

Effect of sulphur content on the recrystallisation behaviour of cold worked low carbon Aluminium-killed strip steels

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The influence of the sulphur and nitrogen content on the static recrystallisation behaviour of cold worked low carbon Al-killed strip steels was investigated. This was in response to the observation made by some users of these steels that the recrystallisation process after cold work was "sluggish" in some steels and, therefore, this was affecting their productivity as the continuous annealing lines had to be run slower and the batch annealing cycles required a higher annealing temperature and thus more energy input.

Two groups of Al-killed low carbon strip steels, one with low (<10 ppm) and the other with medium to high (> 70 ppm) sulphur content were studied. It was found that sulphur had an indirect but significant effect on the recrystallisation behaviour after cold work of these steels. In the high sulphur content steels, the



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sulphur precipitated as manganese sulphide (MnS) and copper sulphide (CuS/Cu₂S) coarse particles. These sulphide particles, particularly MnS, were the favoured nucleation sites for aluminium nitride (AIN) and the heterogeneous nucleation encouraged the precipitation of AIN during coiling after hot rolling. The result was that the AIN in medium to high sulphur content steels was generally associated with coarse sulphides and, therefore, the mean particle size of the AIN/MnS particles in the as-coiled steel prior to cold working and annealing, was generally much coarser than in the steels with low sulphur content. In AIN these low sulphur content steels, the nucleated homogeneously in the matrix or heterogeneously on grain boundaries or dislocations during coiling. Consequently the mean particle size of AIN in these low sulphur content steels was significantly finer, often less than 30 nm in diameter and, therefore, these particles were more effective in retarding the recrystallisation process through Zener-pinning of dislocations and moving recrystallisation fronts.

The effect of lower sulphur content was exacerbated by lower coiling temperatures (~600 °C) i.e. the recrystallisation start time increased with a decrease in coiling temperature as the AIN particles remained small due to a low coarsening rate. On the contrary, no significant sensitivity to coiling temperature was observed in the time for the start of recrystallisation after cold work in medium to high sulphur steels within the coiling temperature range 600 to 650 °C. The conclusion was that sulphur, in the presence of manganese, does not hinder



recrystallisation after cold work in Al-killed steels but promotes heterogeneous nucleation and growth of AlN on the coarser MnS. The effect is that the effective mean particle size of the AlN becomes much larger as it is not isolated homogeneously as in low sulphur steels but is tied to the sulphides. The sulphur (ppm)

dependent empirical expressions for the recrystallisation start times $t_{5\%}$ at isothermal annealing temperature of 610 °C for steels coiled at temperatures of 600 and 650 °C were found to be:

 $t_{5\%} = 33.78 exp(-0.0345S)$ for 600 °C and

 $t_{5\%} = 0.99 exp(-0.008S)$ for 650 °C.

Recrystallisation arrest was observed during the annealing process after cold work of as-quenched specimens. The apparent activation energy of the process that led to the recrystallisation arrest, being 230 kJ mol⁻¹, is of the order of the activation energy for the diffusion of aluminium in ferrite, i.e. 196.5 kJ mol⁻¹. Since the precipitation process of AlN is controlled by the slower diffusion of aluminium, this was an indication of the nucleation/clustering of fine particles of AlN, which consequently halted the recrystallisation process through effective Zener drag. This Zener-pinning effect was used to estimate indirectly the precipitation start and finish times for the AlN during the isothermal annealing after cold work of these steels.

The solubility of AIN in austenite has mostly been studied by others using the Beeghly selective dissolution technique which has often been criticised for its limitations in its lack of sensitivity to the presence of very fine AIN particles, i.e. < 10 nm, and its failure to separate the AIN from other nitrides. In this study the thermoelectric power technique [TEP] was used to study the solubility of AIN in austenite in commercial low carbon Al-killed steels; one group with a medium to higher sulphur content and the other with a lower one. The equilibrium solubility equation thus obtained was determined as:

$$Log[\%Al][\%N] = 2.6 - \frac{9710}{T}$$

where the aluminium and the nitrogen contents are in weight percentages and T is the absolute solution temperature in Kelvin.

It was found that the AIN solubility equation derived here predicted slightly higher solubility temperatures for the AIN if compared to those derived from the Beeghly method. It was also confirmed that the equilibrium solubility of AIN was not sensitive to the sulphur content (unlike the precipitation behaviour), which is in agreement with the results from others obtained through the Beeghly technique.



Key words: Heterogeneous and homogeneous nucleation, Zenerpinning force, mean particle size, activation energy, driving force, static recrystallisation, Avrami exponent and constant.



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times t (in parenthesis) when the recrystallisation resumes after the arrest.

Figure 13.4: Schematic presentation of the apparent incubation time due to the initial slow movement of the recrystallisation fronts in steels HS140-104 and LS2-65 which were coiled at 600 $^{\circ}$ C and isothermally annealed at 610 $^{\circ}$ C, where L \approx L_{min}.

Figure 13.5: (a) Modelled driving force $\triangle Gv$, (b) activation energy $\triangle G^*$ and (c) the critical radius r^* for the homogeneous nucleation of AIN, CuS and MnS in austenite and ferrite. Solid symbols are for austenite and open ones for ferrite.

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Table 2.1: Typical chemical composition of hot rolled low carbon steel sheet and strip⁽⁵⁾.

Table 5.1: The equilibrium solubility limit of MnS in austenite.

Table 10.1: Chemical compositions of the low carbon strip steels that were studied: HS = high sulphur and LS = medium to low sulphur, the first numeral is for the sulphur content and the second for the nitrogen content, both in ppm.

Table 11.1: The hot rolling and coiling schedule simulating the industrial hot rolling and coiling processes on the Gleeble 1500^{TM} , the interpass time is before the rolling pass, RT = reheat temperature, FMH = finishing mill head and F = rolling pass.

Table 12.1: TEP measurements for the various steels that were solution treated at different temperatures for 12 minutes and quenched into water. The TEP values are within an error of \pm 0.033 μ V K⁻¹ while the Δ S value is taken relative to the TEP value at 800°C.

Table 12.2: Empirical expressions for predicting the recrystallisation start time $t_{5\%}$ as a function of the sulphur content in ppm derived from the results in figure 11.21 above; CT = coiling temperature.

Table 12.3: Absolute TEP values for the steels LS2-65 and HS140-104 which were solution treated at 1150 °C for 10 minutes, hot

rolled in 4 passes and then coiled at 600 °C for 1 hour; TEP1 = absolute TEP value immediately after coiling while TEP2 = absolute TEP value after annealing at 800 °C for 2 hours cooled to 600 °C and annealed for 10 minutes and quenched in water. Error = $0.033 \, \mu V \, K^{-1}$.

Table 13.1: TEP coefficients K_{AIN} for AIN obtained from the five steels that were studied and compared to published values.

Table 13.2: Parameters that have been used to calculate the activation energy for the isothermal and homogeneous nucleation of AIN in ferrite ΔG^*_{AIN} .

Table 13.3: Estimated volume fraction V_{ν} of the precipitated AIN during isothermal annealing at 610 °C in 70 percent cold worked steels HS140-104, LS70-38, LS2-65 and HS90-12.

Table 13.4: Isothermal annealing temperatures and times, the modelled particle radii $r_{(t)}$ and their corresponding modelled Zener drag force P_z in J m⁻³ at the point when the recrystallisation resumes after the arrest.

Table 13.5: Parameter values for the calculations of $\triangle Gv$, $\triangle G^*$ and r^* for the homogeneous nucleation of MnS, AlN and Cu₂S.