * (temperature + ambientTemperature)
coolingairkf = 0.00005581 * coolingairTf + 0.02694
coolingairb = 2 / (ambientTemperature + 273.15 + temperature + 273.15)
coolingairv = 0.0000000006054 * coolingairTf ^ 2 + 0.000000102 * coolingairTf + 0.0000121
```



```
coolingairPr = -9.11E-17 * coolingairTf ^ 5 + 5.007E-13 * coolingairTf ^ 4 - 0.00000001021 *
coolingairTf ^ 3 + 0.000000916 * coolingairTf ^ 2 - 0.0003139 * coolingairTf + 0.7153
coolingairhRadiation = solidsurfaceEmissivity * 5.669 * (((temperature + 273.15) / 100) ^ 4 -
((ambientTemperature + 273) / 100) ^ 4) / (temperature - ambientTemperature) !equation 8-126, p471, Holman

coolingairHorizontal_L = 0.25 * blockR * 2
coolingairHorizontal_Gr = g * coolingairb * (temperature - ambientTemperature) *
coolingairHorizontal_L ^ 3 / coolingairv ^ 2
coolingairHorizontal_C = IF coolingairHorizontal_Gr <= 8.0E6 THEN 0.54 ELSE 0.15
coolingairHorizontal_m = IF coolingairHorizontal_Gr <= 8.0E6 THEN 1.0/4.0 ELSE 1.0/3.0
coolingairHorizontal_hConvection = (coolingairkf / coolingairHorizontal_L) * coolingairHorizontal_C
*( (MAX(coolingairHorizontal_Gr, 0.0) * coolingairPr) ^ coolingairHorizontal_m)
coolingairHorizontal_h = coolingairhRadiation + coolingairHorizontal_hConvection

coolingairVertical_L = Cos(potTheta_Radians) / blockZ
coolingairVertical_Gr = (g * coolingairb * (temperature - ambientTemperature) *
coolingairVertical_L ^ 3 / coolingairv ^ 2) * (Cos(potTheta_Radians)) ^ 2
coolingairVertical_C = IF coolingairVertical_Gr <= 1.0E9 THEN 0.59 ELSE 0.1
coolingairVertical_m = IF coolingairVertical_Gr <= 1.0E9 THEN 1.0/4.0 ELSE 1.0/3.0
coolingairVertical_hConvection = coolingairkf / coolingairVertical_L * coolingairVertical_C *
(coolingairVertical_Gr * coolingairPr) ^ coolingairVertical_m
coolingairVertical_h = coolingairhRadiation + coolingairVertical_hConvection

! Boundary Condition Schedule
endTimeCoolingStep00 = 3600*18 ! Block in pot.

endTimeCoolingStep01 = endTimeCoolingStep00 + 3600
endTimeCoolingStep02 = endTimeCoolingStep01 + 3600
endTimeCoolingStep03 = endTimeCoolingStep02 + 3600
endTimeCoolingStep04 = endTimeCoolingStep03 + 3600
endTimeCoolingStep05 = endTimeCoolingStep04 + 3600
endTimeCoolingStep06 = endTimeCoolingStep05 + 3600
endTimeCoolingStep07 = endTimeCoolingStep06 + 3600
endTimeCoolingStep08 = endTimeCoolingStep07 + 3600
endTimeCoolingStep09 = endTimeCoolingStep08 + 3600
endTimeCoolingStep10 = endTimeCoolingStep09 + 3600

! Boundary Conditions
horizontalAirCooling = URAMP(t - 0.0, t - 5.0) * coolingairHorizontal_h * (temperature -
ambientTemperature)
verticalAirCooling = coolingairVertical_h * (temperature - ambientTemperature)
verticalWaterCooling = coolingwaterh * (temperature - coolingwaterWaterTemperature)

blockVerticalCooling = IF t <=endTimeCoolingStep00 THEN
{
0.0
ELSE IF t <=endTimeCoolingStep01 THEN
verticalAirCooling
ELSE IF t <=endTimeCoolingStep02 THEN
verticalWaterCooling
ELSE IF t <=endTimeCoolingStep03 THEN
verticalAirCooling
ELSE IF t <=endTimeCoolingStep04 THEN
verticalWaterCooling
ELSE IF t <=endTimeCoolingStep05 THEN
verticalAirCooling
ELSE IF t <=endTimeCoolingStep06 THEN
verticalWaterCooling
ELSE IF t <=endTimeCoolingStep07 THEN
verticalAirCooling
ELSE IF t <=endTimeCoolingStep08 THEN
verticalWaterCooling
ELSE IF t <=endTimeCoolingStep09 THEN
verticalAirCooling
ELSE IF t <=endTimeCoolingStep10 THEN
verticalWaterCooling}
ELSE 0.0

blockHorizontalCooling = IF t <=endTimeCoolingStep00 THEN
{
horizontalAirCooling
ELSE IF t <=endTimeCoolingStep01 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep02 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep03 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep04 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep05 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep06 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep07 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep08 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep09 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep10 THEN
```





```

                                horizontalContactCooling}
ELSE                               0.0

potVerticalCooling =             IF      t <=endTimeCoolingStep00 THEN
                                verticalAirCooling
                                ELSE     verticalAirCooling

! Other Calculations
liquidFraction = min(max((temperature -slagSolidusTemperature_0)/( slagLiquidusTemperature -
slagSolidusTemperature_0),0),1)

INITIAL VALUES
temperature = initialTemperature
!liquidFraction = initialLiquidFraction

EQUATIONS
temperature:      slagDensity*Cp*dt(temperature) - div(k*grad(temperature)) = 0.0

BOUNDARIES

REGION 1 'Block'
initialTemperature = slagTemperature_AtTap
k = slagk
Cp = slagCp

                                START                (0.0, -blockL)
NOBC(temperature)          ARC(CENTER=0.0, 0.0) TO (blockrt, blockzt)
NOBC(temperature)          LINE    TO    (blockR, blockZ - blockL)
NATURAL(temperature)=blockHorizontalCooling LINE    TO    (0.0, blockZ - blockL)
NOBC(temperature)          LINE    TO    CLOSE

{REGION 1 'Block'
k = slagk
Cp = slagCp

                                START                (0.0, -blockL)
NATURAL(temperature)=blockVerticalCooling ARC(CENTER=0.0, 0.0) TO (blockrt, blockzt)
NATURAL(temperature)=blockVerticalCooling LINE    TO    (blockR, blockZ - blockL)
NATURAL(temperature)=blockHorizontalCooling LINE    TO    (0.0, blockZ - blockL)
NOBC(temperature)          LINE    TO    CLOSE}

REGION 2 'Pot'
initialTemperature = potTemperature_BeforeTap
k = steelk
Cp = steelCp

                                START                (0.0, -blockL)
NOBC(temperature)          ARC(CENTER=0.0, 0.0) TO (blockrt, blockzt)
NOBC(temperature)          LINE    TO    (blockR, blockZ - blockL)
NATURAL(temperature)=horizontalAirCooling LINE    TO    (blockR + potWallThickness /
cos(potTheta_Radians), blockZ - blockL)
NATURAL(temperature)=verticalAirCooling LINE    TO    (blockrt +
potWallThickness * sin(potPhi_Radians), blockzt - potWallThickness * cos(potPhi_Radians))
NATURAL(temperature)=verticalAirCooling ARC(CENTER=0.0, 0.0) TO (0.0, -blockL -
potWallThickness)
NOBC(temperature)          LINE    TO    CLOSE

FEATURE 'Block Vertical Outer Surface'
START                (0.0, -blockL)
ARC(CENTER=0.0, 0.0) TO (blockrt, blockzt)
LINE    TO    (blockR, blockZ - blockL)

FEATURE 'Pot Vertical Outer Surface'
START                (0.0, -blockL - potWallThickness)
ARC(CENTER=0.0, 0.0) TO (blockrt + potWallThickness * sin(potPhi_Radians), blockzt -
potWallThickness * cos(potPhi_Radians))
LINE    TO    (blockR + potWallThickness / cos(potTheta_Radians), blockZ - blockL)

FEATURE 'Pot Horizontal edge'
START                (blockR + potWallThickness / cos(potTheta_Radians), blockZ - blockL)
LINE    TO    (blockR, blockZ - blockL)

FEATURE 'Block Horizontal Outer Surface'
START                (blockR, blockZ - blockL)
LINE    TO    (0.0, blockZ - blockL)

FEATURE 'Block Centre Line'
START                (0.0, -blockL)
LINE TO (0.0, -blockL + blockZ)

TIME
0.0 BY 1E-1 TO endTimeCoolingStep00
{endTimeCoolingStep00 BY 1E-1 TO endTimeCoolingStep10}

MONITORS
FOR t= 0, 0.1, 0.5, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 15, 20, 25, 30, 45, 60, 300, 600 BY 600 TO 3600
BY 3600 TO ENDTIME
CONTOUR(temperature) RANGE=(0, 1700)
! CONTOUR(temperature) PAINTED RANGE=(0, 1700)
! CONTOUR(liquidFraction2) PAINTED RANGE=(0.0, 1.0)
! CONTOUR(liquidFraction) PAINTED fixed RANGE=(0.0, 1.0)
```



```
! ELEVATION(temperature) ON 'Block Vertical Outer Surface'
! ELEVATION(verticalAirCooling) ON 'Block Vertical Outer Surface'
! ELEVATION(temperature) ON 'Block Horizontal Outer Surface'
! ELEVATION(horizontalAirCooling) ON 'Block Horizontal Outer Surface'
! ELEVATION(slagk) ON 'block horizontal outer surface'
! ELEVATION(steelCp, steelk, temperature) ON 'Block Vertical Outer Surface'

PLOTS
! FOR t= 0 BY 24*3600 TO ENDTIME

! CONTOUR(temperature) RANGE=(0, 1700)
! CONTOUR(temperature) PAINTED RANGE=(0, 1700)
! CONTOUR(max(min(liquidFraction, 1.0), 0.0)) PAINTED RANGE=(0.0, 1.0)
! ELEVATION(temperature) ON 'Block Vertical Outer Surface' EXPORT FORMAT "#d,#1"
! ELEVATION(verticalAirCooling) ON 'Block Vertical Outer Surface'
! ELEVATION(temperature) ON 'Block Horizontal Outer Surface'
! ELEVATION(horizontalAirCooling) ON 'Block Horizontal Outer Surface'
! TABLE(temperature) EXPORT FORMAT "#t,#x,#y,#1"

! PLANT BLOCK
! FOR t = 0, 0.25, 0.5, 1, 2, 4, 8, 16, 32, 64, 100, 200, 400, 800, 1600, 3200
! ELEVATION(liquidFraction) ON 'Block Centre Line' EXPORT FORMAT "#y, #1"
! ELEVATION(verticalAirCooling) ON 'Block Vertical Outer Surface' EXPORT FORMAT "#d, #1"
! ELEVATION(horizontalAirCooling) ON 'Block Horizontal Outer Surface' EXPORT FORMAT "#d, #1"
! ELEVATION(coolingairHorizontal_h) ON 'Block Horizontal Outer Surface' EXPORT FORMAT "#d, #1"
! ELEVATION(steelk) ON 'Block Vertical Outer Surface' EXPORT FORMAT "#d, #1"

SUMMARY
REPORT potMassCapacity
REPORT tapMass
REPORT potR
REPORT potTheta_Degrees
REPORT potTheta_Radians
REPORT potWallThickness

REPORT potVolume
REPORT blockVolume
REPORT potPhi_Degrees
REPORT potPhi_Radians
REPORT blockL
REPORT blockrt
REPORT blockzt
REPORT blockVolume_SphericalCap
REPORT blockVolume_ConicalSection
REPORT blockR
REPORT blockZ
REPORT slagSolidusTemperature_0
REPORT slagLiquidusTemperature

FOR t = ENDTIME
! TRANSFER(temperature, liquidFraction) FILE="transfer.dat"
! TRANSFER(temperature) FILE="transfer.dat"

HISTORIES
! HISTORY(SURF_INTEGRAL(blockVerticalCooling, 'Block Vertical Outer Surface') +
SURF_INTEGRAL(potVerticalCooling, 'Pot Vertical Outer Surface') ) AS 'HeatLossV' EXPORT FORMAT "#t#r,#i"
! HISTORY(SURF_INTEGRAL(blockHorizontalCooling, 'Block Horizontal Outer Surface') +
SURF_INTEGRAL(horizontalAirCooling, 'Pot Horizontal edge')) AS 'HeatLossH' EXPORT FORMAT "#t#r,#i"
! HISTORY(VOL_INTEGRAL(slagEnthalpy*slagDensity, 1)+VOL_INTEGRAL(steelCp*slagDensity*(temperature-
25), 2)) AS 'TotalEnergy' EXPORT FORMAT "#t#r,#i"
{Tap 37 & 38 C10 thermocouple positions}
! HISTORY(temperature) AT (0,0.13) AT(0,-0.020) AT(0,-0.120) AS 'ThermocouplePositions' EXPORT
FORMAT "#t#r,#i"

! TAP 42 & / 38
! HISTORY(SURF_INTEGRAL(verticalAirCooling, 'Block vertical outer surface')) AS 'Heatlosses from Pot
surface' EXPORT FORMAT "#t,#1"
! HISTORY(temperature) AT (0.0, -blockL - potWallThickness) AT(blockrt + potWallThickness *
sin(potPhi_Radians), blockzt - potWallThickness * cos(potPhi_Radians)) AT(blockR + potWallThickness /
cos(potTheta_Radians), blockZ - blockL) AS 'Pot surface temperatures' EXPORT FORMAT "#t#r,#i"

! PLANT TAPS
! HISTORY(temperature) AT (0.0, blockZ - blockL) AT (blockR, blockZ - blockL) AT (blockrt, blockzt)
AT (0.0, -blockL) AS 'Block surface temperatures' EXPORT FORMAT "#t #r, #i"

END
```



## File 2: slag block

```
! #####
! UP Slag Block Cooling 3b.pde
!
! Client: Department of Material Science and Metallurgical Engineering
! University of Pretoria
! Date: 2007-02-13
! Drawing: ???
!
! Developer: Johan Zietsman
!
! Model Type: Transient
! Model Purpose: The model is part of Hanlie Kotze's Ph.D. study. Chris Pistorius requested Ex
Mente to assist with the development of a
! model based on Hanlie's formulation. Hanlie would then use this model to
! generate results required for the completion of her Ph.D.
! The purpose of the model is to describe the cooling and solidification
! of a high-TiO2 slag block as it moves through the various cooling stages
! and cooling conditions on the plant.
! The model must take into account the following factors:
! - tap temperature,
! - tap composition,
! - cooling conditions.
! The model must also be able to describe a small pilot-plant configuration
! and the full-scale industrial configuration.
!
! Units: Temperature [°C]
! Distance [m]
! Time [s]
! Fraction []
! Chemical Composition [%]
! Density [kg/m3]
! Mass [kg]
! Heat Capacity [J/(kg.°C)]
! Thermal Conductivity [W/(m.°C)]
! #####

TITLE
    'UP Slag Block Cooling 5b'

COORDINATES
    YCYLINDER

SELECT
    ERRLIM = 1E-4
    SMOOTHINIT
    GRAY

VARIABLES
    temperature (1700)

DEFINITIONS
    ! #####
    ! Constants
    ! #####

    g = 9.81 ! Gravitational acceleration [m/s2]
    sigma = 5.669E-8 ! Stefan-Boltzmann constant [W/(m2.K4)]
    slagDensity = 3800.0 ! Slag density [kg/m3]

    ! #####
    ! Model Inputs
    ! #####

!
! %% Tap 37
! {
!     slagFeOContent_Analysed = 13.25
!     slagTemperature_AtTap = 1669.0
!     tapmass = 1002.00}

!
! %% Tap 38
! {
!     slagFeOContent_Analysed = 13.92
!     slagTemperature_AtTap = 1668.0
!     tapmass = 1365.00}

!
! %%% TAP42 C9 WATERcooling
! {
!     slagFeOContent_Analysed = 11.99
!     slagTemperature_AtTap = 1709.0
!     tapMass = 1017.0}

!
! %%% Tap 64 C9 AIRcooling
! {
!     slagFeOContent_Analysed = 11.46
!     slagTemperature_AtTap = 1683.0
!     tapMass = 1420.0}
```



```
! Big tap
slagFeOContent_Analysed = 10.10
slagTemperature_AtTap = 1678.0
tapMass = 18200.0

potTemperature_BeforeTap = 25.0
slagFeOContent = slagFeOContent_Analysed
slagLiquidusTemperature = (0.2351*slagFeOContent^2 - 11.24*slagFeOContent + 1664.1)
slagSolidusTemperature_0 = (0.0364*slagFeOContent^2 - 4.845*slagFeOContent + 1502.7)

! Domain Dimension Inputs (1.5 t)
{
potMassCapacity = 1500.0
potR = 0.523
potTheta_Degrees = 15.0
potTheta_Radians = (potTheta_Degrees / 360.0) * 2.0 * PI
potWallThickness = 0.04}

! Domain Dimension Inputs (20 t)
potMassCapacity = 20000.0
potR = 1.175
potTheta_Degrees = 15.0
potTheta_Radians = (potTheta_Degrees / 360.0) * 2.0 * PI
potWallThickness = 0.11

! #####
! Domain Dimensions
! #####

! Domain Dimension Calculations
potVolume = potMassCapacity / slagDensity
blockVolume = tapMass / slagDensity
potPhi_Degrees = 90.0 - potTheta_Degrees
potPhi_Radians = (potPhi_Degrees / 360.0) * 2.0 * PI
blockL = ((potVolume - PI * potR ^ 3.0 / 3.0 / tan(potTheta_Radians)) / (-PI *
(cos(potTheta_Radians) ^ 3.0 / 3.0 / tan(potTheta_Radians) + PI / 3.0 * (2.0 + cos(potPhi_Radians)) * (1.0
- cos(potPhi_Radians)) ^ 2.0)) ^ (1.0 / 3.0)
blockrt = blockL*cos(potTheta_Radians)
blockzt = blockL*(1.0 - sin(potTheta_Radians)) - blockL
blockVolume_SphericalCap = (PI/3.0) * blockL^3.0 * (2.0 + cos(potPhi_Radians)) * (1.0 -
cos(potPhi_Radians))^2.0
blockVolume_ConicalSection = (PI/3.0) / tan(potTheta_Radians) * (potR^3.0 - blockrt^3.0)
blockR = ((3.0*tan(potTheta_Radians)/PI)*(blockVolume - blockVolume_SphericalCap) +
blockrt^3.0)^(1.0/3.0)
blockZ = (blockR - blockrt)/tan(potTheta_Radians) + blockzt + blockL

! #####
! Conditions
! #####

ambientTemperature = 25.0

! #####
! Physical Properties
! #####

k = 0.0 ! Thermal conductivity
Cp = 0.0 ! Heat capacity

! Slag
slagk = (max((0.00175*temperature+0.3),0.3))*1

slagCp_Solid = - 0.0314*slagFeOContent^2 - 0.4042*slagFeOContent^2 + 908.51
slagCp_Liquid = 0.0561*SlagFeOContent^2 - 3.3668*slagFeOContent + 1044.3
slagEnthalpy_At25 = - 139.51*slagFeOContent^2 - 1086.1*slagFeOContent + 515805

slagCp= if (temperature <= slagSolidusTemperature_0) THEN slagCp_Solid
ELSE IF (temperature >= slagLiquidusTemperature) THEN slagCp_Liquid
ELSE (slagEnthalpy_At25 + slagCp_Liquid*(slagLiquidusTemperature - 25) -
slagCp_Solid*(slagSolidusTemperature_0 - 25)) / (slagLiquidusTemperature-slagSolidusTemperature_0)

! Steel
! steelk = 54.0 ! Carbon Steel 0.5% (Holman)
! steelCp = 465.0 ! Carbon Steel 0.5% (Holman)

! #####
! Initial Conditions
! #####

!initialTemperature = 25.0
TRANSFER("transfer_1.dat", initialTemperature)

! #####
! Boundary Conditions
! #####

! Water cooling.
waterVapourDensity = 0.5863 ! At 107°C (Holman)
```





```
coolingairHorizontal_C = IF coolingairHorizontal_Gr <= 8.0E6 THEN 0.54 ELSE 0.15
coolingairHorizontal_m = IF coolingairHorizontal_Gr <= 8.0E6 THEN 1.0/4.0 ELSE 1.0/3.0
coolingairHorizontal_hConvection = coolingairkf / coolingairHorizontal_L * coolingairHorizontal_C *
(coolingairHorizontal_Gr * coolingairPr) ^ coolingairHorizontal_m
coolingairHorizontal_h = coolingairhRadiation + coolingairHorizontal_hConvection

coolingairVertical_L = Cos(potTheta_Radians) / blockZ
coolingairVertical_Gr = (g * coolingairb * (temperature - ambientTemperature) *
coolingairVertical_L ^ 3 / coolingairv ^ 2) * (Cos(potTheta_Radians)) ^ 2
coolingairVertical_C = IF coolingairVertical_Gr <= 1.0E9 THEN 0.59 ELSE 0.1
coolingairVertical_m = IF coolingairVertical_Gr <= 1.0E9 THEN 1.0/4.0 ELSE 1.0/3.0
coolingairVertical_hConvection = coolingairkf / coolingairVertical_L * coolingairVertical_C *
(coolingairVertical_Gr * coolingairPr) ^ coolingairVertical_m
coolingairVertical_h = coolingairhRadiation + coolingairVertical_hConvection

! Boundary Condition Schedule

!
PLANT BLOCKS LANE 1
{
endTimeCoolingStep00 = 3600*17.75 ! Block in pot.
endTimeCoolingStep01 = endTimeCoolingStep00 + 3600*0.25 ! Air cooling
endTimeCoolingStep02 = endTimeCoolingStep01 + 3600*6.6
endTimeCoolingStep03 = endTimeCoolingStep02 + 3600*0.1
endTimeCoolingStep04 = endTimeCoolingStep03 + 3600*75.7
endTimeCoolingStep05 = endTimeCoolingStep04 + 3600*0.1
endTimeCoolingStep06 = endTimeCoolingStep05 + 3600*0.1
endTimeCoolingStep07 = endTimeCoolingStep06 + 3600*58.8
endTimeCoolingStep08 = endTimeCoolingStep07 + 3600*0.3
endTimeCoolingStep09 = endTimeCoolingStep08 + 3600*7.2
endTimeCoolingStep10 = endTimeCoolingStep09 + 3600*0.5
endTimeCoolingStep11 = endTimeCoolingStep10 + 3600*145.2
endTimeCoolingStep12 = endTimeCoolingStep11 + 3600*0.12
endTimeCoolingStep13 = endTimeCoolingStep12 + 3600
endTimeCoolingStep14 = endTimeCoolingStep13 + 3600}

!
PLANT BLOCKS LANE 2
endTimeCoolingStep00 = 3600*18 ! Block in pot.
endTimeCoolingStep01 = endTimeCoolingStep00 + 3600*24*15 ! Air cooling
! endTimeCoolingStep02 = endTimeCoolingStep00 + 3600*24*15 ! Water cooling
! endTimeCoolingStep03 = endTimeCoolingStep02 + 3600*1 ! Air cooling

!
TAP 42 AIR & WATER COOLING
{
endTimeCoolingStep00 = 3600*17.75 ! Block in pot.
endTimeCoolingStep01 = endTimeCoolingStep00 + 3600*0.25 ! Air cooling
endTimeCoolingStep02 = endTimeCoolingStep01 + 3600*15 ! Water cooling
endTimeCoolingStep03 = endTimeCoolingStep02 + 3600*0.5
endTimeCoolingStep04 = endTimeCoolingStep03 + 3600*1.2
endTimeCoolingStep05 = endTimeCoolingStep04 + 3600*0.5
endTimeCoolingStep06 = endTimeCoolingStep05 + 3600*1.1
endTimeCoolingStep07 = endTimeCoolingStep06 + 3600*0.6
endTimeCoolingStep08 = endTimeCoolingStep07 + 3600*1.0
endTimeCoolingStep09 = endTimeCoolingStep08 + 3600*0.7
endTimeCoolingStep10 = endTimeCoolingStep09 + 3600*1.0
endTimeCoolingStep11 = endTimeCoolingStep10 + 3600*1.0
endTimeCoolingStep12 = endTimeCoolingStep11 + 3600*1.0
endTimeCoolingStep13 = endTimeCoolingStep12 + 3600*1.2
endTimeCoolingStep14 = endTimeCoolingStep13 + 3600*1.0}

! Boundary Conditions
horizontalAirCooling = URAMP(t - 0.0, t - 1.0) * coolingairHorizontal_h * (temperature -
ambientTemperature)
horizontalContactCooling = (coolingairkf / (0.050) + coolingairhRadiation) * (temperature -
ambientTemperature)
verticalAirCooling = coolingairVertical_h * (temperature - ambientTemperature)
verticalWaterCooling = coolingwaterh * (temperature - coolingwaterWaterTemperature)

blockVerticalCooling =
IF t <=endTimeCoolingStep00 THEN
0.0
ELSE IF t <=endTimeCoolingStep01 THEN
verticalAirCooling
{
ELSE IF t <=endTimeCoolingStep02 THEN
verticalWaterCooling
ELSE IF t <=endTimeCoolingStep03 THEN
verticalAirCooling
ELSE IF t <=endTimeCoolingStep04 THEN
verticalWaterCooling
ELSE IF t <=endTimeCoolingStep05 THEN
verticalAirCooling
ELSE IF t <=endTimeCoolingStep06 THEN
verticalWaterCooling
ELSE IF t <=endTimeCoolingStep07 THEN
verticalAirCooling
ELSE IF t <=endTimeCoolingStep08 THEN
verticalWaterCooling
ELSE IF t <=endTimeCoolingStep09 THEN
verticalAirCooling
ELSE IF t <=endTimeCoolingStep10 THEN
verticalWaterCooling
```



```

ELSE IF t <=endTimeCoolingStep11 THEN
verticalAirCooling
ELSE IF t <=endTimeCoolingStep12 THEN
verticalWaterCooling
ELSE IF t <=endTimeCoolingStep13 THEN
verticalAirCooling
ELSE IF t <=endTimeCoolingStep14 THEN
verticalWaterCooling}
ELSE
0.0

blockHorizontalCooling = IF t <=endTimeCoolingStep00 THEN
horizontalAirCooling
ELSE IF t <=endTimeCoolingStep01 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep02 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep03 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep04 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep05 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep06 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep07 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep08 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep09 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep10 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep11 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep12 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep13 THEN
horizontalContactCooling
ELSE IF t <=endTimeCoolingStep14 THEN
horizontalContactCooling}
ELSE
0.0

potVerticalCooling = IF t <=endTimeCoolingStep00 THEN
verticalAirCooling
ELSE
verticalAirCooling

! Other Calculations
liquidFraction = min(max((temperature -slagSolidusTemperature_0)/( slagLiquidusTemperature -
slagSolidusTemperature_0),0),1)
liquidFraction2 = max(min(liquidFraction, 1.0), 0.0)

INITIAL VALUES
temperature = initialTemperature

EQUATIONS
temperature: slagDensity*cp*dt(temperature) - div(k*grad(temperature)) = 0.0

BOUNDARIES

{REGION 1 'Block'
initialTemperature = slagTemperature_AtTap
initialLiquidFraction = 1.0
k = slagk
Cp = slagCp

NOBC(temperature) START (0.0, -blockL)
ARC(CENTER=0.0, 0.0) TO (blockrt, blockzt)
NOBC(temperature) LINE TO (blockR, blockZ - blockL)
NATURAL(temperature)=blockHorizontalCooling LINE TO (0.0, blockZ - blockL)
NOBC(temperature) LINE TO CLOSE}

REGION 1 'Block'
k = slagk
cp = slagCp

NATURAL(temperature)=blockVerticalCooling ARC(CENTER=0.0, 0.0) TO (blockrt, blockzt)
NATURAL(temperature)=blockVerticalCooling LINE TO (blockR, blockZ - blockL)
NATURAL(temperature)=blockHorizontalCooling LINE TO (0.0, blockZ - blockL)
NOBC(temperature) LINE TO CLOSE

{REGION 2 'Pot'
initialTemperature = potTemperature_BeforeTap
initialLiquidFraction = 0.0
k = steelk
Cp = steelCp

NOBC(temperature) START (0.0, -blockL)
ARC(CENTER=0.0, 0.0) TO (blockrt, blockzt)
NOBC(temperature) LINE TO (blockR, blockZ - blockL)

```



```
NATURAL(temperature)=horizontalAirCooling LINE TO (blockR + potWallThickness /
cos(potTheta_Radians), blockZ - blockL)
NATURAL(temperature)=verticalAirCooling LINE TO (blockrt +
potWallThickness * sin(potPhi_Radians), blockzt - potWallThickness * cos(potPhi_Radians))
NATURAL(temperature)=verticalAirCooling ARC(CENTER=0.0, 0.0) TO (0.0, -blockL -
potWallThickness)
NOBC(temperature) LINE TO CLOSE}

FEATURE 'Block Vertical Outer Surface'
START (0.0, -blockL)
ARC(CENTER=0.0, 0.0) TO (blockrt, blockzt)
LINE TO (blockR, blockZ - blockL)

FEATURE 'Block Horizontal Outer Surface'
START (blockR, blockZ - blockL)
LINE TO (0.0, blockZ - blockL)

FEATURE 'Block Centre Line'
START (0.0, -blockL)
LINE TO (0.0, -blockL + blockZ)

FEATURE 'Block Roundend Radius'
START (0.0, 0.0)
LINE TO (blockrt, blockzt)

TIME
{0.0 BY 1E-1 TO endTimeCoolingStep00}
endTimeCoolingStep00 BY 1E-1 TO endTimeCoolingStep01

MONITORS
FOR t= 0, 0.1, 0.5, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 15, 20, 25, 30, 45, 60, 300, 600 BY 600 TO 3600
BY 3600 TO ENDTIME
CONTOUR(temperature) RANGE=(0, 1700)
CONTOUR(temperature) PAINTED RANGE=(0, 1700)
! CONTOUR(liquidFraction2) PAINTED RANGE=(0.0, 1.0)
! ELEVATION(temperature) ON 'Block Vertical Outer Surface'
! ELEVATION(verticalAirCooling) ON 'Block Vertical Outer Surface'
! ELEVATION(temperature) ON 'Block Horizontal Outer Surface'
! ELEVATION(horizontalAirCooling) ON 'Block Horizontal Outer Surface'

PLOTS
FOR t=endTimeCoolingStep00 BY 3600*24 TO ENDTIME
! CONTOUR(temperature) RANGE=(0, 1700)
! CONTOUR(temperature) PAINTED RANGE=(0, 1700)
! CONTOUR(max(min(liquidFraction, 1.0), 0.0)) PAINTED RANGE=(0.0, 1.0)
! ELEVATION(temperature) ON 'Block Vertical Outer Surface'
! ELEVATION(verticalAirCooling) ON 'Block Vertical Outer Surface'
! ELEVATION(temperature) ON 'Block Vertical Outer Surface' EXPORT FORMAT "#d, #1"
! ELEVATION(horizontalAirCooling) ON 'Block Horizontal Outer Surface'

! FOR t=0 BY 3600.00*24 TO ENDTIME
! ELEVATION(temperature) ON 'Block Vertical Outer Surface' EXPORT FORMAT "#x,#y,#1"

! PLANT BLOCK
! FOR t = 0, 0.1, 0.5, 1, 2, 4, 6, 8, 10, 15, 20, 25, 30, 45, 60, 80, 100, 200, 300, 400, 500, 600 BY
600 TO ENDTIME
! ELEVATION(liquidFraction) ON 'Block Centre Line' EXPORT FORMAT "#y, #1"
! ELEVATION(temperature, coolingairvertical_h) ON 'Block Vertical Outer Surface' EXPORT FORMAT "#d,
#1, #2"
! CONTOUR(temperature) RANGE=(0,1700) EXPORT FORMAT "#x, #y, #1"
! ELEVATION(temperature) ON 'Block Roundend Radius' EXPORT FORMAT "#d, #1"

SUMMARY
REPORT potMassCapacity
REPORT tapMass
REPORT potR
REPORT potTheta_Degrees
REPORT potTheta_Radians
REPORT potWallThickness

REPORT potVolume
REPORT blockVolume
REPORT potPhi_Degrees
REPORT potPhi_Radians
REPORT blockL
REPORT blockrt
REPORT blockzt
REPORT blockVolume_SphericalCap
REPORT blockVolume_ConicalSection
REPORT blockR
REPORT blockZ
REPORT slagSolidusTemperature_0
REPORT slagLiquidusTemperature

{FOR t = ENDTIME
TRANSFER(temperature, liquidFraction) FILE="transfer.dat"}
```





```
HISTORIES
!   HISTORY(SURF_INTEGRAL(blockVerticalCooling, 'Block Vertical Outer Surface') / SURF_INTEGRAL(1,
'Block Vertical Outer Surface')) AS 'VerticalHeatLoss'
!   HISTORY(VOL_INTEGRAL(liquidFraction2, 1) / VOL_INTEGRAL(1, 1)) AS 'TotalLiquidFraction'

      {Tap 37 & 38 C10 thermocouple positions}
!   HISTORY(temperature) AT (0,0.13)   AT(0,-0.020) AT(0,-0.120)   AS 'ThermocouplePositions' EXPORT
FORMAT "#t#r,#i"
!   HISTORY(temperature, coolingairVertical_h, coolingwaterh) AT (0.0, blockZ - blockL) AT (blockR,
blockZ - blockL) AT (blockrt, blockzt) AT (0.0, -blockL) AS 'Block surface temperatures' EXPORT FORMAT
"#t #r, #i"

!   TAP 42
!   HISTORY(SURF_INTEGRAL(blockHorizontalCooling, 'Block Horizontal Outer Surface')) AS 'Heatloss to
ground' EXPORT FORMAT "#t,#1"
!   HISTORY(SURF_INTEGRAL(blockVerticalCooling, 'Block Vertical Outer Surface')) AS 'Heatloss to
surroundings' EXPORT FORMAT "#t,#1"

!   PLANT TAPS
!   HISTORY(temperature, coolingairvertical_h) AT (0.0, blockZ - blockL) AT (blockR, blockZ - blockL)
AT (blockrt, blockzt) AT (0.0, -blockL) AS 'Block surface temperatures' EXPORT FORMAT "#t #r, #i"

END
```

## 5.6 List of expressions used to calculate the heat transfer coefficient in spray cooling. <sup>37,38</sup>

### 5.7 Nomenclature

$d_{32}$  = Sauter mean drop diameter, m

(the Sauter mean diameter is the drop diameter with a volume to surface area which is equal that of the entire water spray)

$Nu$  = Nusselt number, dimensionless

$Pr$  = Prandtl number, dimensionless

$Re_{32}$  = Reynolds number based on the Sauter Mean Diameter<sup>sss</sup>, dimensionless

$T$  = temperature, °C

$Q''$  = volumetric flux of the cooling water, m<sup>3</sup>s<sup>-1</sup>/m<sup>2</sup>

$q''$  = heat flux, W/m<sup>2</sup>

$k$  = thermal conductivity, W/m<sup>2</sup>/°C

$\sigma$  = surface tension of water, N/m

$\rho$  = density, kg/m<sup>3</sup>

$\nu$  = drop speed, m/s

$\mathcal{G}$  = kinematic viscosity of water, m<sup>2</sup>/s

$h_{fg}$  = latent heat of vaporisation, J/kg

### Subscripts

*incip* = point of incipient nucleate boiling

*crit* = point of critical heat flux

*min* = point of minimum heat flux (Leidenfrost point)

*dfb* = point of departure from film boiling

*s* = hot surface

*w* = water

*v* = vapour

$$\text{Re}_{32} = \frac{Q'' d_{32}}{g}$$

$$\text{Nu}_f = 2.512 \text{Re}_{32}^{0.76} \text{Pr}_w^{0.56}$$

$$T_{incip} = 13.43 \text{Re}_{32}^{0.167} \text{Pr}_w^{0.123} \left( \frac{k_w}{d_{32}} \right)^{0.22} + T_w$$

$$T_{crit} = 18 \left( \rho_v h_{fg} Q'' \left( \frac{\sigma}{\rho_w Q''^2 d_{32}} \right)^{0.198} \right)^{\frac{1}{5.55}} + T_w$$

$$T_{min} = 204.895 Q''^{0.066} \nu^{0.138} d_{32}^{-0.035} + T_w$$

$$T_{dfb} = 280.762 Q''^{0.087} \nu^{0.11} d_{32}^{-0.035} + T_w$$

$$q''_{dfb} = 6100300 Q''^{0.588} \nu^{0.244}$$

$$q''_{min} = 3324400 Q''^{0.544} \nu^{0.324}$$

$$q''_{crit} = 122.4 \rho_g h_{fg} Q'' \left( 1 + 0.0118 \left( \frac{\rho_w}{\rho_g} \right)^{0.25} \left( \frac{\rho_w C p_w (T_w^{sat} - T_w)}{\rho_g h_{fg}} \right) \right) \left( \frac{\sigma}{\rho_w Q''^2 d_{32}} \right)^{0.0198}$$

$$N_2 = \frac{q''_{dfb} - q''_{min}}{(T_{dfb} - T_{min})^2}$$

$$N_1 = -2N_2 (T_{min})$$

$$N_0 = q''_{min} - N_1 (T_{min}) - N_2 (T_{min})^2$$

$$T_s > T_{dfb} : h = \frac{63.25(T_s - T_w)^{1.691} Q^{0.264} d_{32}^{-0.062}}{(T_s - T_w)}$$

$$T_s = T_{dfb} : h = \frac{q_{dfb}''}{(T_s - T_w)}$$

$$T_s < T_{dfb}; T > T_{min} : h = \frac{N_0 + N_1(T_s - T_w) + N_2(T_s - T_w)^2}{(T_s - T_w)}$$

$$T_s = T_{min} : h = \frac{q_{min}''}{(T_s - T_w)}$$

$$T_s < T_{min}; T > T_{crit} : h = \frac{q_{crit}''}{(T_s - T_w)} - \frac{q_{crit}'' - q_{min}''}{(T_{crit} - T_{min})^3} \times$$

$$\frac{(T_{crit}^3 - 3T_{crit}^2 T_{min} + 6T_{crit} T_{min}(T_s - T_w) - 3(T_{crit} + T_{min})(T_s - T_w)^2 + 2(T_s - T_w)^3)}{(T_s - T_w)}$$

$$T_s = T_{crit} : h = \frac{q_{crit}''}{(T_s - T_w)}$$

$$T_s < T_{crit}; T > T_{incip} : h = \frac{0.0000187(T_s - T_w)^{5.55}}{(T_s - T_w)}$$

$$T_s < T_{incip} : h = \frac{Nu_w k_w}{d_{32}}$$

## 5.8 Tap information and composition of the two thermocouple blocks

### Tap chemistry

	TiO <sub>2</sub>	FeO	Al <sub>2</sub> O <sub>3</sub>	CaO	Cr <sub>2</sub> O <sub>3</sub>	MgO	MnO	SiO <sub>2</sub>	V <sub>2</sub> O <sub>5</sub>	Total ****
Tap 37	85.42	13.25	0.71	0.04	0.11	0.89	1.51	1.01	0.45	104.02
Tap 38	84.21	13.92	0.75	0.04	0.15	1.26	1.45	0.94	0.45	102.94

### Tap details

	Block mass (kg)	Tap rate (kg/min)	Tap temperature (°C)
Tap 37	1,002	204.6	1669
Tap 38	1,365	345.7	1668

---

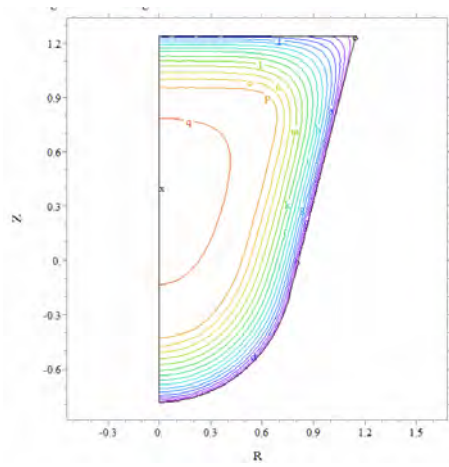
\*\*\*\* The total exceeds 100% because all titanium, including Ti<sup>3+</sup>, is reported as Ti<sup>4+</sup> (TiO<sub>2</sub>)



### 5.9 Cross sections water vs. air cooling

Water

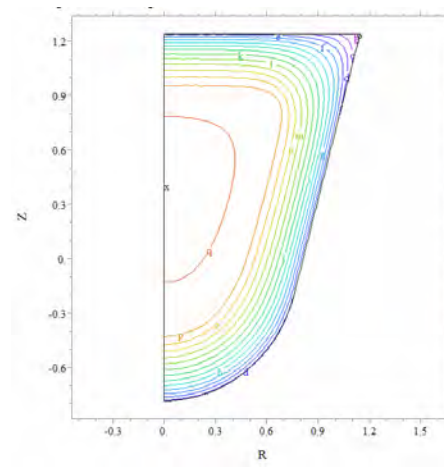
Air



temperature

max	1.70
r:	1.70
q:	1.60
p:	1.50
o:	1.40
n:	1.30
m:	1.20
l:	1.10
k:	1.00
j:	0.90
i:	0.80
h:	0.70
g:	0.60
f:	0.50
e:	0.40
d:	0.30
c:	0.20
b:	0.10
a:	0.00
min	0.00

Scale = E3

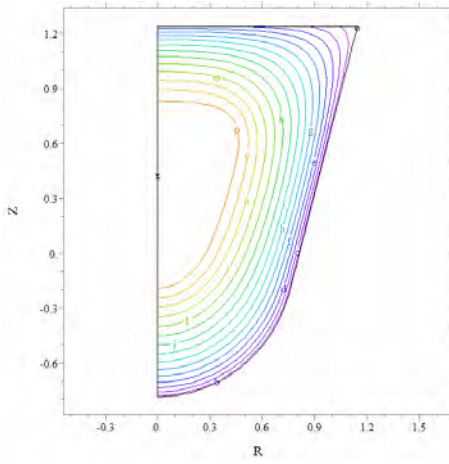


temperature

max	1.70
r:	1.70
q:	1.60
p:	1.50
o:	1.40
n:	1.30
m:	1.20
l:	1.10
k:	1.00
j:	0.90
i:	0.80
h:	0.70
g:	0.60
f:	0.50
e:	0.40
d:	0.30
c:	0.20
b:	0.10
a:	0.00
min	0.00

Scale = E3

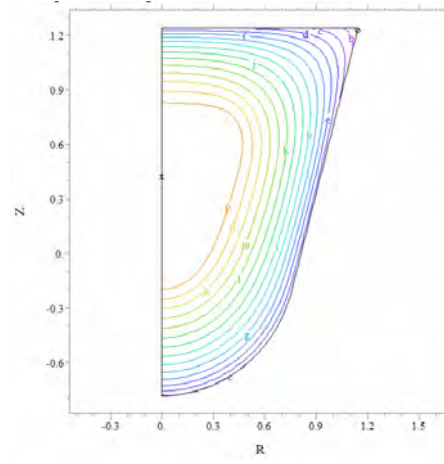
1 day



temperature

max	1.70
r:	1.70
q:	1.60
p:	1.50
o:	1.40
w:	1.30
m:	1.20
l:	1.10
k:	1.00
j:	0.90
i:	0.80
h:	0.70
g:	0.60
f:	0.50
e:	0.40
d:	0.30
c:	0.20
b:	0.10
a:	0.00
min	0.00

Scale = E3

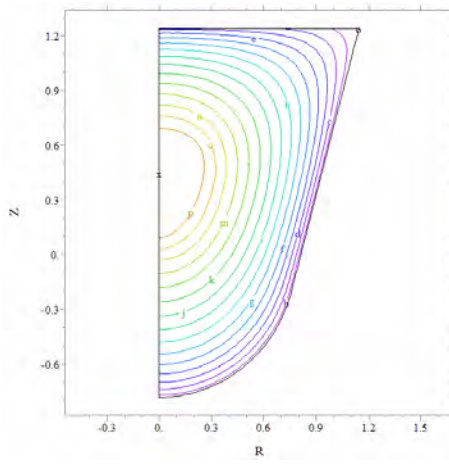


temperature

max	1.70
r:	1.70
q:	1.60
p:	1.50
o:	1.40
w:	1.30
m:	1.20
l:	1.10
k:	1.00
j:	0.90
i:	0.80
h:	0.70
g:	0.60
f:	0.50
e:	0.40
d:	0.30
c:	0.20
b:	0.10
a:	0.00
min	0.00

Scale = E3

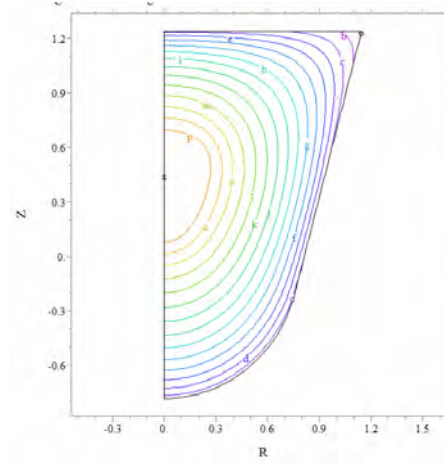
2 days



temperature

max	1.70
r:	1.70
q:	1.60
p:	1.50
o:	1.40
n:	1.30
m:	1.20
l:	1.10
k:	1.00
j:	0.90
i:	0.80
h:	0.70
g:	0.60
f:	0.50
e:	0.40
d:	0.30
c:	0.20
b:	0.10
a:	0.00
min	0.00

Scale = E3



temperature

max	1.70
r:	1.70
q:	1.60
p:	1.50
o:	1.40
n:	1.30
m:	1.20
l:	1.10
k:	1.00
j:	0.90
i:	0.80
h:	0.70
g:	0.60
f:	0.50
e:	0.40
d:	0.30
c:	0.20
b:	0.10
a:	0.00
min	0.00

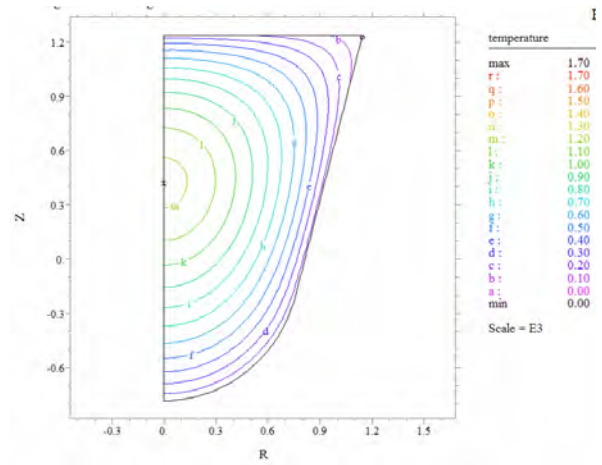
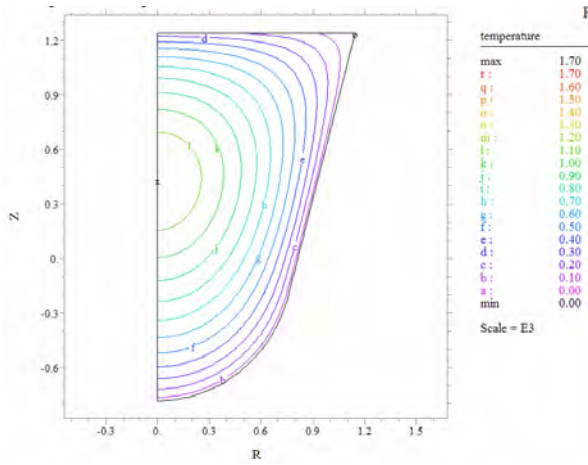
Scale = E3

3 days

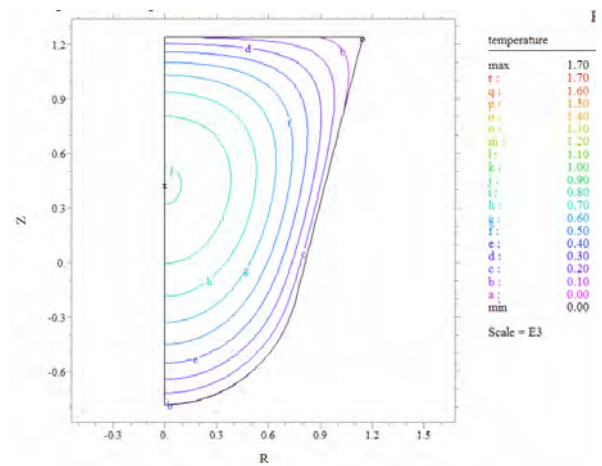
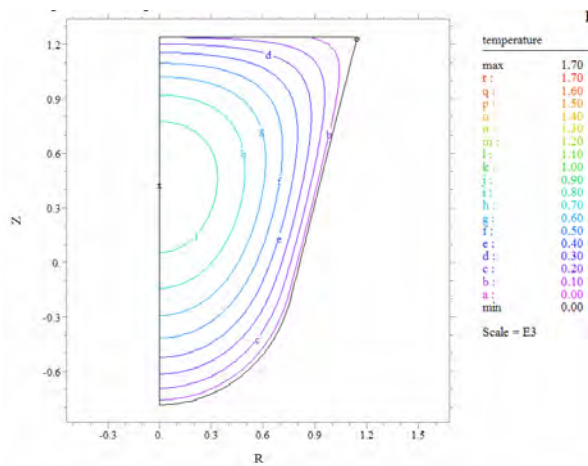


Water

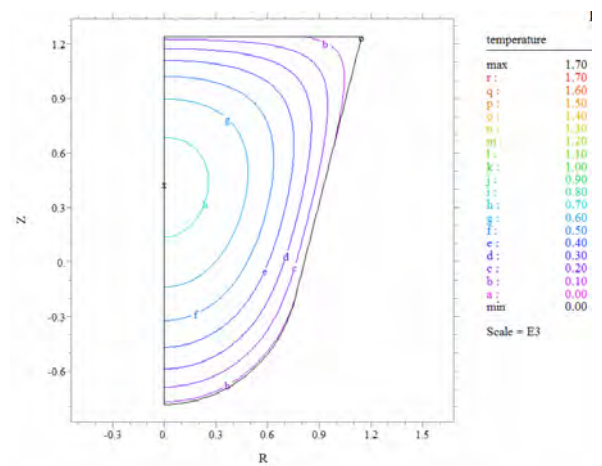
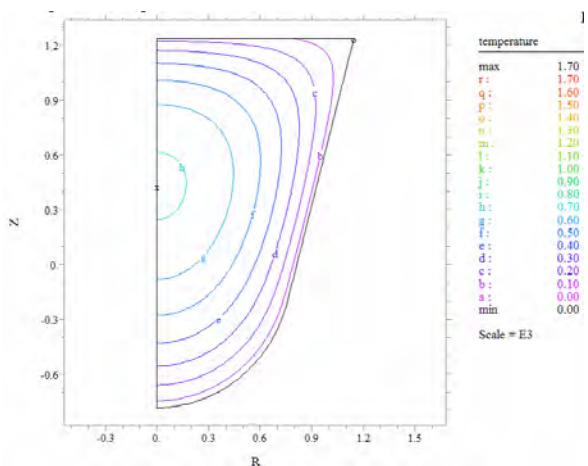
Air



4 days



5 days

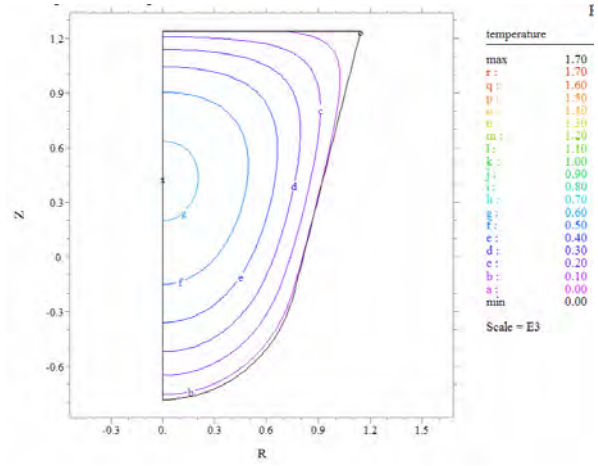
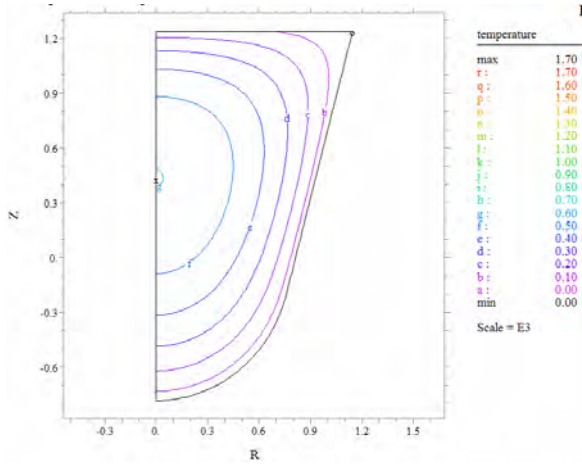


6 days

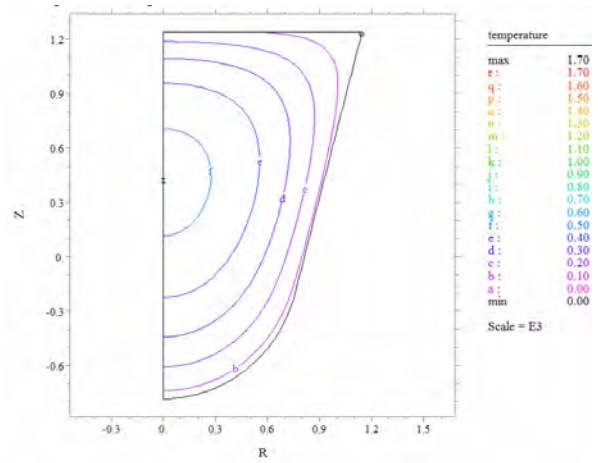
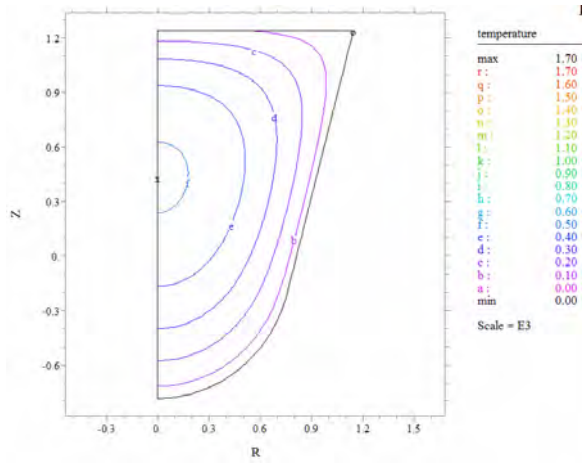


Water

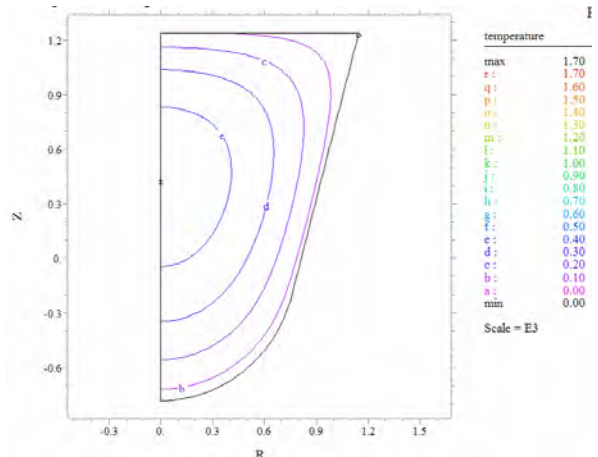
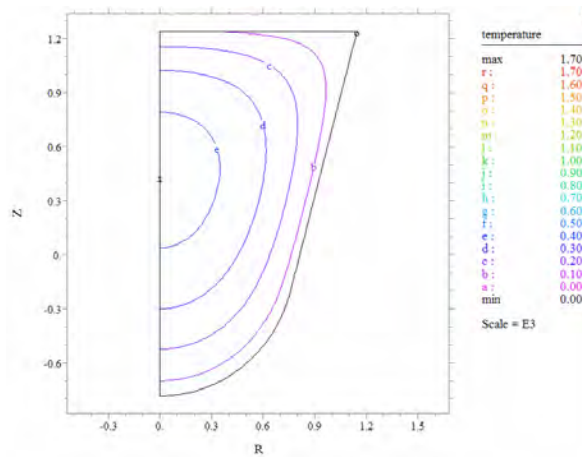
Air



7 days



8 days

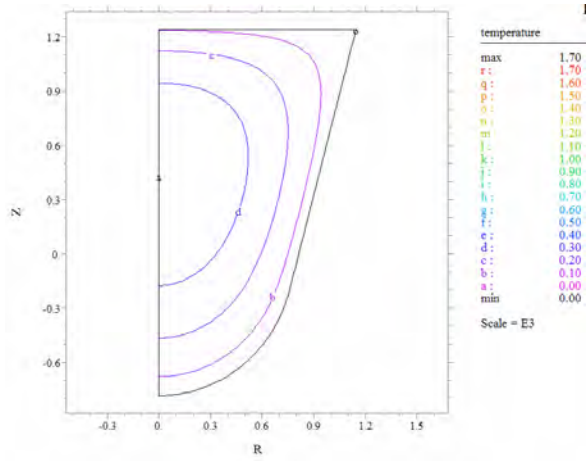


9 days

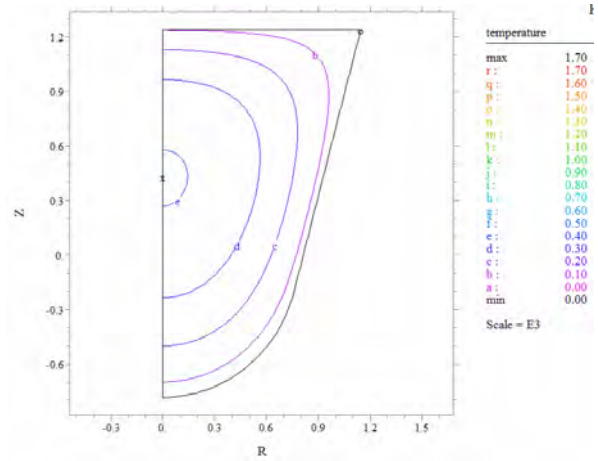




### Water



### Air



10 days

## 6 References

- 1 [http://en.wikipedia.org/wiki/Titanium\\_dioxide](http://en.wikipedia.org/wiki/Titanium_dioxide). (Accessed 15 September 2006)
- 2 T. P. Battle, D. Nguyen and J.W. Reeves: "The Processing of titanium-containing ores." *The Paul E. Queneau International Symposium*. Vol. I: Fundamental Aspects., pp. 925-945, 1993.
- 3 H. Kotzé, D. Bessinger and J. Beukes: "Ilmenite smelting at Tigor SA." *The Journal of the South African Institute for Mining and Metallurgy*, March 2006, vol. 106, pp. 165-170.
- 4 PC Pistorius and C Coetzee: "Physicochemical aspects of titanium slag production and solidification." *Metallurgical and Materials Transactions Series B*, vol. 34B, pp. 581-588B (2003).
- 5 Bessinger, D., Geldenhuis, J.M.A. and Pistorius P.C.: "Phase changes in the decrepitation of solidified high titania slags." *Heavy Minerals 2005*, Society for Mining, Metallurgy and Exploration, 2005. pp. 213-219.
- 6 Bessinger, D., Geldenhuis, J.M.A., Pistorius, P.C., Mulaba, A., Hearne, G.: "The decrepitation of solidified high titania slags." *Journal of Non-Crystalline Solids*, vol. 282, no. 1, pp. 132-142 (2001).
- 7 J.P.R de Villiers, J. Göske and A. Tuling: "Disintegration in high grade titania slags: low temperature oxidation reactions and associated fracture mechanics of pseudobrookite." *Mineral Processing and Extractive Metallurgy (Trans. Inst. Min. Metall C)*, June 2005, vol. 114, pp. C73-C79.
- 8 M Gous: "An overview of the Namakwa Sands ilmenite smelting operations." *The Journal of the South-African Institute of Mining and Metallurgy*, June 2006, vol. 106, pp. 379-384.
- 9 P.C. Pistorius: "Ilmenite smelting – the basics." *The 6th International Heavy Minerals Conference 'Back to Basics'*. The Southern African Institute of Mining and Metallurgy, 2007. pp. 75-83.
- 10 J.H. Zietsman and P.C. Pistorius: "Process mechanisms in ilmenite smelting." *Journal of the South African Institute of Mining and Metallurgy*, vol. 104, pp. 653-660 (2004).
- 11 M. Guéguin and F. Cardarelli: "Chemistry and mineralogy of titania-rich slags. Part 1—hemo-ilmenite, sulphate, and upgraded titania slags." *Mineral processing and extractive metallurgy review*, vol. 28, pp. 1-58 (2007).
- 12 K. Borowiec, A.E. Grau, M. Guéguin and J.-F. Turgeon: "Method to upgrade titania slag and resulting product." United States Patent no. 5,830,420 (1998).
- 13 H. Elstad, J.M. Eriksen, A. Hildal, T. Rosenqvist, and S. Seim: "Equilibrium between titania slags and metallic iron." *The 6th International Heavy Minerals Conference 'Back to Basics'*. The Southern African Institute of Mining and Metallurgy, 2007. pp. 35-42.

- 14 G. Eriksson and A.D. Pelton: "Critical evaluation and optimization of the thermodynamic properties and phase diagrams of the MnO-TiO<sub>2</sub>, MgO-TiO<sub>2</sub>, FeO-TiO<sub>2</sub>, Ti<sub>2</sub>O<sub>3</sub>-TiO<sub>2</sub>, Na<sub>2</sub>O-TiO<sub>2</sub> and K<sub>2</sub>O-TiO<sub>2</sub> systems." *Metallurgical Transactions B*, vol. 24B, pp. 795-805 (1993).
- 15 G. Eriksson, A.D. Pelton, E. Woermann and E. Ender: "Measurement and thermodynamic evaluation of phase equilibria in the Fe-Ti-O system." *Berichte der Bunsengesellschaft für physikalische Chemie*, vol. 100, pp. 1839-1849 (1996).
- 16 J. Pesl and R.H. Eric: "High-temperature phase relations and thermodynamics in the iron-titanium-oxygen system." *Metallurgical and Materials Transactions B*, vol. 30B, pp. 695-705 (1999).
- 17 C.W. Bale, P. Chartrand, S.A. Degterov, G. Eriksson, K. Hack, R. Ben Mahfoud, J. Melançon, A.D. Pelton and S. Petersen: "FactSage Thermochemical Software and Databases." *Calphad*, vol. 26, pp. 189-228 (2002).
- 18 S. Jahanshahi, L. Zhang and D. Bessinger: "Development of MPE package and its application in ilmenite smelting." *Heavy Minerals 2003*. South African Institute of Mining and Metallurgy, 2003. pp. 119-126.
- 19 D.J. Fourie, J.J. Eksteen and J.H. Zietsman: "Calculation of FeO-TiO<sub>2</sub>-Ti<sub>2</sub>O<sub>3</sub> liquidus isotherms pertaining to high titania slags." *Journal of the South African Institute of Mining and Metallurgy*, vol. 105, pp. 695-710 (2005).
- 20 P.C. Pistorius: "Fundamentals of freeze lining behaviour in ilmenite smelting." *Journal of the South African Institute of Mining and Metallurgy*, vol. 103, pp. 509-514 (2003).
- 21 J.M.A. Geldenhuis and P.C. Pistorius. "The use of commercial oxygen probes during the production of high titania slags." *Journal of the South African Institute of Mining and Metallurgy*, vol. 99, pp. 41-47 (1999).
- 22 P.C. Pistorius and T. Motlhamme: "Oxidation of high-titanium slags in the presence of water vapour." *Minerals Engineering*, vol. 19, pp. 232-236 (2006).
- 23 I.E. Grey, L.M.D. Cranswick, C. Li, T.J. White, and L.A. Bursill: "New M<sub>3</sub>O<sub>5</sub>-anatase intergrowth structures formed during low-temperature oxidation of anosovite." *Journal of Solid State Chemistry*, vol. 150, pp. 128-138 (2000).
- 24 J.P. van Dyk, N.M. Vegter, C.P. Visser, T. de Lange, J.D. Winter, E.A. Walpole and J. Nell: "Beneficiation of titania slag by oxidation and reduction treatment." United States patent no. 6,803,024 (2004).
- 25 Errol G. Kelly, David J. Spottiswood: "Introduction to mineral processing." *Australian Mineral Foundation*, pp. 461, (1995).
- 26 P.C. Pistorius: "The relationship between FeO and Ti<sub>2</sub>O<sub>3</sub> in ilmenite smelter slags." *Scandinavian Journal of Metallurgy*, 2002, vol. 31, pp. 120-125.

- 27 R.M. Viljoen, J.T Smit, I. du Plessis and V.Ser: "The development and application of in-bed compression breakage principles." *Minerals Engineering*, vol. 14, No. 5, pp. 465 – 471, 2001.
- 28 H Brundiek and F Poeschl: "A roller mill for cement and blast furnace slag in theory and practice." *37<sup>th</sup> IEEE Cement Industry Technical Conference*, June 1995, San Juan, Puerto Rico. pp. 197-223.
- 29 H.S. Carslaw and J.C. Jaeger: *Conduction of heat in solids*, second edition. Clarendon, 1959. pp. 282-286.
- 30 J.P. Holman. "Heat Transfer." *McGraw-Hill Book Co*, SI Metric Edition, pp. 635, (1989).
- 31 A. Hamasaiid, M.S. Dargusch, C.J. Davidson, S. Tovar, T. Loulou, F. Rezaï-Aria, and G. Dour: "Effect of mold coating materials and thickness on heat transfer in permanent mold casting of aluminum alloys." *Metallurgical and Materials Transactions A*, vol. 38A, pp. 1303-1316 (2007).
- 32 R. Le Goff, G. Poutot, D. Delaunay, R. Fulchiron, and E. Koscher: "Study and modeling of heat transfer during the solidification of semi-crystalline polymers." *International Journal of Heat and Mass Transfer*, vol. 48, pp. 5417–5430 (2005).
- 33 M. Bahrami, J.R. Culham, M.M. Yananovich and G.E. Schneider: "Review of thermal joint resistance models for nonconforming solid surfaces." *Applied Mechanics Reviews*, vol. 59, pp. 1-12 (2006).
- 34 M.G. Cooper, B.B. Mikic and M.M. Yovanovich: "Thermal contact conductance." *International Journal of Heat and Mass Transfer*, vol. 12, pp. 279-300 (1969).
- 35 J.J. Salgon, F. Robbe-Valloire, J. Blouet and J. Bransier: "A mechanical and geometrical approach to thermal contact resistance." *International Journal of Heat and Mass Transfer*, vol. 40, pp. 1121-1129 (1997).
- 36 J.P. Holman. "Heat Transfer." *McGraw-Hill Book Co*, SI Metric Edition, pp. 331-345, (1989).
- 37 W.P. Klinzing, J.C. Rozzi and I. Mudawar: "Film and transition boiling correlations for quenching of hot surfaces with water sprays." *J. Heat Treating*, (1992) vol. 9, pp. 91-103.
- 38 I. Mudawar and W.S. Valentine: "Determination of the local quench curve for spray-cooled metallic surfaces." *J. Heat Treating*, (1989), vol. 7, pp. 107-121.
- 39 T. Tran, C. Solnordal and C. Nexhip: "Determination of thermal conductivity of titania slags." *Unpublished CSIRO Minerals Report*, DMR-2229, June 2003.
- 40 Y. Lee and D. Deming: "Evaluation of thermal conductivity temperature corrections applied in terrestrial heat flow studies." *Journal of Geophysical Research*, vol. 103, pp. 2447-2454 (1998).

- 41 J.R. Smyth and T.C. McCormick: "Crystallographic data for Minerals", in *Mineral Physics and Crystallography, a handbook of physical constants*. American Geophysical Union, 1995. p. 3.
- 42 M.D. Lind and R.M. Housley: "Crystallization Studies of Lunar Igneous Rocks: Crystal Structure of Synthetic Armalcolite." *Science*, vol. 175, no. 4021, pp. 521-523 (1972).
- 43 W.D. Kingery, H.K. Bowen and D.R. Uhlmann: *Introduction to ceramics*, second edition. John Wiley & Sons, 1976. pp. 612-643.
- 44 H.J. Siebeneck, D.P.H. Hasselman, J.J. Cleveland and R.C. Bradt: "Effects of grain size and microcracking on the thermal diffusivity of  $MgTi_2O_5$ ." *Journal of the American Ceramic Society*, vol. 60, pp. 336-338 (1977).
- 45 H.J. Siebeneck, D.P.H. Hasselman, J.J. Cleveland and R.C. Bradt: "Effects of microcracking on the thermal diffusivity of  $Fe_2TiO_5$ ." *Journal of the American Ceramic Society*, vol. 59, pp. 241-244 (1976).